X-ray Spectnoscopy and Structure o Water 524 Lars G.M. Pertersson Stockholm University

526

Photon energy [eV]

527

528

525

"HDL/LDL" Fluctuations in Ambient and Supercooled Water

- XAS indicates *predominant asymmetrical H-bonding* with fewer H-bonds than in tetrahedral model Wernet *et al.*, Science **304** (2004) 995
- XES shows *two motifs*: strongly tetrahedral ("LDL") and very disordered ("HDL"); 25:75 Tokushima *et al.*, Chem. Phys. Lett. **460** (2008) 387
- SAXS shows *density fluctuations* Enhanced upon cooling Huang *et al.*, PNAS **106**, 15214 (2009); JCP **133**, 134504 (2010)
- **XRD** shows *accelerated transition to LDL* One on supercooling to T~227 K

LCLS free-electron laser; Sellberg et al., Nature 510, 381 (2014)



XAS and XES



X-ray Absorption Spectroscopy of Water



Two-Dimensional Water Structures



Nordlund et al., Phys. Rev. B 80, 233404 (2009)

Main-edge; Collapse of 2nd Shell High Density Form

X-ray Raman scattering of high pressure ices Strong increase in main-edge



Main-edge; Collapse of 2nd Shell High Density Form



Water should have a collapsed 2nd shell High Density Liquid (HDL)

Pylkkänen et al., J. Phys. Chem. B 2010, 114, 3804

Bondlength with the Ruler



X-ray Absorption Spectroscopy of Water



Post-edge shifted to Higher energy in the liquid

Shorter H-bond distances

XES H₂O: Temperature dependence



- Intensity transferred from 1b₁' to 1b₁" as temperature is increased (fewer H-bonds)
- NO broadening, NO new peaks: Either tetrahedral OR very disordered

Tokushima et al., Chem. Phys. Lett. 460 (2008) 387

Connection between XAS/XRS and XES



Tokushima et al., Chem. Phys. Lett. 460 (2008) 387

Huang et al., PNAS. 106 (2009) 15214

Two Local Structures: Enthalpy vs. Entropy

- **Tetrahedral loses intensity** with temperature, but peak at fixed energy
- **Distorted gains intensity and disperses** with temperature in both XES and XAS
- Energy taken up through:
 - Thermal excitation of distorted species
 - Breaking up a fraction of tetrahedral species

В Gas phase 1b." 1b,' 65°C 38°C 10°C 7°C Crystalline loe Ice formed at -190 °C Amorphous Ice 527 6 525 526 528 524 Photon energy [eV] Energy shifts consistent 4°C — 90°C — ∆E=0.2 eV XAS H₂O 535 540 Energy (eV)

Tokushima *et al.*, Chem. Phys. Lett. **460** (2008) 387 Huang *et al.*, PNAS **106**, 15214 (2009)

XES H₂O

Participator Decay in XES Experimental evidence of broken H-bonds



When exciting on the pre-edge the excited electron stays localized Excited electron may decay back – Participator decay Similar energy emitted – Vibrational losses OH stretch $v_{0 \rightarrow 1}$ 0.45 eV *close to gas phase symmetric* 0.453 eV and asymmetric $v_{0 \rightarrow 1}$ (0.465 eV) \rightarrow weakly interacting Liquid IR band peak position 0.422 eV

Harada et al., PRL **111**, 193001 (2013)

OH length

Isotope Effects: In HDO OH Preferentially "Broken"



In HDO in H₂O or D₂O preferentially OH is broken/weakly interacting

Femtosecond pump-probe vibrational spectroscopy may not sample all situations

Harada et al., PRL **111**, 193001 (2013)

Bimodality in Simulations? Local Structure Index (LSI)

Order molecules according to distance from molecule *i* as $r_1 < r_2 < r_3 < \cdots < r_{n(i)} < 3.7 \text{ Å} < r_{n(i)+1}$ where n(i) is the number of molecules that are within 3.7 Å from molecule *i*



$$\Delta(j;i) = r_{j+1} - r_j$$

Average: $\overline{\Delta}(i)$ is the average of $\Delta(j;i)$ over all neighbors *j* of molecule *i* within cutoff

LSI:

$$I(i) = \frac{1}{n(i)} \sum_{j=1}^{n(i)} \left[\varDelta(j;i) - \overline{\varDelta}(i) \right]^2$$

Measures degree of order

Shiratani & Sasai *J Chem Phys* **104**, 7671 (1996) Shiratani & Sasai *J Chem Phys* **108**, 3264 (1998) Appignanesi et al. *Eur. Phys. J. E* **29**, 305 (2009) Accordino et al. *Eur. Phys. J. E* **34**, 48 (2011) Wikfeldt et al., PCCP **13**, 19918 (2011)

Bimodality in Inherent Structure

Inherent structure obtained by energy minimization



Bimodality in Inherent Structure



Inherent Structure: Quench to 0 K, i.e. minimize the energy for each dump

Bimodal

LSI inherent structure High LSI fixed (LDL) Low LSI disperses (HDL)

Ambient conditions: ~75% HDL, 25% LDL (≡ XAS, XES)

T [K]

Widom line: 50% HDL, 50% LDL Maximum fluctuations $(\kappa_{\rm T}, C_{\rm P})$

Wikfeldt et al., PCCP 13, 19918 (2011)

Temperature Dependence LSI vs XES/XAS



Bimodality with similar temperature dependence in LSI inherent structure and in XES/XAS Smeared out in real structure



Dynamics Semiclassical Approximation

Need reliable technique to compute XES including life-ti vibrational interference for large clusters with many degree of freedom

• treat the nuclear degrees of freedom in the time domain

• sum over classical trajectories with QM initial condition

$$\sigma^{class}(\omega') = \sum_{traj} \sum_{F} \left| D_{F}^{+class}(\omega') \right|$$

• approximate nuclear Hamiltonians with corresponding (

$$D_{F}^{+class}(\omega') = \int_{0}^{\infty} dt D_{NI}^{+}(0) D_{FN}^{\prime+}(t) e^{-i \int_{0}^{t} (E_{F}(\tau) - E_{N}(\tau)) d\tau} e^{-\Gamma t'} e^{-i}$$

Sample structure model

Sample QM O-H and momentum distributions for the two hydrogens

Run trajectories and sum to get one spectrum Sample many to get spectrum for the model...



Ljungberg et al., PRB 82, 245115 (2010)

Computed XES



$$\sigma(\omega') \propto \sum_{f} \left| \sum_{n} \frac{\langle f | D'_{FN} | n \rangle \langle n | D_{NI} | i \rangle}{\omega' - (E_n - E_f) + i\Gamma} \right|^2$$

1b,"

527

Photon energy [eV]

----- H₂O

 D_2O

Kramers-Heisenberg expression: Interference – or dynamical – effects modify intensity distribution – but do not give new features

> Isotope difference simply more interference for D_2O making $1b_1$ peak symmetric

> Core-shifted second component on either asymmetric or symmetric background

Background - TDDFT



- Linear response TDDFT
- In its simplest form (Tamm-Dancoff) approximation:

 $\mathbf{A} \mathbf{X} = \boldsymbol{\omega} \mathbf{X}$

where

$$\mathbf{A} = \delta_{ij} \delta_{ab} \left(\varepsilon_a - \varepsilon_i \right) + K_{ia,jb}$$
$$K_{ia,jb} = \iint \psi_i^*(\mathbf{r}_1) \psi_a^*(\mathbf{r}_1) \left(\frac{1}{r_{12}} + \frac{\partial^2 E_{xC}}{\partial \rho \ (\mathbf{r}_1) \partial \rho \ (\mathbf{r}_2)} \right) \psi_j(\mathbf{r}_2) \psi_b(\mathbf{r}_2)$$





Extension to X-ray emission spectroscopy

2

1 Perform SCF calculation on core ionized state

Perform TDDFT calculations and X-ray emission transitions appear as negative eigenvalues

Can we build TDDFT model as accurate as EOM-CCSD?

Modification of Functionals

Best results from modification of a hybrid functional:

B^xLYP = X HF + (0.92 - X) B + 0.08 S + 0.19 VWN + 0.81 LYP Optimize the fraction of HF exchange: fitted to EOM-CCSD data

optimum value is 66%: B⁶⁶LYP

Wadey & Besley, JCTC, 10, 4557 (2014)

Improved Transition Energies

Excitation	EXP./eV	B3LYP / eV	B ⁶⁶ LYP / eV
CH ₄ 1t ₂ →1a ₁	276.3	282.7	276.4
$NH_3 2a_1 \rightarrow 1a_1$	395.1	401.9	395.3
$H_2O 1b_2 \rightarrow 1a_1$	527.0	535.3	527.6
CH₃OH 2a″→1a'	527.8	535.9	527.7
HF 1π→1σ	678.6	687.3	678.5

u6-311G** basis set

- Much closer agreement with experiment

Wadey & Besley, JCTC, 10, 4557 (2014)

TDDFT Computed XES



XES (no dynamics) computed from an optPBE-vdW PIMD dump *gives one peak*, but high and low LSI structures reproduce the experimental split



Besley & Pettersson, unpublished

O-O Pair-Correlation from EXAFS



Structure factor determined by Bergmann et al. J. Chem. Phys. **128**, 089902 (2008) from EXAFS wiggles in X-ray Raman Spectroscopy of water.

The derived PCF was highly peaked towards short distances compared to RMC and EPSR fits to neutron data (Soper) and x-ray data (Hura et al.)

Wikfeldt et al., J. Chem. Phys. **132**, 104513 (2010)

Combine XRD+ND with EXAFS



Fit simultaneously EXAFS signal and PCF

Joint solution possible

EXAFS very sensitive to strong, directional H-bonds

XRD+ND sees all situations

EXAFS enforces a subclass of structures with well-defined H-bonds

Complementary information





Wide Angle Region Q>0.5 Å⁻¹



Temperature Dependence



L. Skinner et al. J. Chem. Phys. 141, 214507 (2014)



Number of molecules in first shell



L. Skinner et al. J. Chem. Phys. 141, 214507 (2014)

Structure models



g₂ height of 2nd shell in the O-O Pair Correlation function Directly related to amount and ordering of tetrahedral water (LDL)

Soper and Ricci, Phys. Rev. Lett. 84, 2881 (2000)

Temperature dependence of g_{OO}(r)



Skinner et al. JCP 141, 214507 (2014)

Temperature dependence of g₂(r): Tetrahedrality not important above 46 °C



Compressibility



2nd shell well defined when water becomes anomalous

SAXS: Normal Liquid vs Water



Long Range Correlations



Long Range Correlations



Compressibility and Structure



Simulation at 298 K

Two Structural Environments



A. Nilsson and L. G. M. Pettersson, Nature Communications 6 8998 (2015)

Long range correlations of two structures

4.6

4.6

2

 $\langle \cdot \rangle$

4.5



Water is a correlated material below 320 K



Temperature

We live in the "funnel" of an ADP



Nilsson and Pettersson, Nature Communications 6, 8998 (2015)

Challenges

Character of ADP



Dynamics within H-bonds Breaking and forming H-bonds Dynamics in long-length-scale fluctuations Fragile to Strong Transition

Potential long-range coupling with dynamics

Para and Ortho Water