

# Inelastic Neutron Scattering Indirect Geometry Instrumentation

Eric School  
“Neutron Science and Instrumentation  
Neutrons for Chemistry And Materials  
Science Applications”

Timmy Ramirez-Cuesta  
Neutron Scattering Division  
Oak Ridge National Laboratory

July 9<sup>th</sup>, 2018



# Myself



Start new ✕

Distance ?  
22,566.65 km ( 14,022.27 mi )

3D 👤 🎯

− + 📍 🌍

IBCAO Landsat / Copernicus PGC/NASA U.S. Geological Survey Data SIO, NOAA, U.S. Navy, NGA, GEBCO Camera : 16,198 km 16°06'53"N 80°47'42"W 100%

# Myself



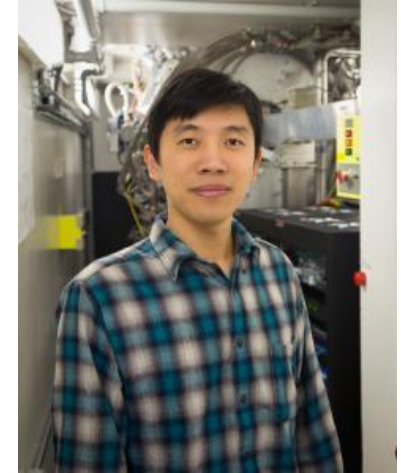
# Myself

- MSc Mathematics
- BSc Physics
- PhD Physical Chemistry
- Postdoc at University of Milwaukee
- Assistant professor Argentina
- Fellow CONICET, Argentina
- Postdoc Reading University
- TOSCA Instrument Scientist at ISIS

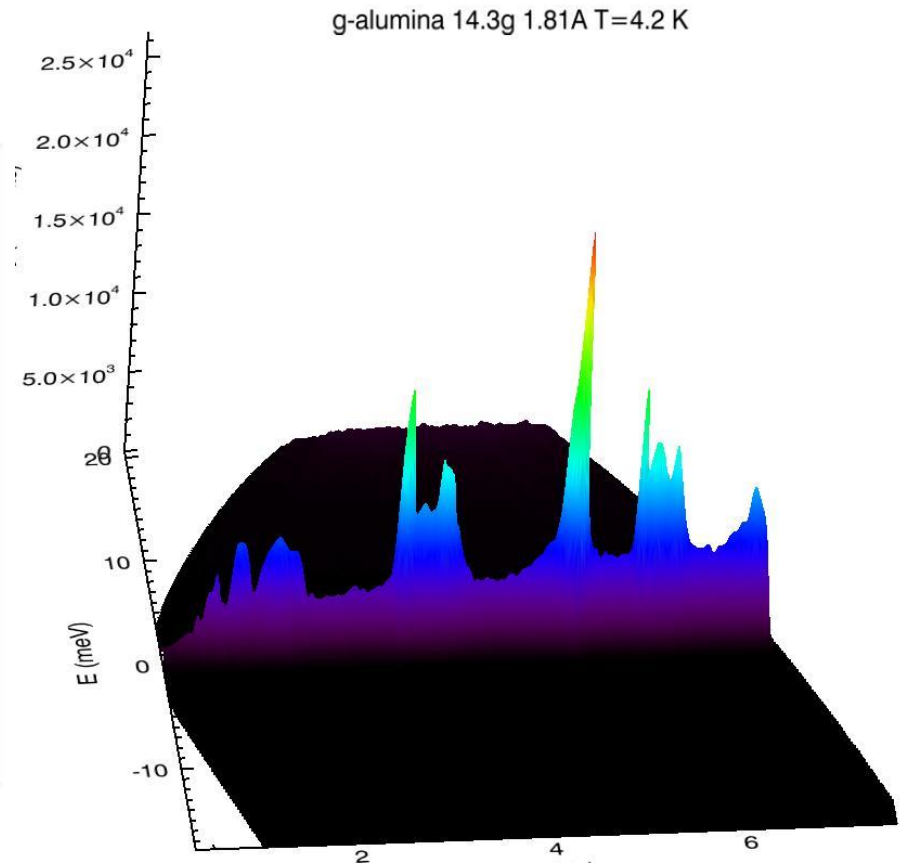
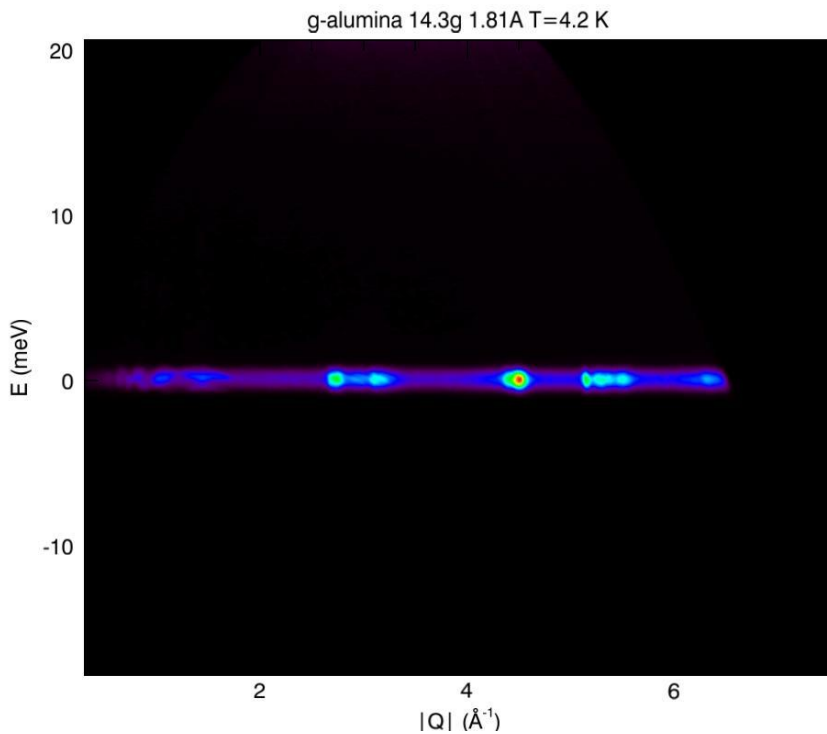


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- NScD, ORNL



# The $S(Q,\omega)$ Map



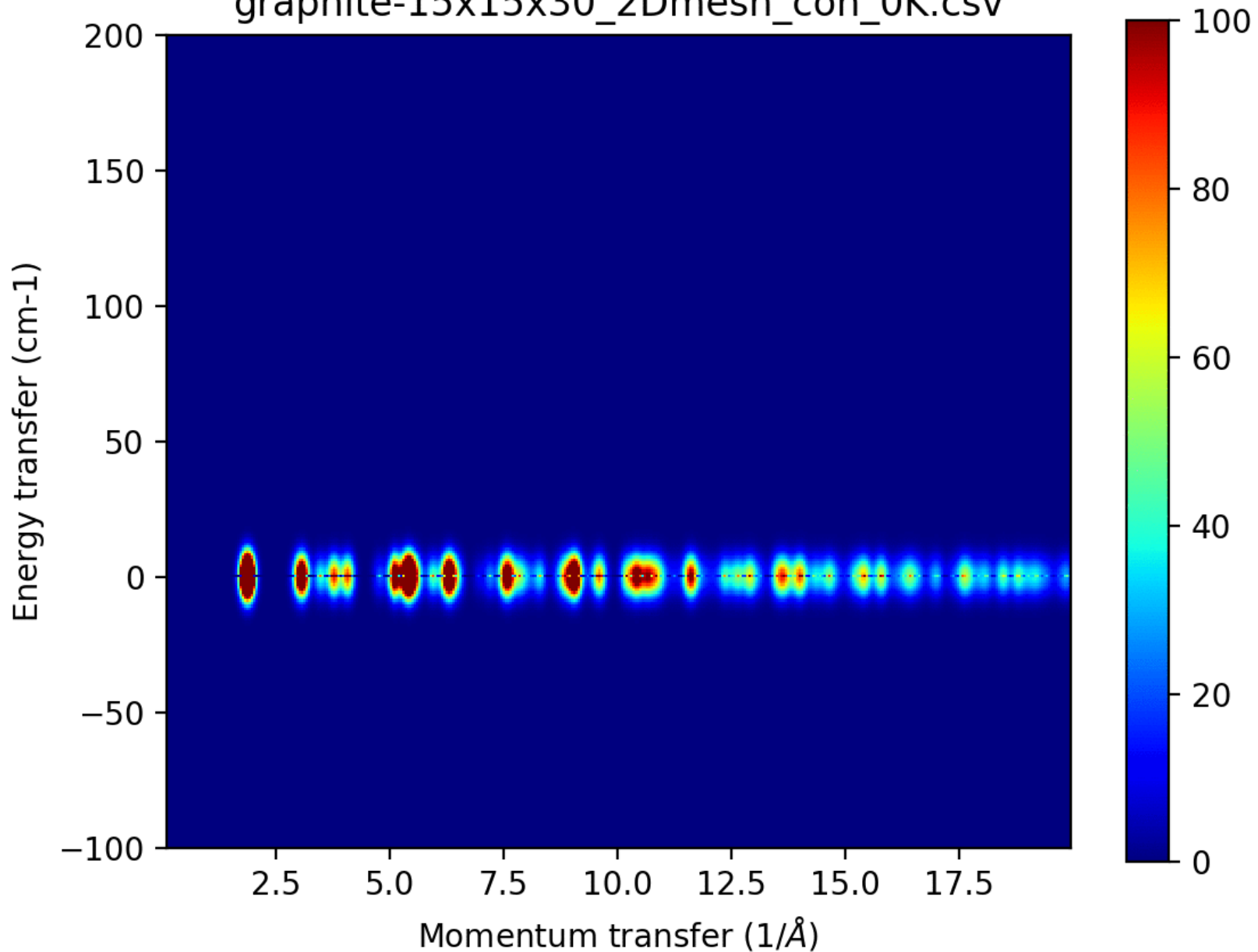
$\omega=0$

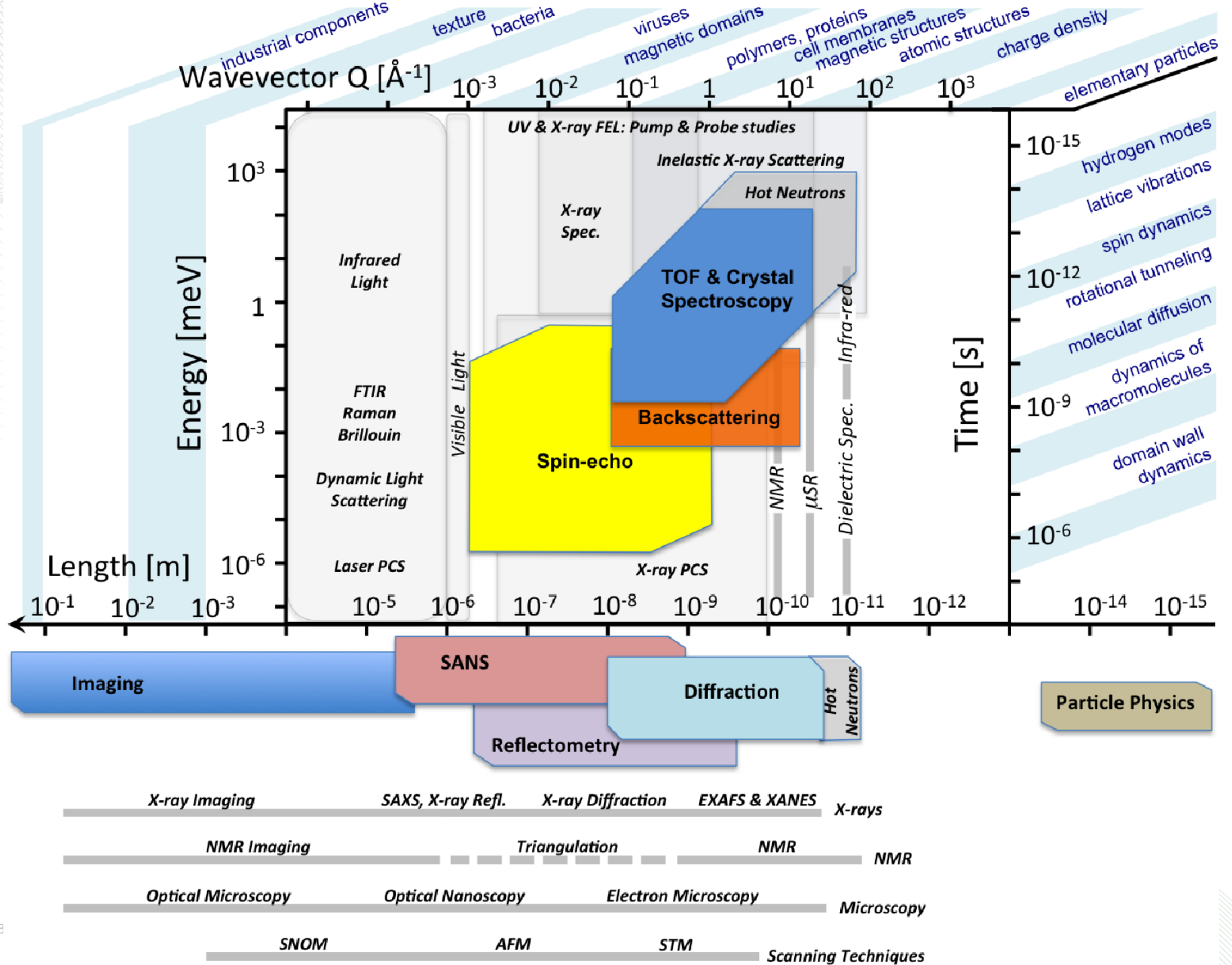
Elastic Scattering  
Structural Information

Diffraction

# "Let there be light"

graphite-15x15x30\_2Dmesh\_coh\_0K.csv



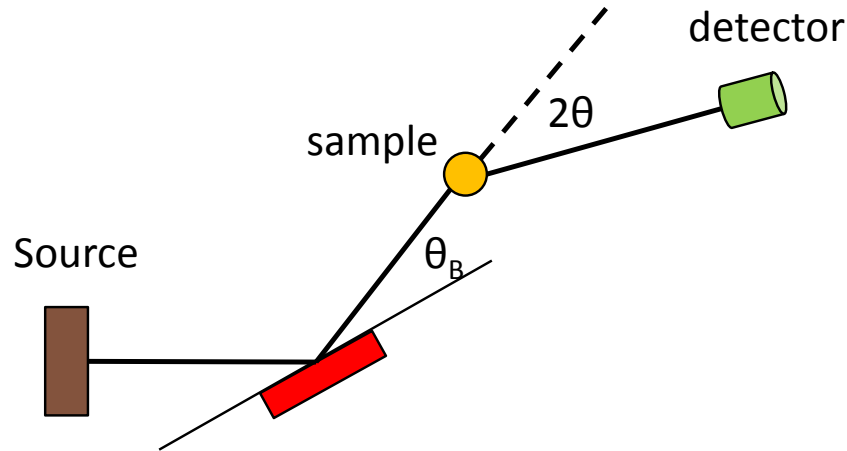




# The easy part: Neutron Diffraction

## Instrument layouts

CW

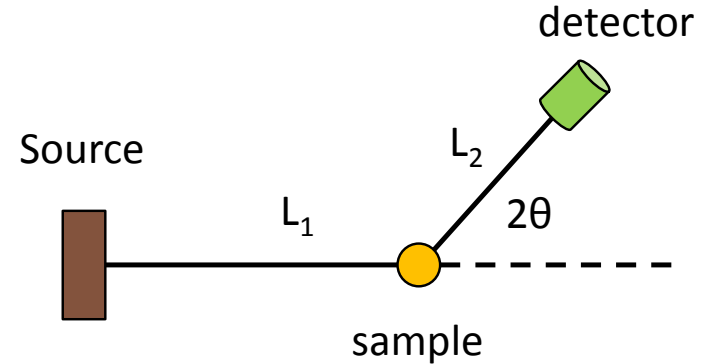


$$| = \frac{2d_c \sin \theta_B}{n}$$

$$\frac{\Delta\lambda}{\lambda} = \frac{\delta d}{d} + \cot \theta \delta\theta$$

Correlation between  $|$  and  $\theta_B$

TOF



$$| = \frac{3956}{v} = \frac{3956 (t-t_0)}{L_1+L_2}$$

$$\Delta\lambda \sim \delta t_0, \delta t, \delta L$$

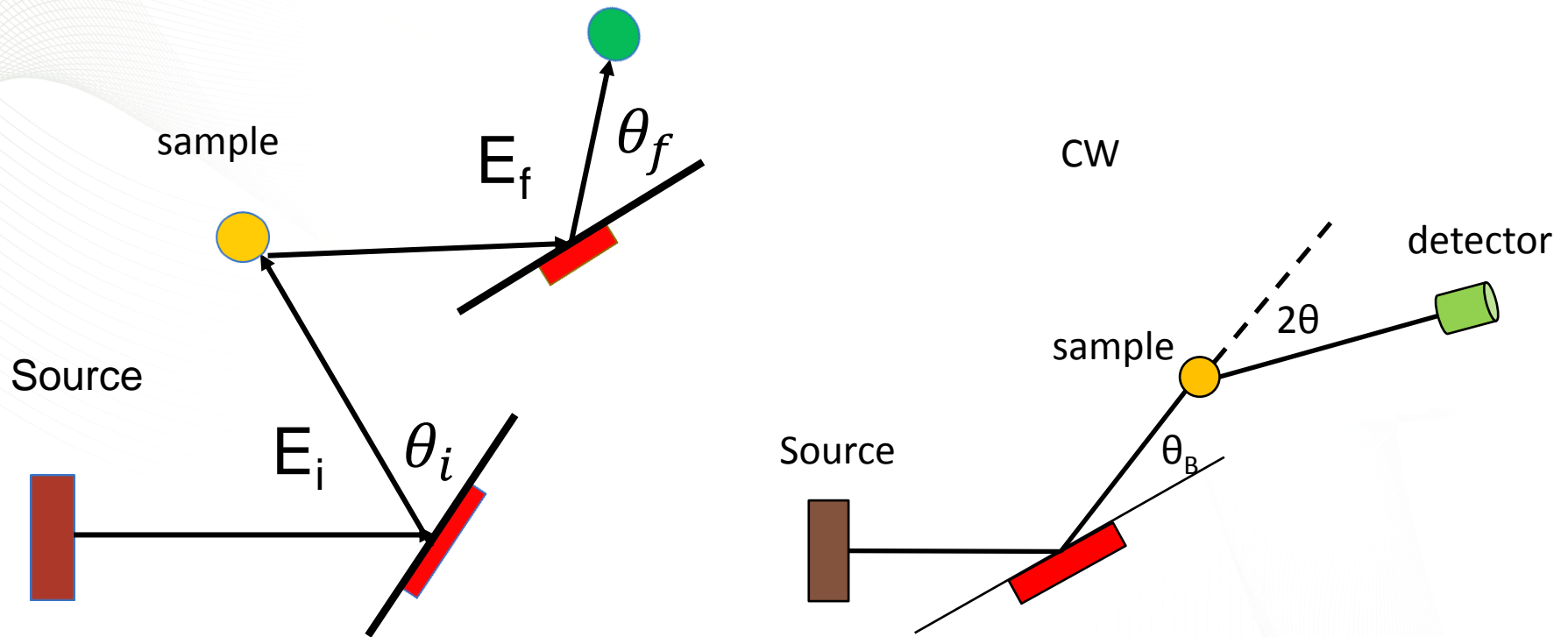
No correlation between  $|$  and  $\theta$



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# The hard part: Neutron Spectroscopy

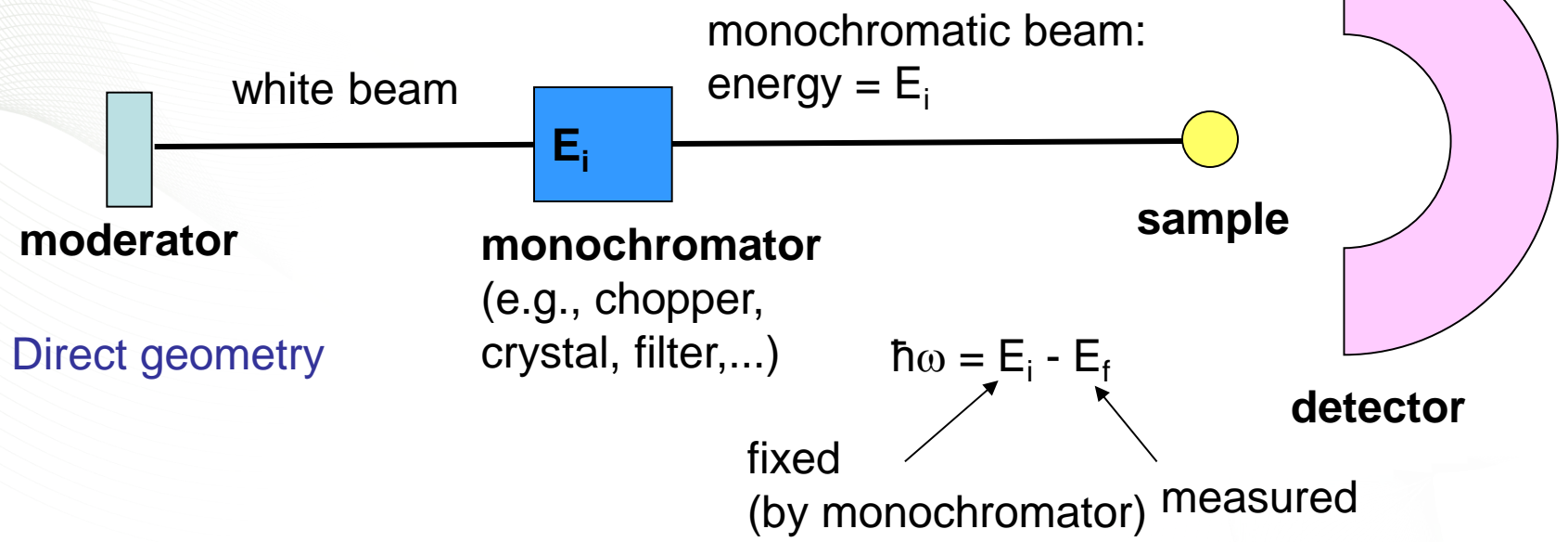


In order to discriminate the final energy, we need 2 crystals instead of one!

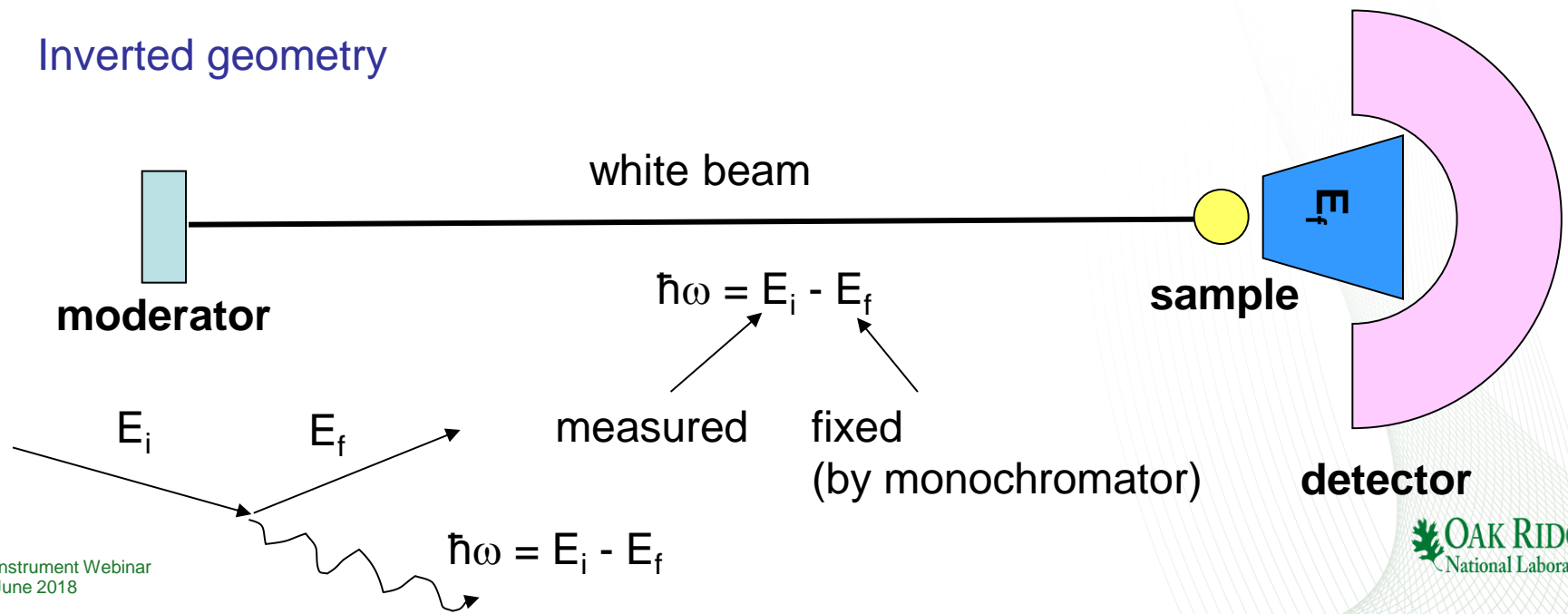
This is a triple axis instrument (in a continuous source)

We need to work twice as hard!

## Spectrometer configurations

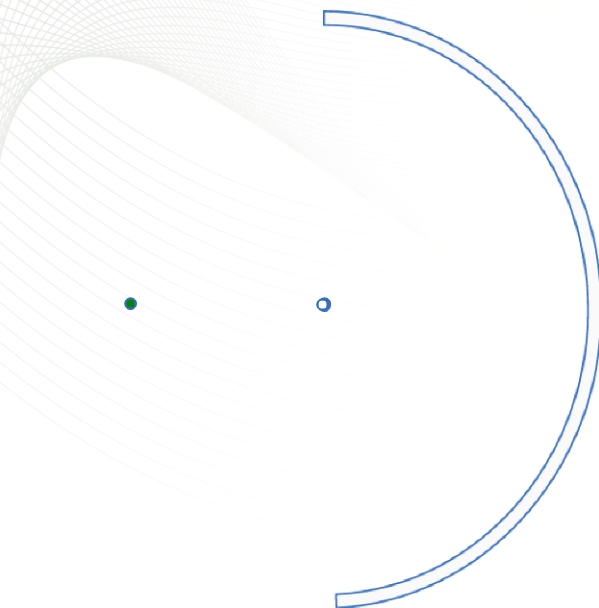


### Inverted geometry



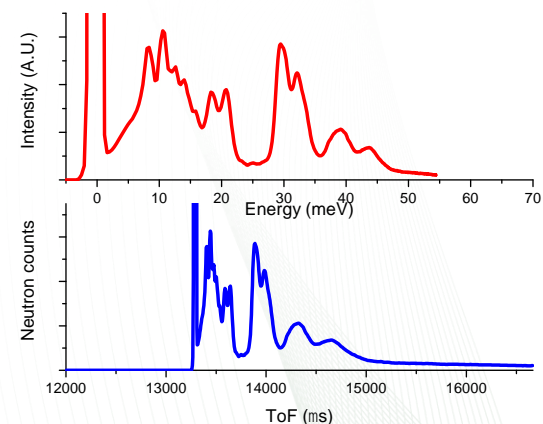
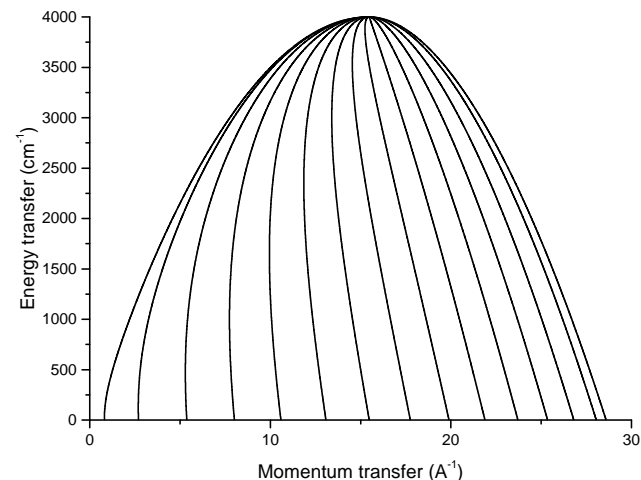
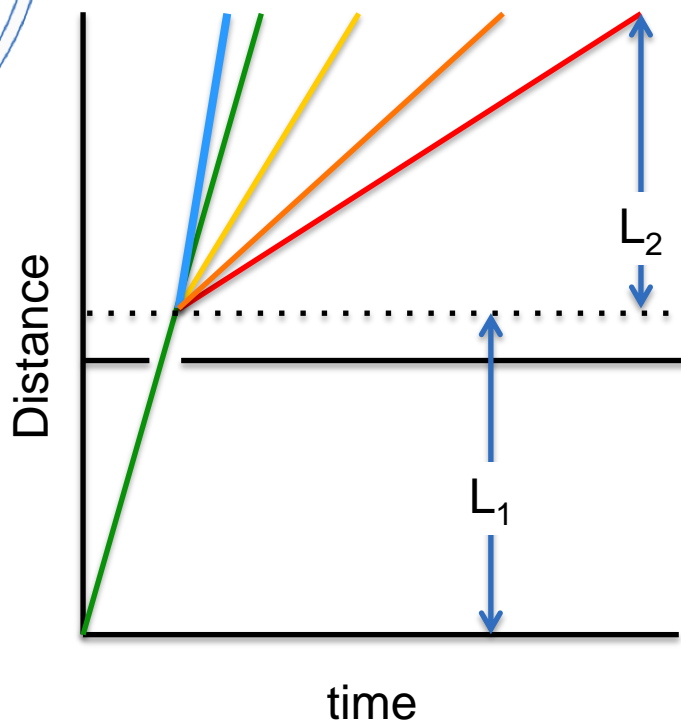
# How to measure INS (1)

## Direct Geometry Instrumentation



Direct geometry instruments measure Q trajectory is determined by the angle and energy transfer.  
 Examples: ARCS, CNCS, HYSPEC, SEQUIOA

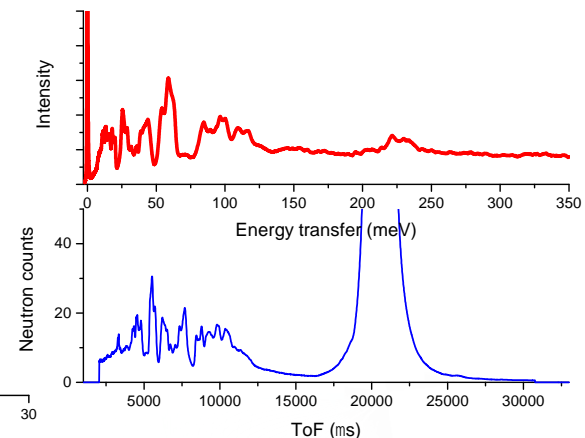
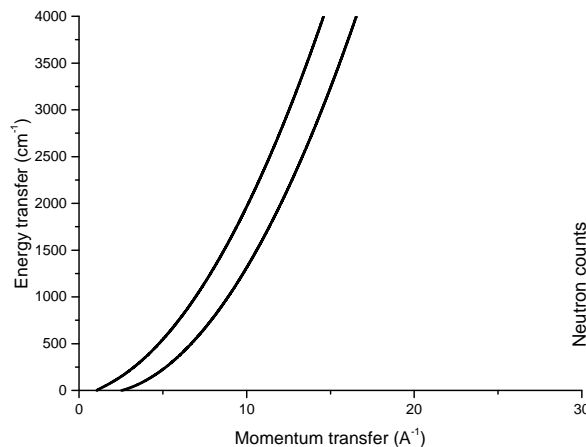
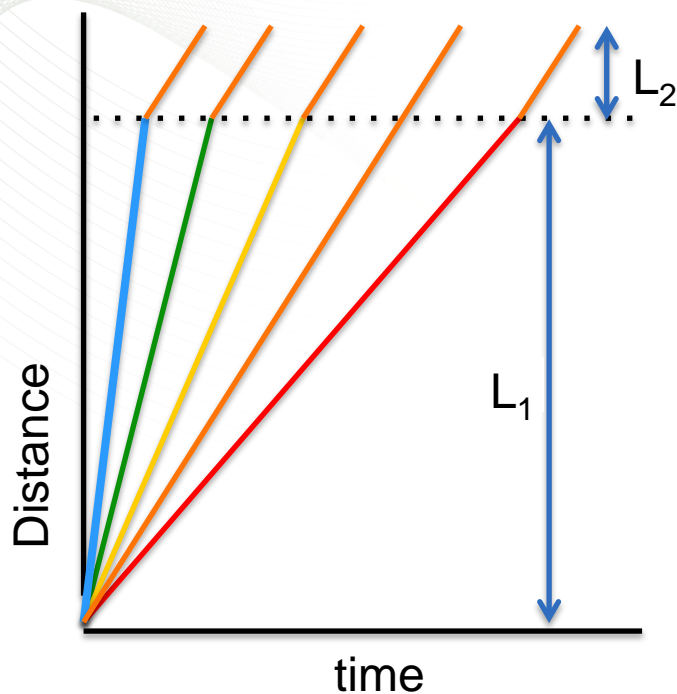
Incident neutron beam is monochromatic determining the incident energy  $E_1$ . That determines  $T_1$ . We measure the ToF and we can work out  $T_2$ .



Resolution is almost constant in units of  $E_i$

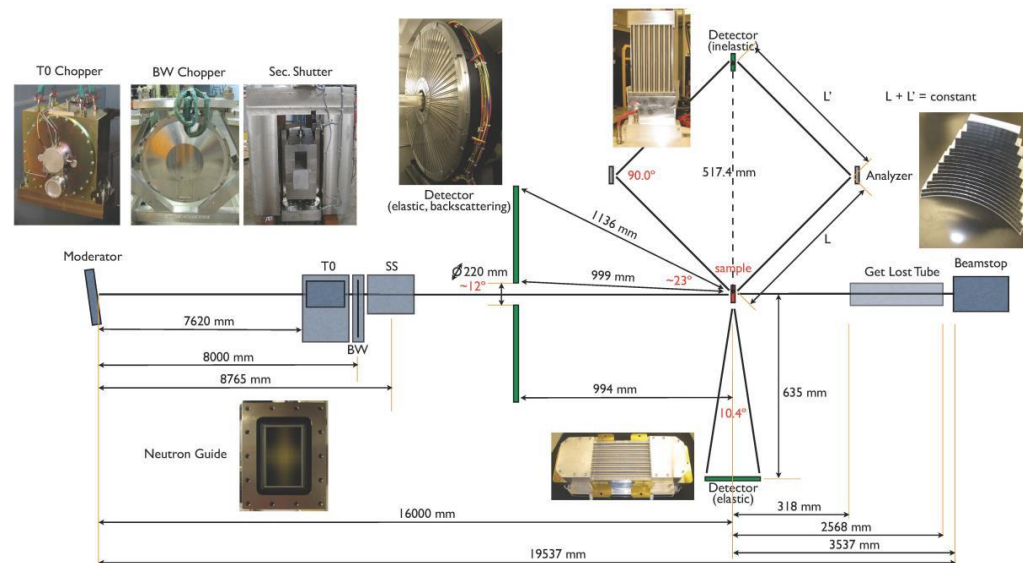
# How to measure INS (2)

## Indirect Geometry Instrumentation



Resolution is almost constant in units of  $\Delta\omega/\omega \sim 1.5\%$

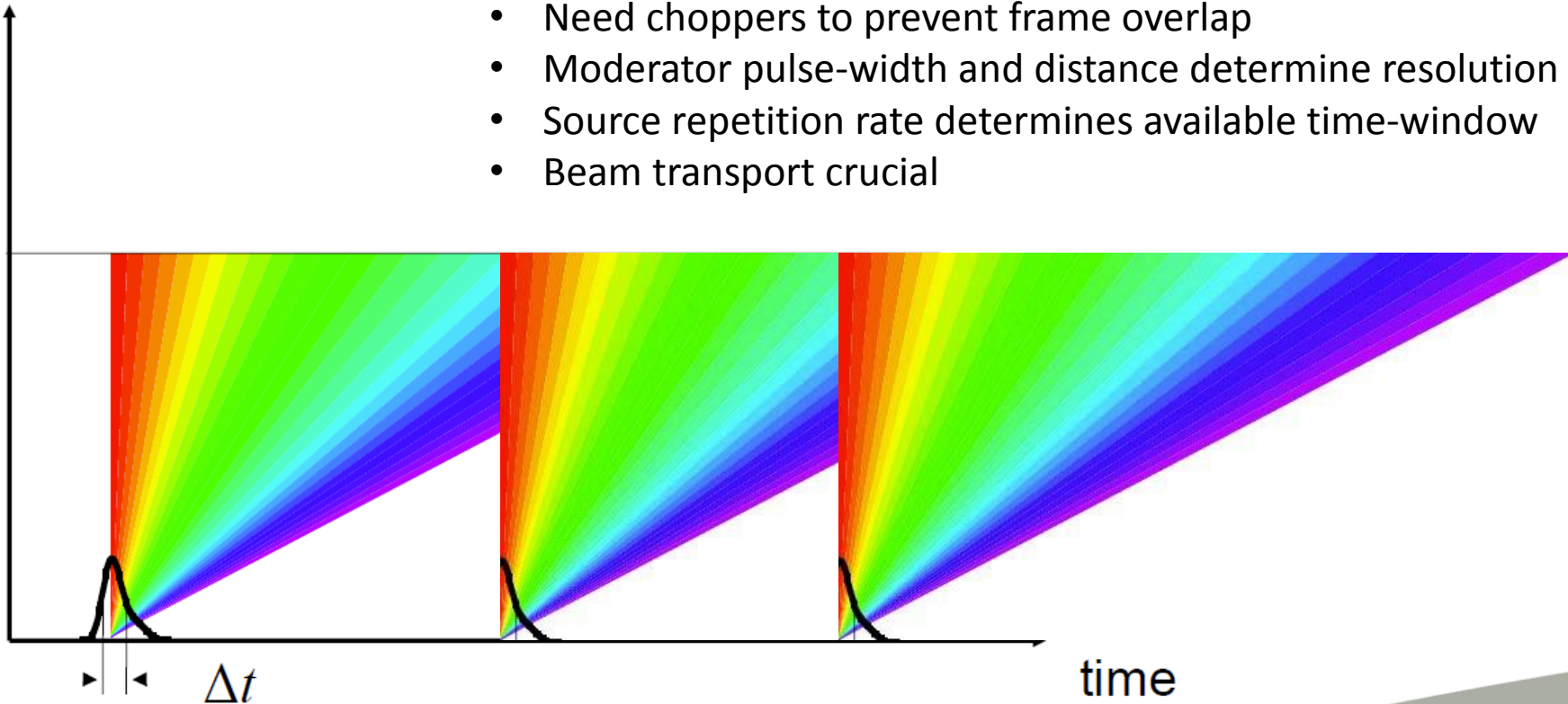
Incident neutron beam is white. We fix the energy of the scattered neutrons using an analyzer and filter device. That fixes  $T_2$ . We measure the ToF and we can work out  $T_1$ .



# TOF method

distance

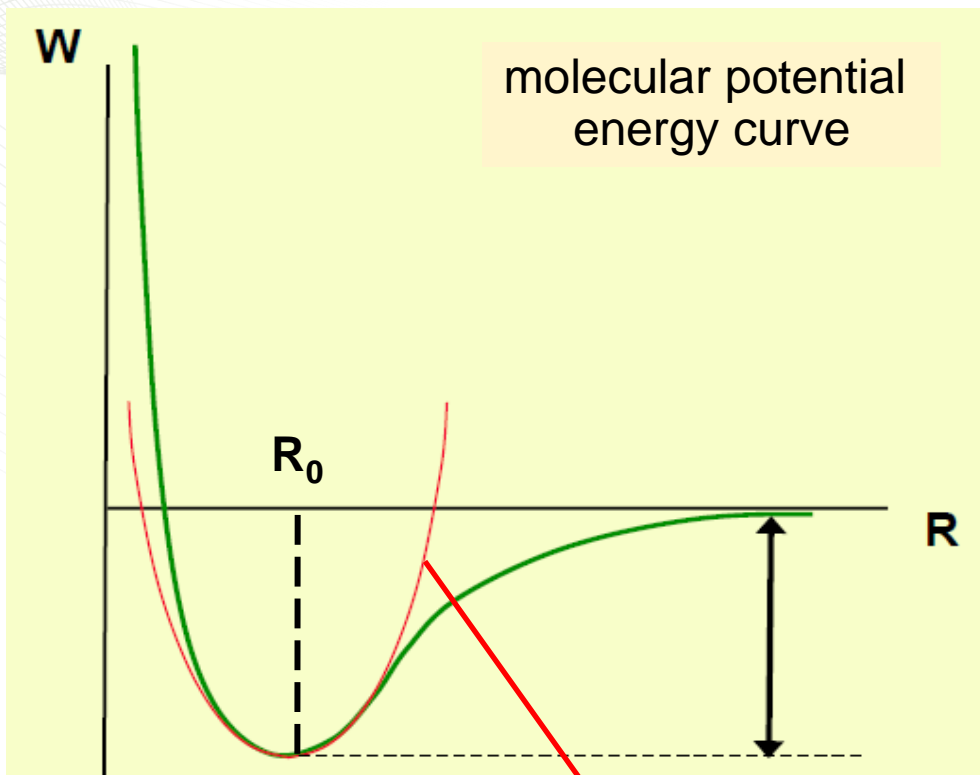
- Use distance to separate wavelengths
- Need choppers to prevent frame overlap
- Moderator pulse-width and distance determine resolution
- Source repetition rate determines available time-window
- Beam transport crucial



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# Vibrational spectroscopy



As a first approximation, a chemical bond between two atoms can be thought of as a spring connecting two masses:



Classical mechanics shows that this system vibrates with a characteristic frequency:

$$\nu = \left(\frac{1}{2\pi}\right) \sqrt{k/\mu}$$

where  $\mu$  is the reduced mass:

$$\frac{1}{\mu} = \frac{1}{m_A} + \frac{1}{m_B}$$

# Vibrational spectroscopy

Dynamics at the atomic level is determined by quantum mechanics rather than by classical mechanics. The relevant problem here is the quantum harmonic oscillator.

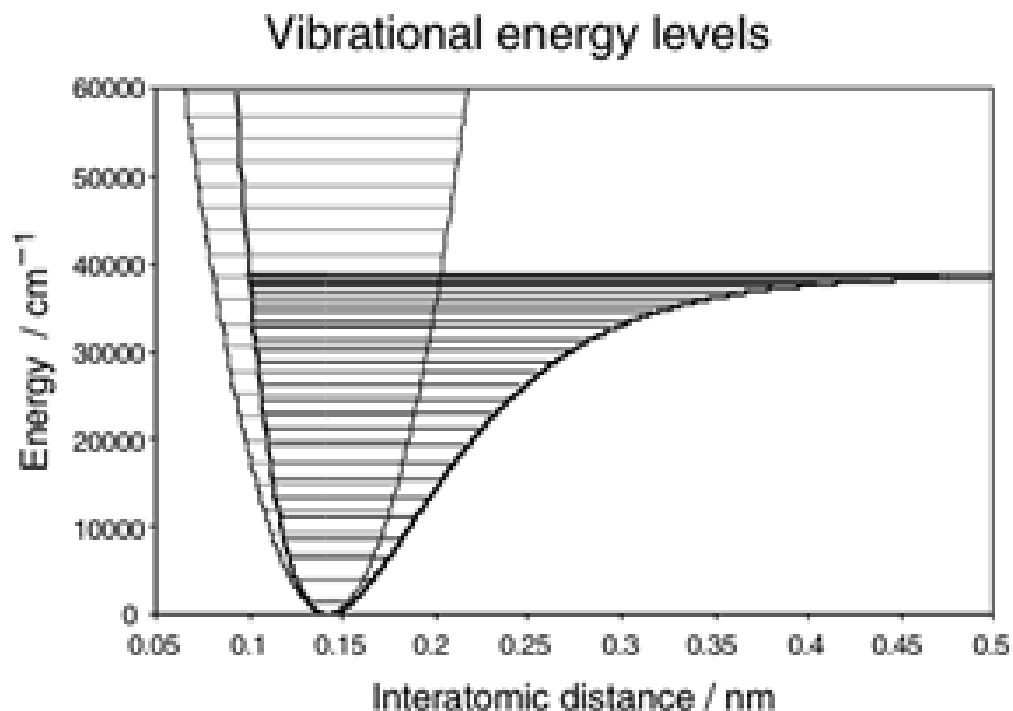
This is still an elementary problem of quantum mechanics. The energy levels of the oscillator are quantized and given by:

$$E_n = h\nu \left( n + \frac{1}{2} \right)$$

$$(n = 0, 1, 2, 3, \dots)$$

but the characteristic frequency,  $\nu$ , is still given by the classical value:

$$\nu = \left( \frac{1}{2\pi} \right) \sqrt{k/\mu}$$





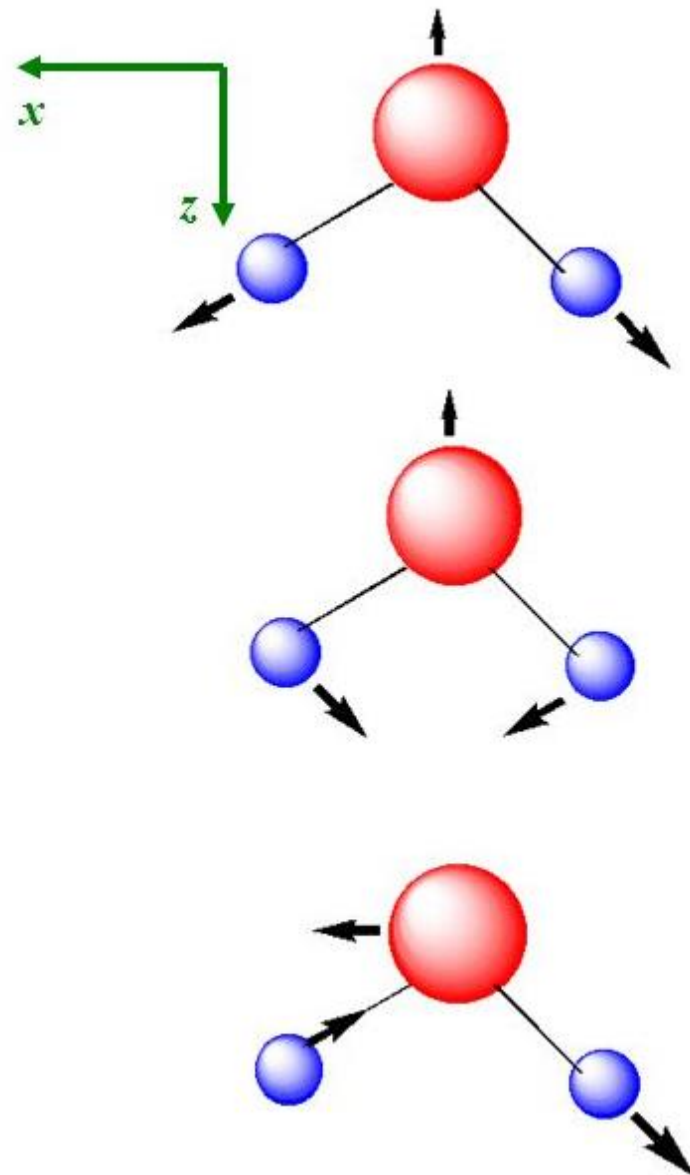
# Vibrational spectroscopy

- A molecule with N atoms is a collection of N masses connected with harmonic springs.
- Classical mechanics tells us that such a system has  $3N$  degrees of freedom.
- Three of these degrees of freedom correspond to translation of the molecule (position of its center of gravity in space), and three correspond to the orientation of the molecule in space (rotation about the center of gravity). This leaves

$$3.N - 3 - 3 = 3.N - 6$$

vibrational modes.

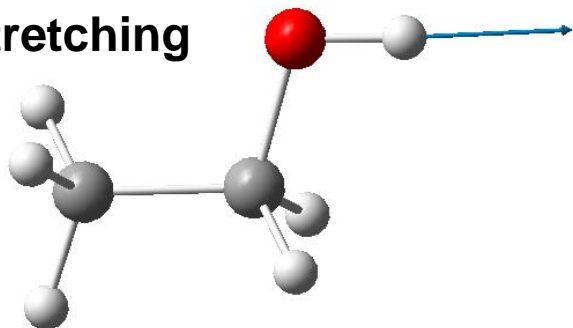
- For example  $\text{H}_2\text{O}$  ( $N=3$ ) has  $3 \cdot 3 - 6 = 3$  modes of vibration.



# Vibrational spectroscopy

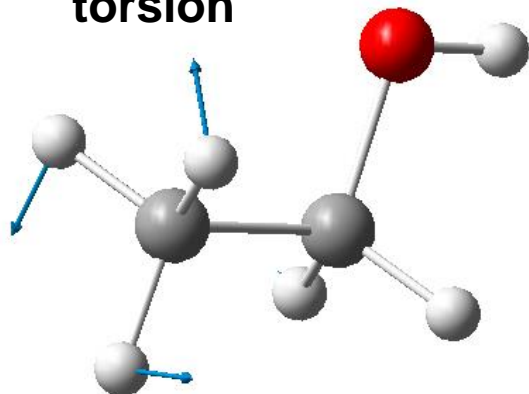
## Several types of vibrational modes in molecules

**stretching**



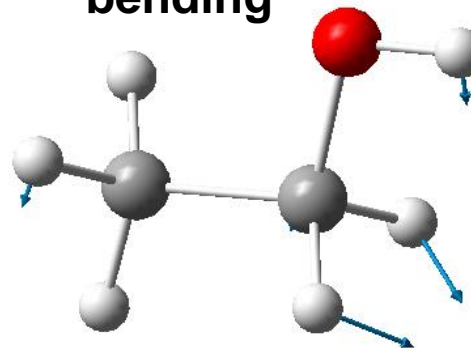
(bond distance changes)

**torsion**



(rotation about bond axis;  
dihedral angle changes)

**bending**



(bond angle changes)

Two quantities define a vibrational mode:

- frequency
- set of atomic displacements

Notice that in a normal mode of vibration all atoms move in phase.

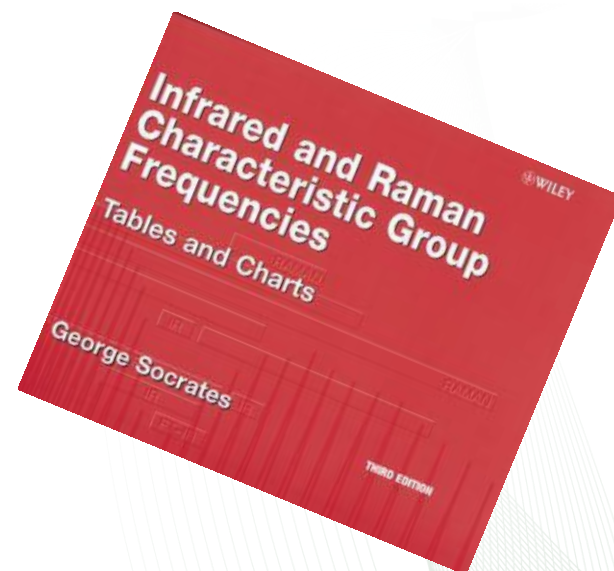
# Vibrational spectroscopy

Molecular vibrations are useful to chemists because:

- they depend on molecular structure and interatomic or intermolecular forces (chemical bonding)
- specific bonds and functional groups are easily identified (analytical tool)

Table 1 Absorption frequencies of some common bonds (shown in bold type)

<i>bond</i>		<i>type of compound</i>	<i>frequency</i>
$\text{—}\overset{\text{H}}{\underset{\text{H}}{\text{C}}}\text{—H}$	(stretch)	alkanes	2800–3000
$\text{=}\overset{\text{H}}{\text{C}}\text{—H}$	(stretch)	alkenes, aromatics	3000–3100
$\text{=}\overset{\text{H}}{\text{C}}\text{—H}$	(stretch)	alkynes	3300
$\text{—O—H}$	(stretch)	alcohols, phenols	3600–3650 (free) 3200–3500 (H-bonded) (broad)
$\text{—O—H}$	(stretch)	carboxylic acids	2500–3300
$\text{—N—H}$	(stretch)	amines	3300–3500 (doublet for $\text{NH}_2$ )
$\text{—}\overset{\text{O}}{\parallel}{\text{C}}\text{—H}$	(stretch)	aldehydes	2720 and 2820
$\text{—}\overset{\text{H}}{\text{C}}=\overset{\text{H}}{\text{C}}\text{—}$	(stretch)	alkenes	1600–1680
$\text{—}\overset{\text{H}}{\text{C}}=\overset{\text{H}}{\text{C}}\text{—}$	(stretch)	aromatics	1500–1600
$\text{—C}\equiv\text{C—H}$	(stretch)	alkynes	2100–2270
$\text{—}\overset{\text{O}}{\parallel}{\text{C}}\text{—}$	(stretch)	aldehyde, ketones, carboxylic acids	1680–1740
$\text{—C}\equiv\text{N}$	(stretch)	nitriles	2220–2260
$\text{C—N}$	(stretch)	amines	1180–1360
$\text{—C—H}$	(bending)	alkanes	1375 (methyl)
$\text{—C—H}$	(bending)	alkanes	1460 (methyl and methylene)
$\text{—C—H}$	(bending)	alkanes	1370 and 1385 (isopropyl split)

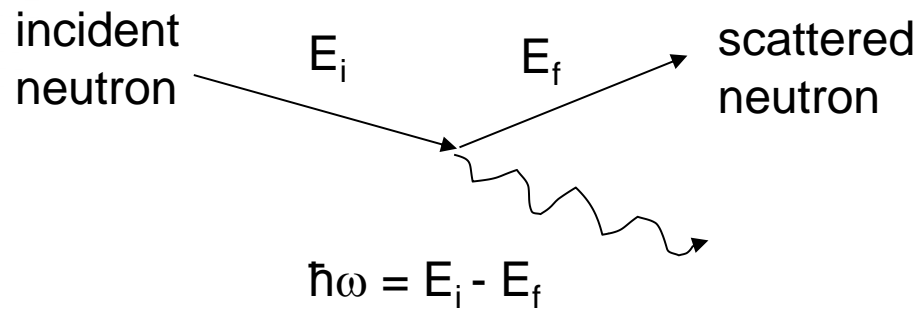


# Vibrational spectroscopy

## How do we observe vibrational modes experimentally ?

Crystallographers use diffraction of some form of radiation (light, electron, x-ray, neutron,...) to obtain information on the periodic arrangement of atoms in space. The wavelength of the radiation is comparable to interatomic distances.

Spectroscopists use (inelastic) scattering of radiation (light, x-ray, neutron,...) to excite vibrational modes. The energy of the radiation is comparable to the energy associated with the vibrational excitations.



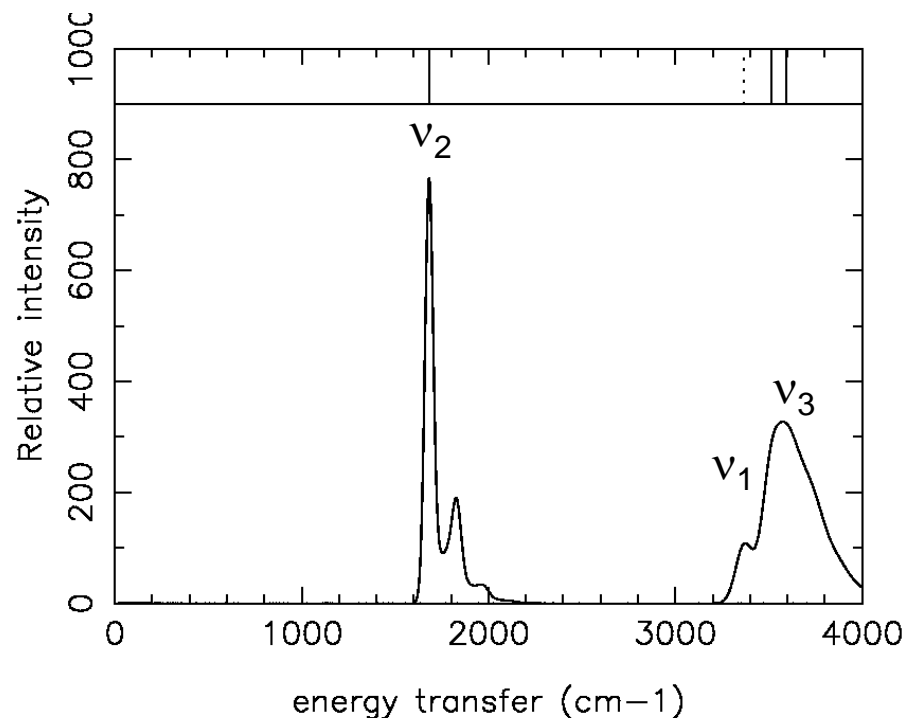
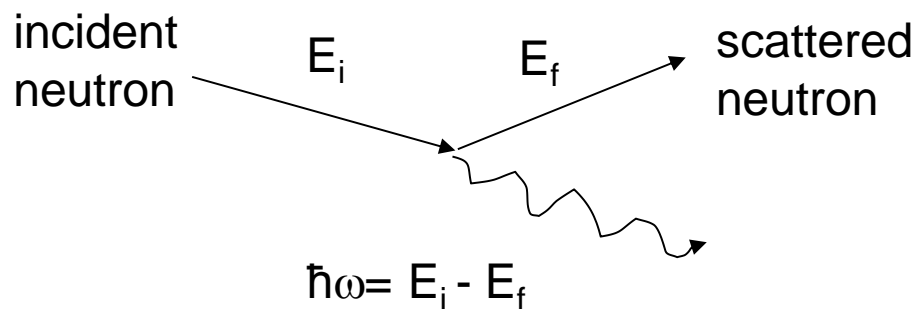
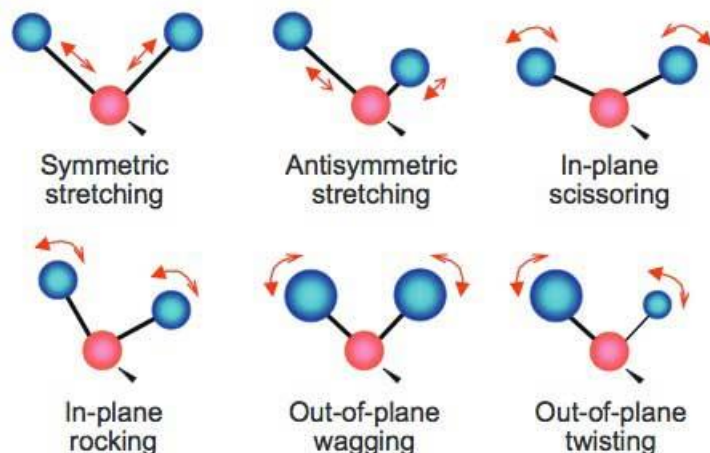
(conservation of energy)

Upon interacting with a vibrational mode, the incident neutron loses energy (from  $E_i$  to  $E_f$ ). The difference in kinetic energy is used to create a vibrational quantum.

Momentum is also exchanged !

# Vibrational spectroscopy

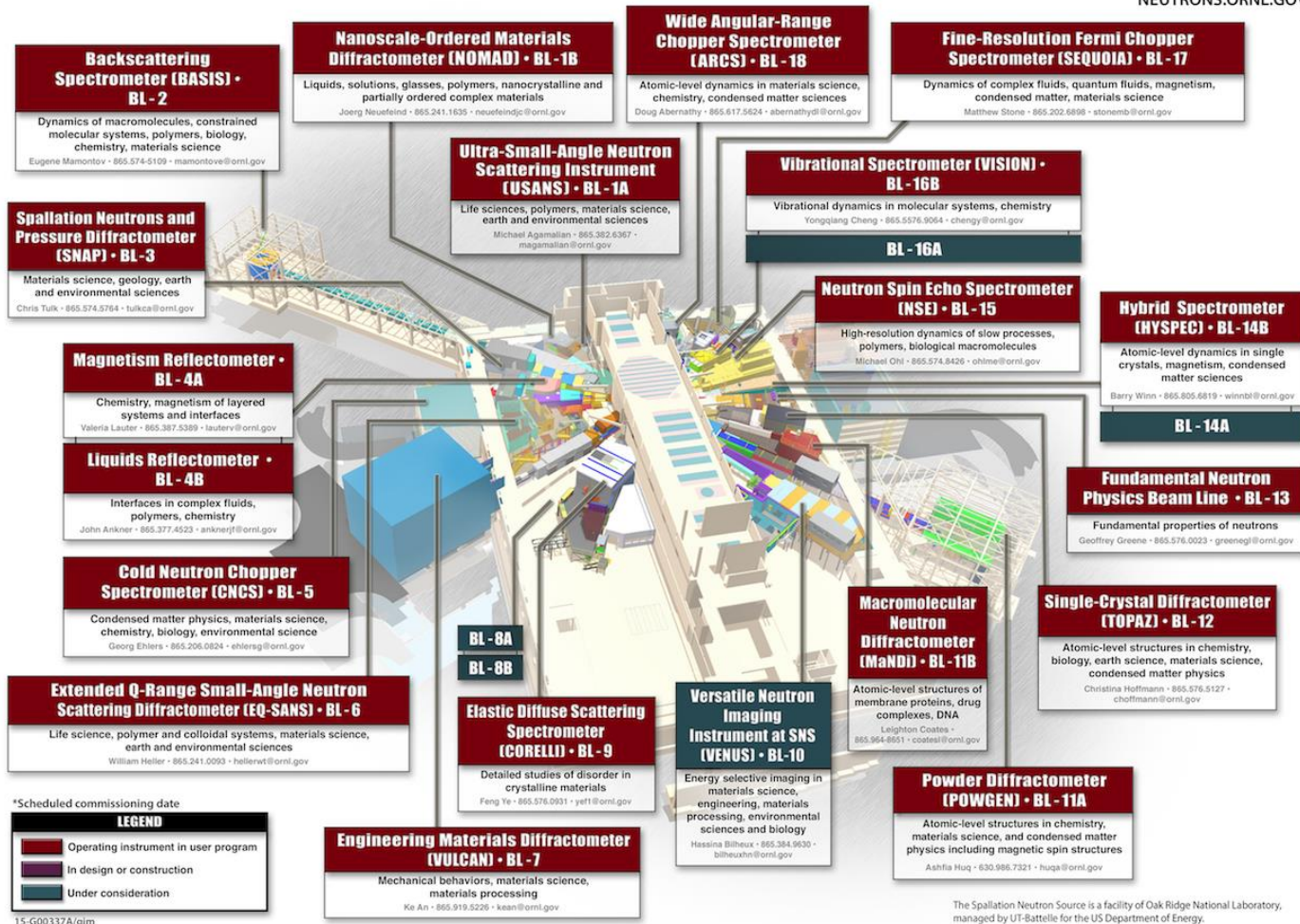
As long as we have a way to determine  $E_i$  and  $E_f$  and the number of particles with energy  $E_i$  and  $E_f$ , we can determine the number of excitations (vibrational modes) created with an energy of  $\hbar\omega = E_i - E_f$ . The result is the vibrational spectrum:



# Vibrational spectroscopy

VISION (INS)	Raman/Infrared
Measures dynamics of nuclei (direct)	Measures response of electrons (indirect)
No selection rules	Selection rules apply
Great sensitivity to H	Cannot always see H
High penetration (bulk probe)	Low penetration (surface probe)
Easy access to low energy range (librational and translational modes)	Low energy cutoff applies (on the order of $100 \text{ cm}^{-1}$ )
Q trajectories in the $(\omega, Q)$ map; averaging over the Brillouin zone	Gamma point only
Weighted by neutron scattering cross section	Weighted by change in polarizability or dipole moment
Easy to simulate/calculate	Difficult to simulate/calculate
No energy deposition in sample	Heating, photochemistry, ...

# SNS Instrument Suite



15-G00337A/gjm

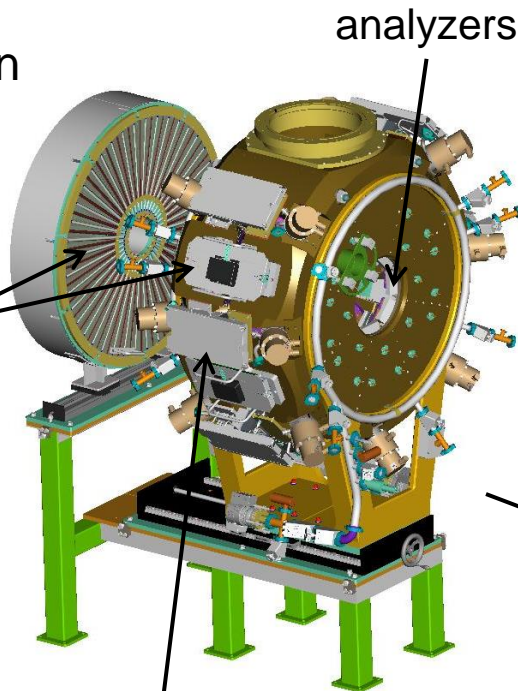
The Spallation Neutron Source is a facility of Oak Ridge National Laboratory, managed by UT-Battelle for the US Department of Energy.

# VISION

neutron beam

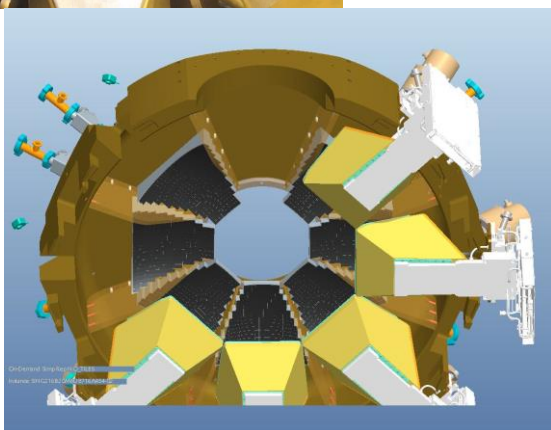
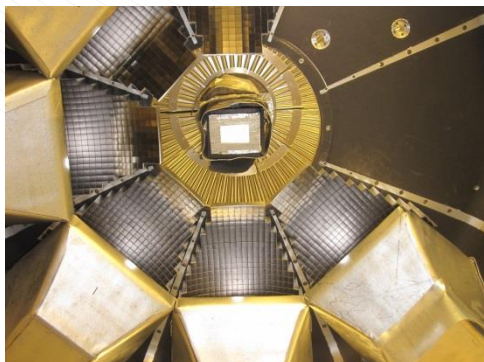
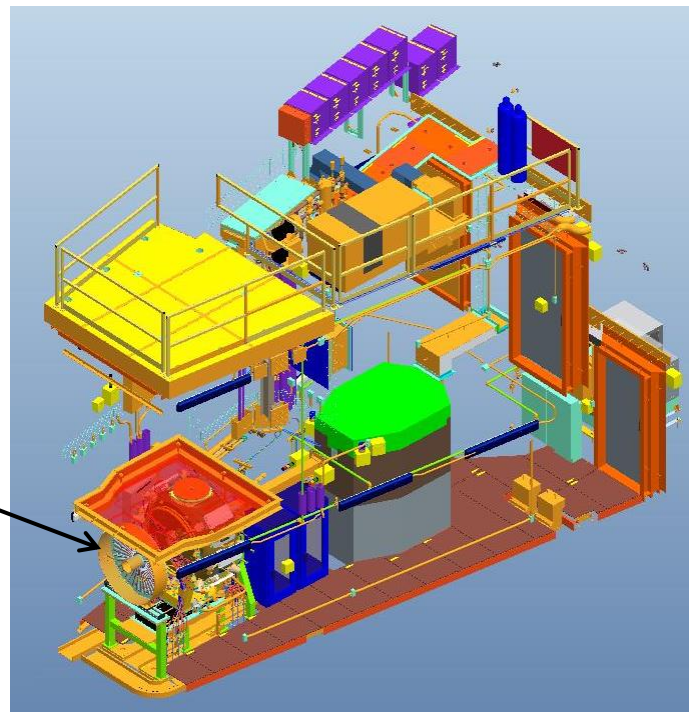


diffraction detectors



analyzers

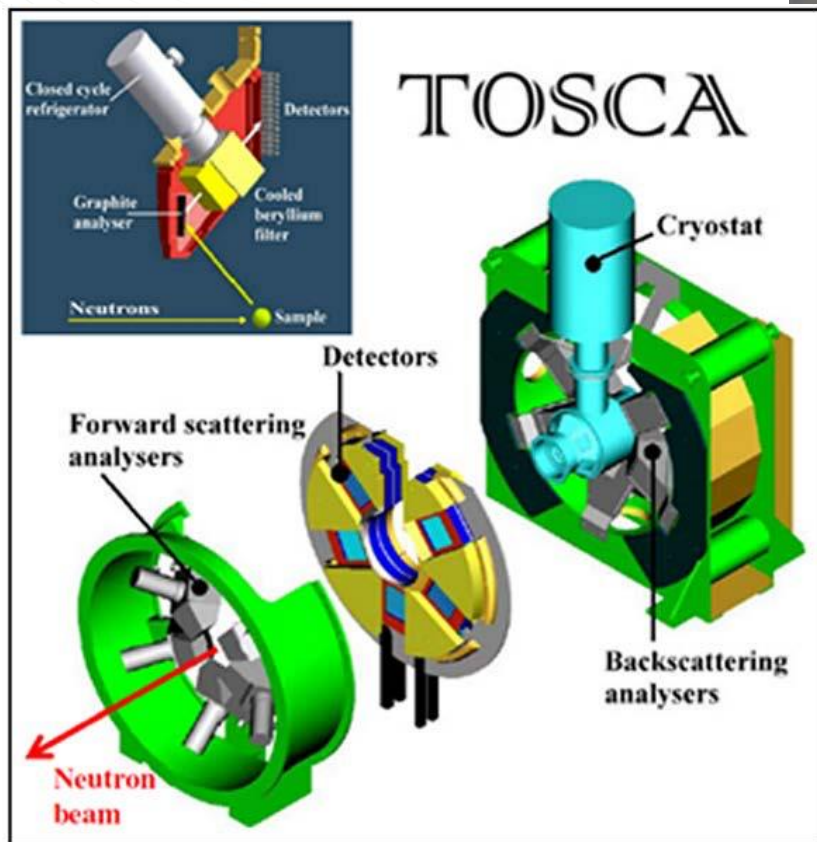
inelastic detectors



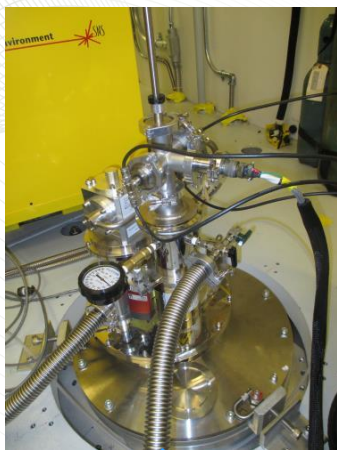
- Vibrational spectroscopy with neutrons
- Beam line started commissioning 3 years ago
- **Multifunctional beam line: simultaneous spectroscopy and diffraction**
- Dynamic range: 0-1000 meV; resolution:  $< 1.5\%$
- Diffraction:  $1.5 - 30 \text{ \AA}^{-1}$
- Temperature range: 5-700K
- Sample environment: high pressure, electric field, gas loading, ...
- Great sensitivity to hydrogen, no selection rules, penetration through matter, ...



# TOSCA



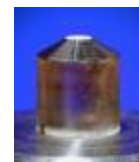
# Sample environment



JANIS closed-cycle refrigerator (5-600K)



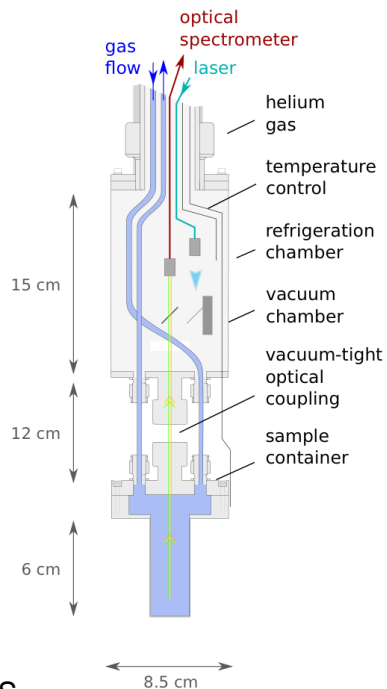
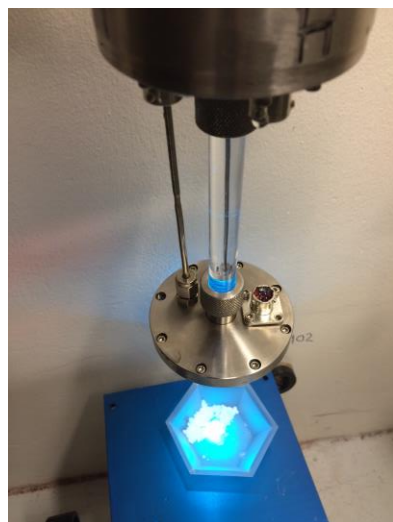
ortho/para H<sub>2</sub> converter



Largest single crystal diamond for DAC!



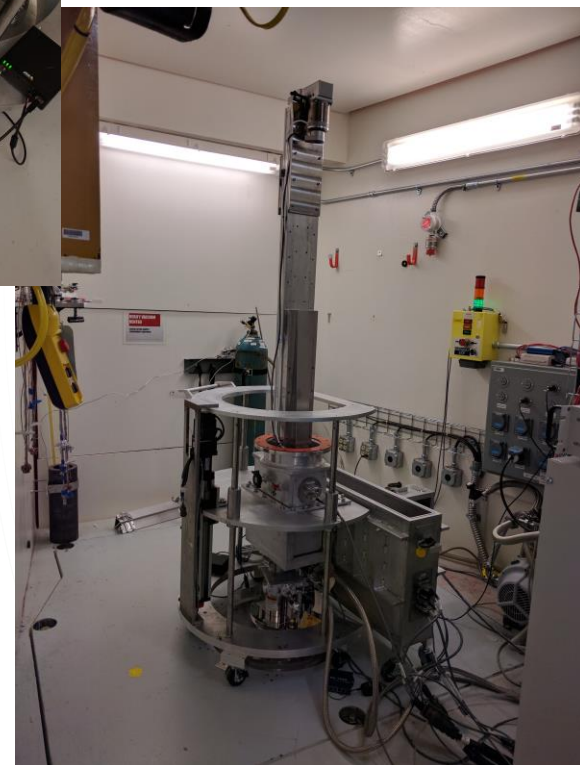
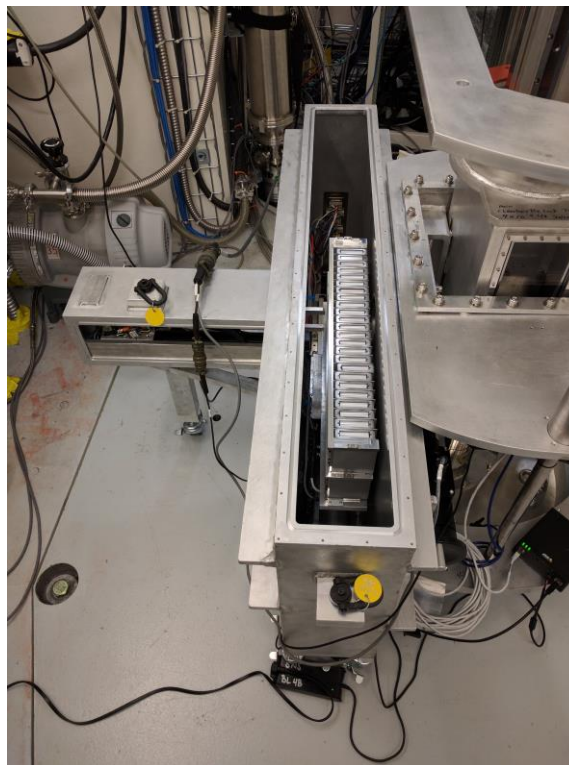
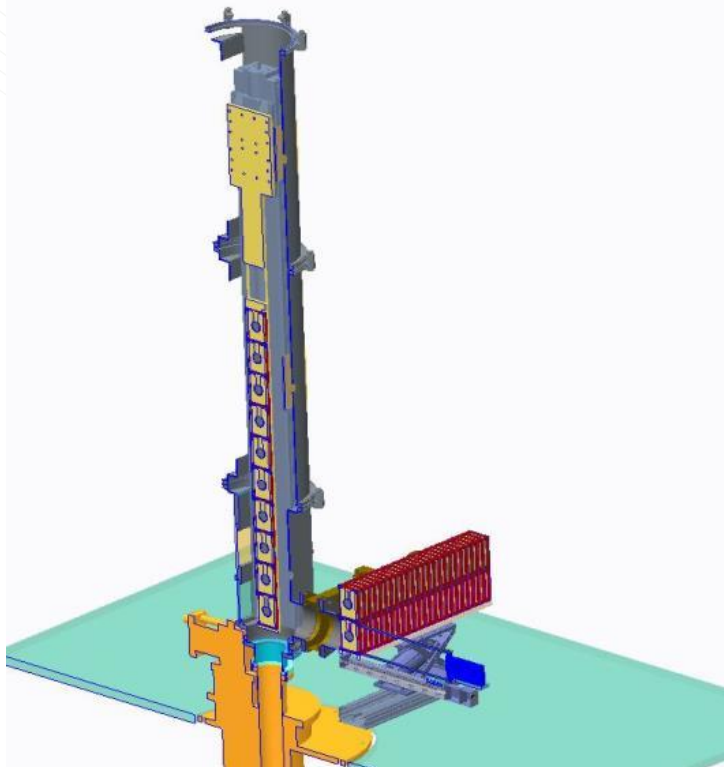
in situ electrochemical impedance spectroscopy (EIS)





# VISION Sample changer

The high throughput rate of VISION requires very rapid sample changes to make the best use of neutron beamtime and run mail-in program. A sample changer has been tested in January 2017 and will be commissioned this cycle.



# High Throughput

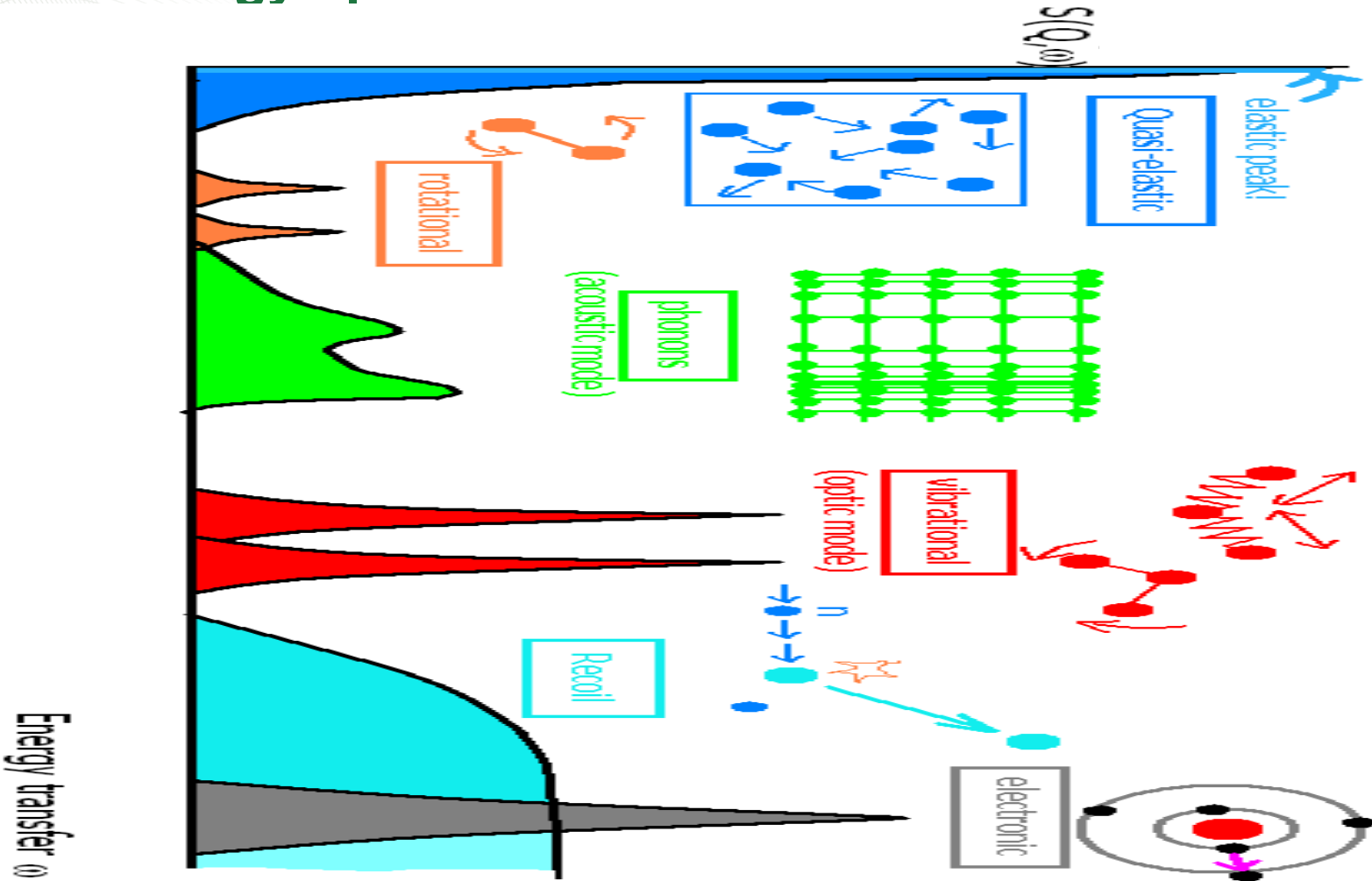
## Challenges

- Large volume of Data
- How to handle large number of samples
  - Sample changers
  - Sample environment
  - Gas handling
- How to model and interpret the results

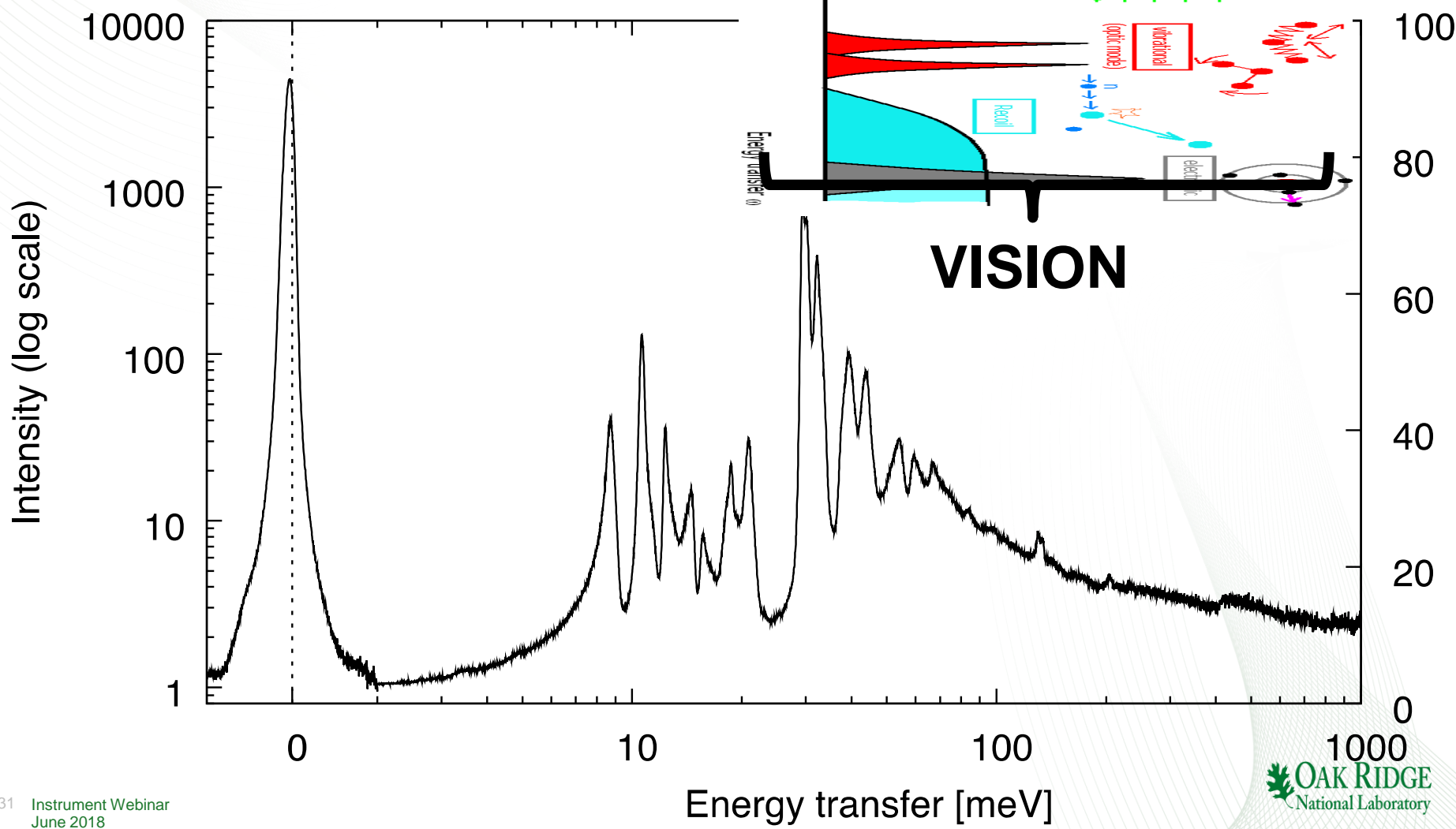
## Opportunities

- Databases and Libraries
- Parametric studies
- Kinetic studies
- In-situ studies
- Small signals in large backgrounds
- Modeling

# The energy spectra

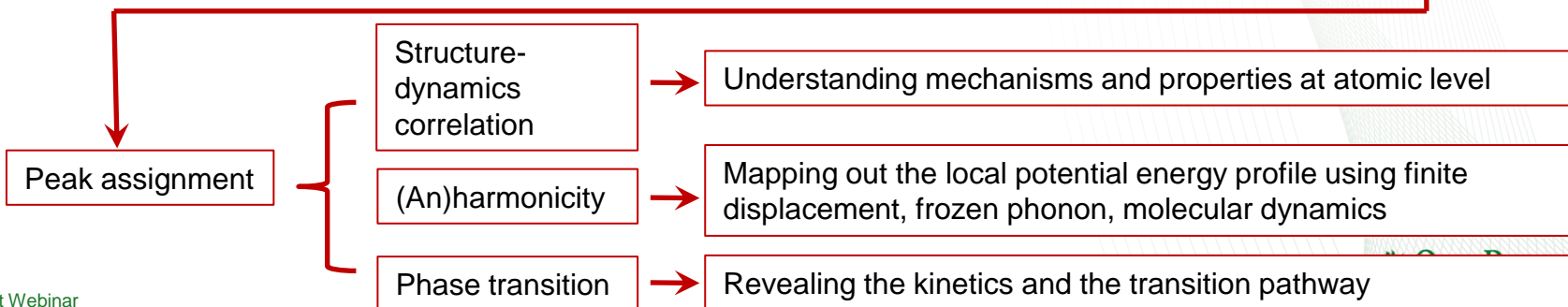
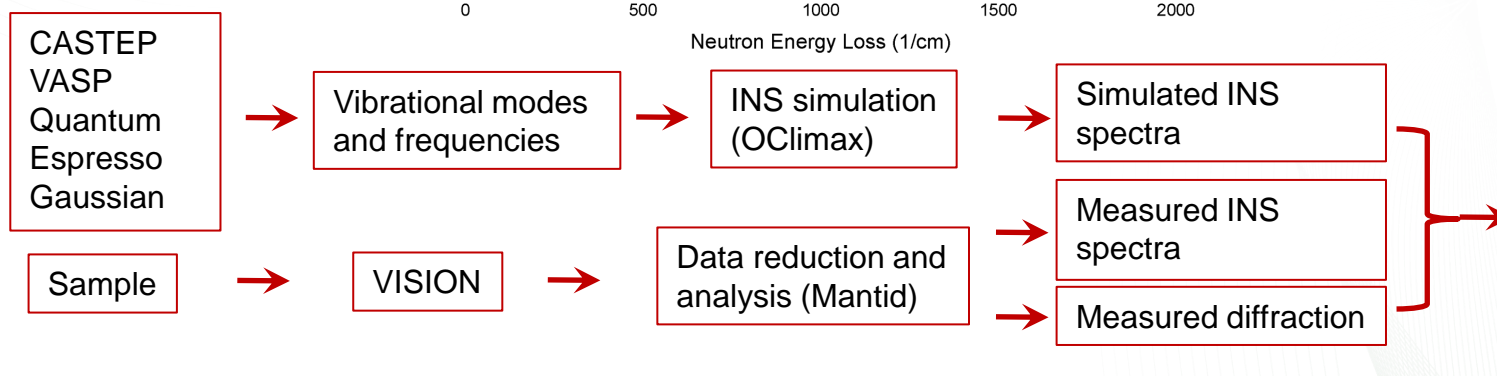
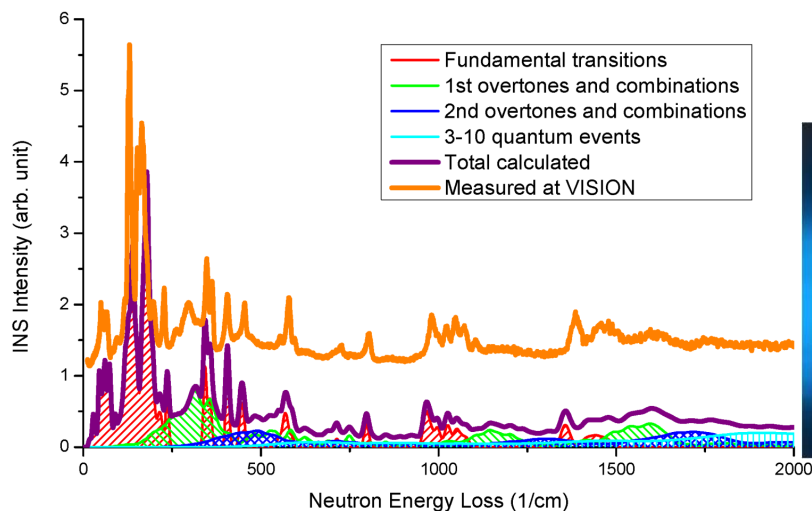
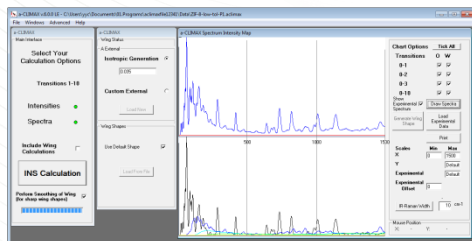


# The energy spectra (according to VISION)



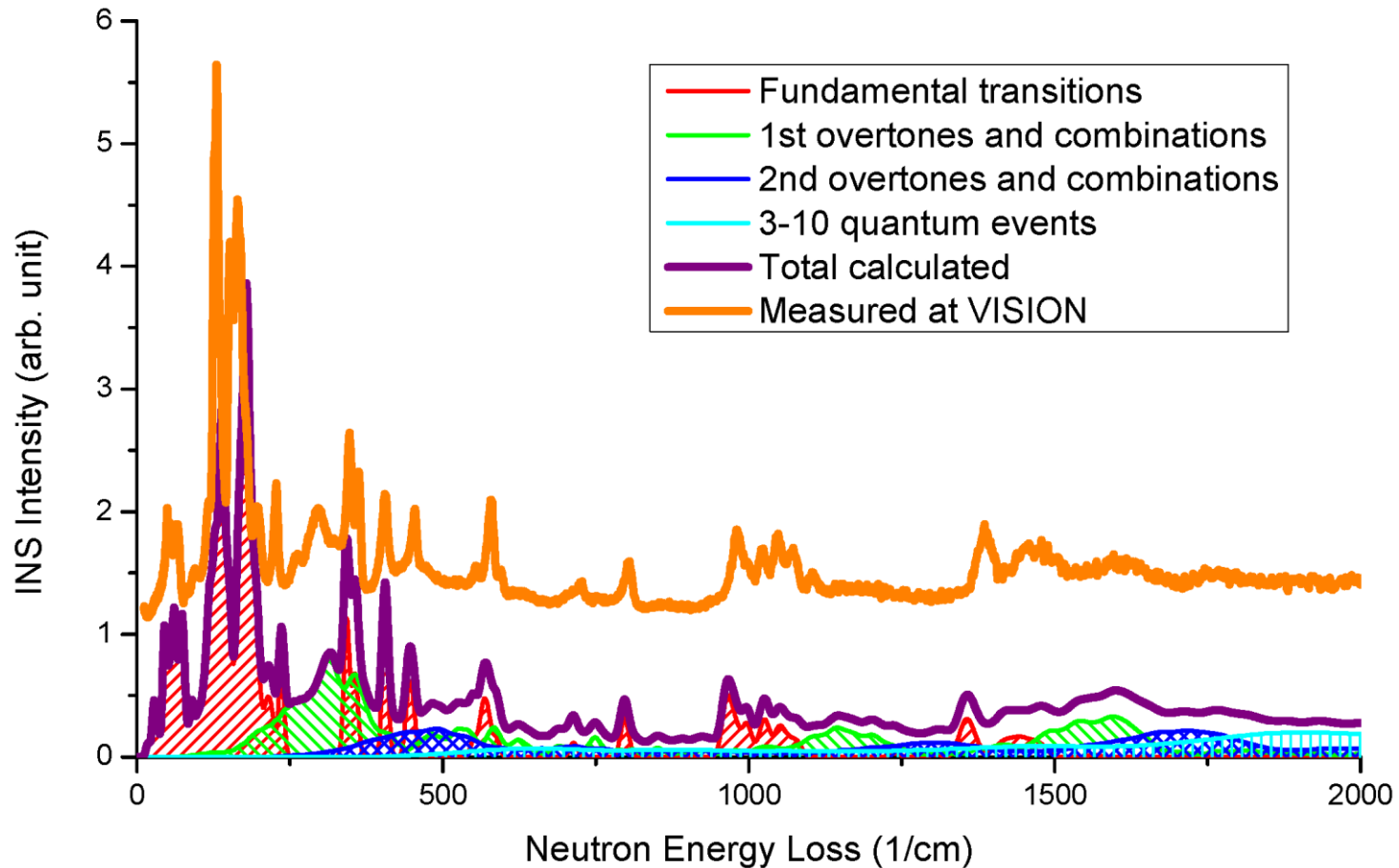
# Integrated modeling for data interpretation

Today this is what we do at **VISION** using **VirtuES** (Virtual Experiments in Spectroscopy)  
1600 cores cluster dedicated to VISION



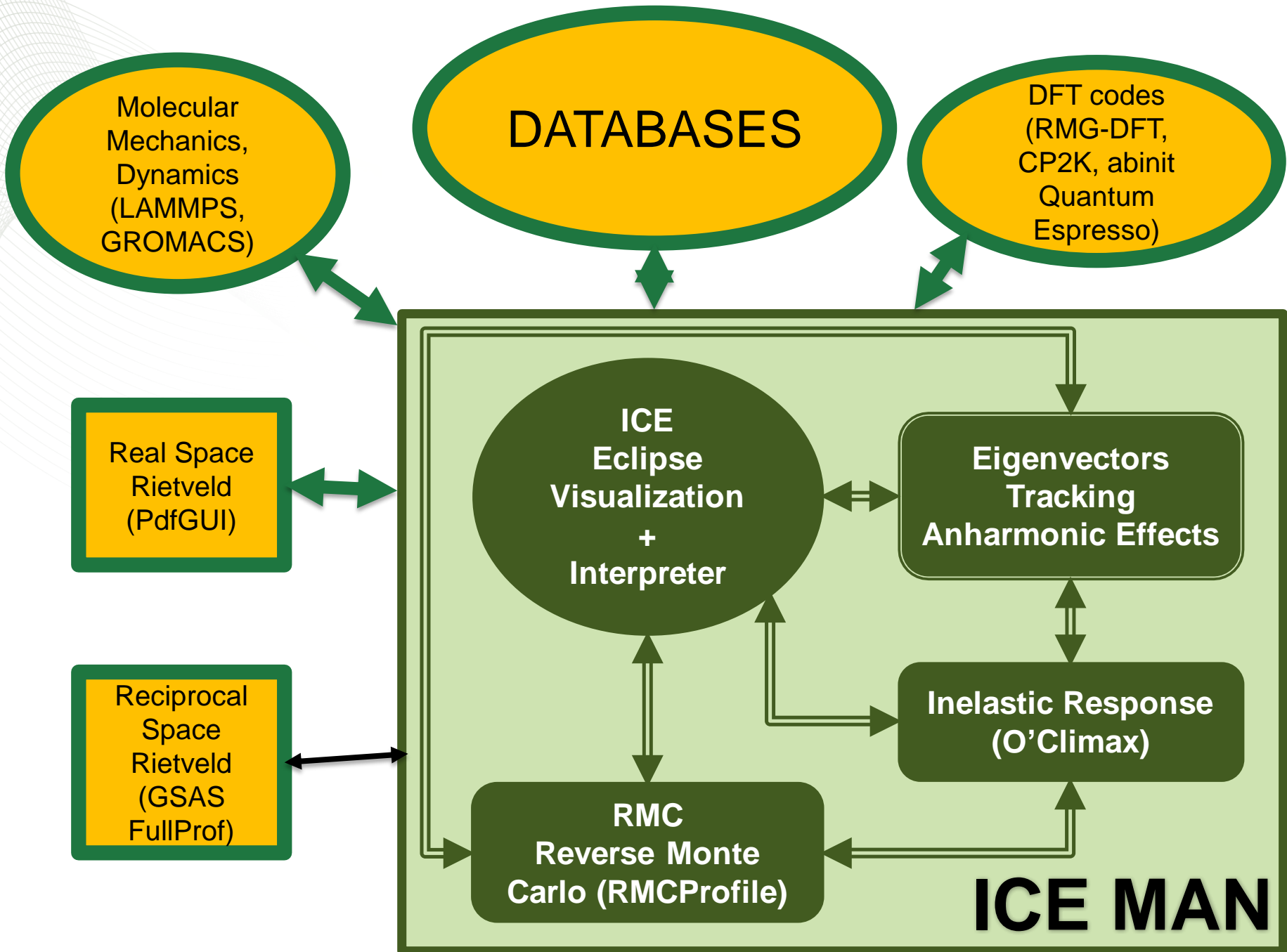


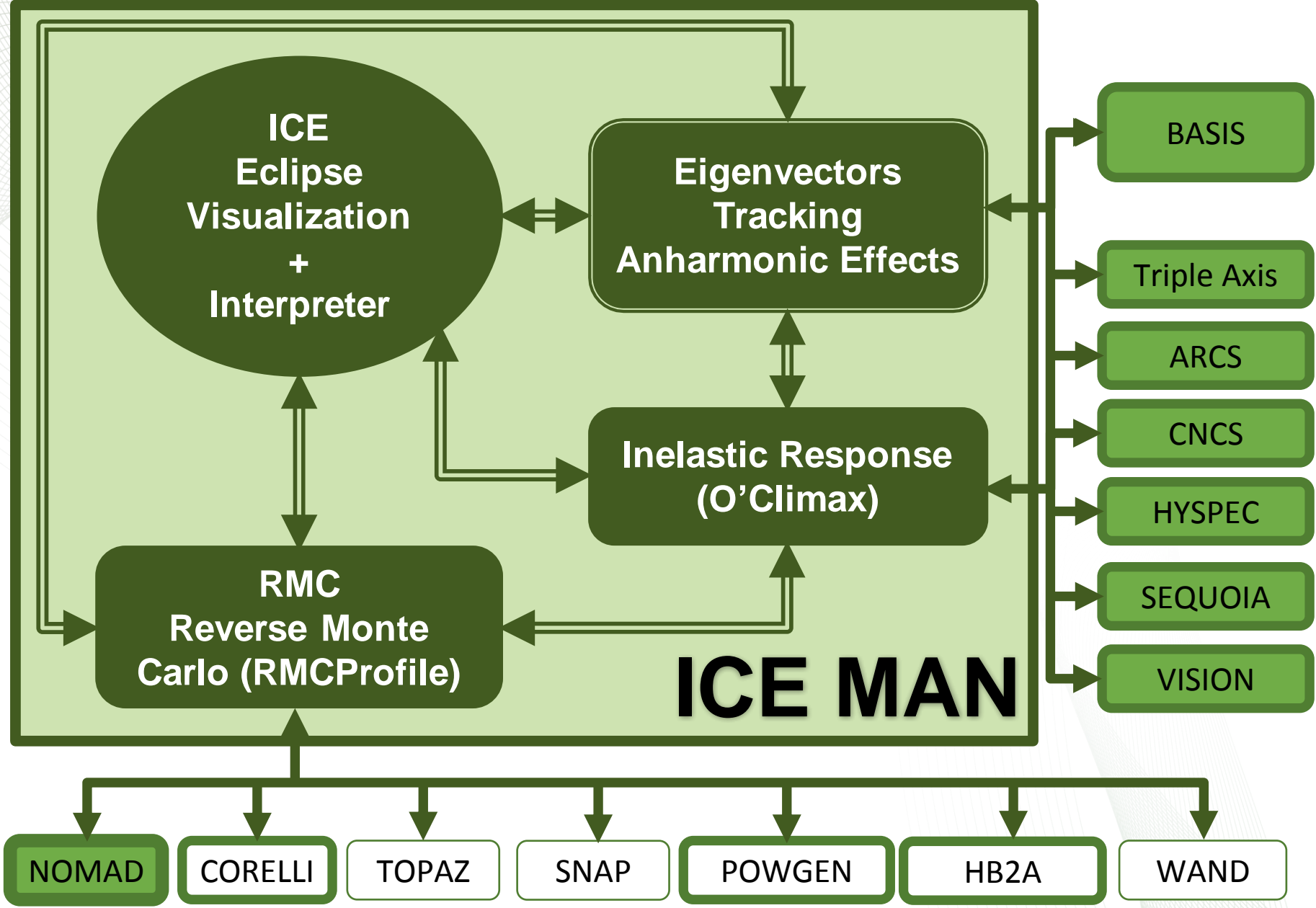
# Comparison



sample is hexamethylbenzene







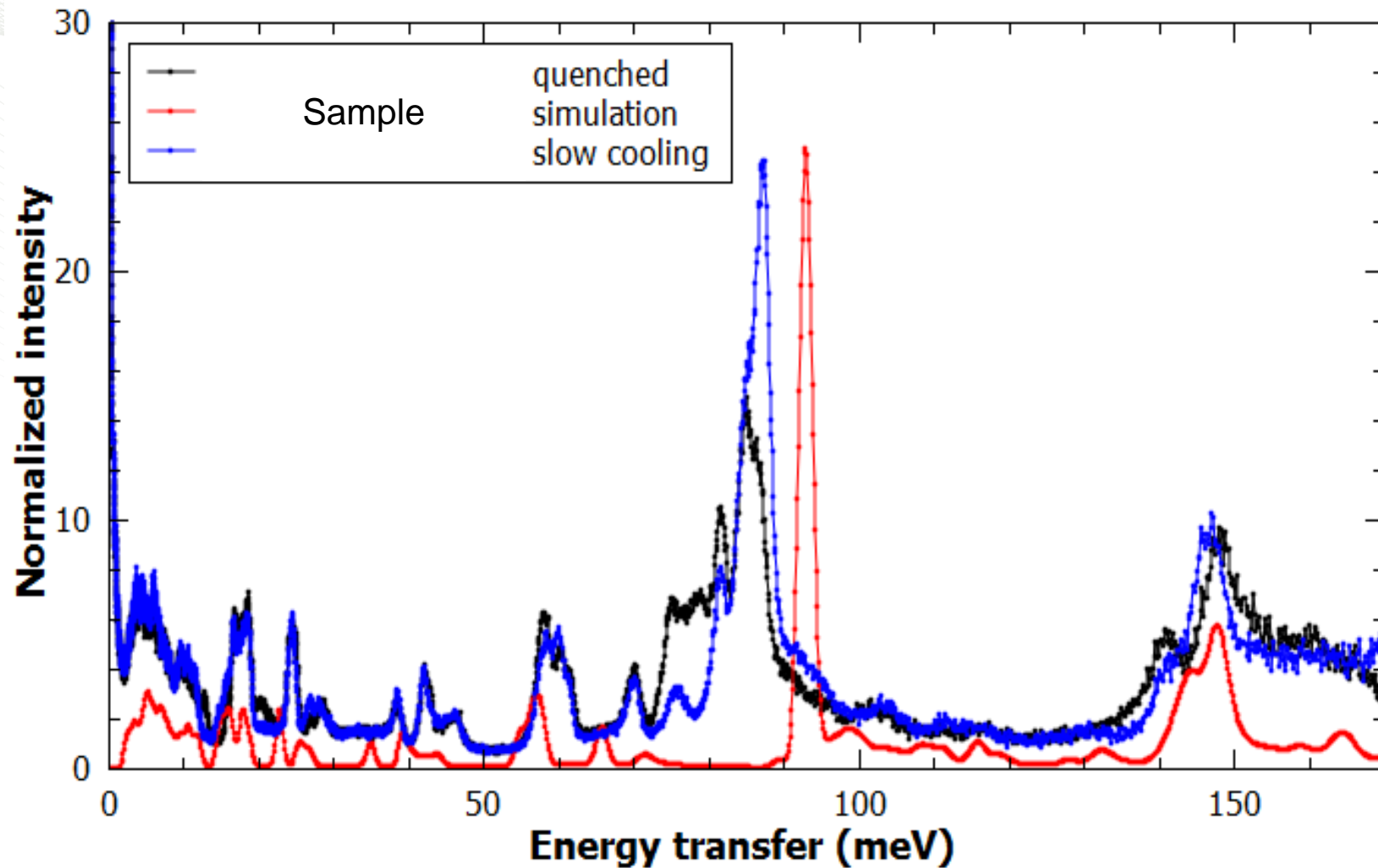
# VirtuES helped users to make decisions on-the-fly

```
[yyc@or-condo-login02 CF3SO2OH]$ ls -lhtr
-rw-r--r-- 1 yyc users 3.6K Nov  4 15:50 cell
-rw-r--r-- 1 yyc users 1.1K Nov  4 15:50 param
-rw-r--r-- 1 yyc users 3.9K Nov  4 15:51 _PhonDOS.cell
-rw-r--r-- 1 yyc users  735 Nov  4 15:52 _PhonDOS.param
-rw-r----- 1 yyc users 1.1M Nov  4 16:46 castep
-rw-r----- 1 yyc users 7.3M Nov  5 06:15 _PhonDOS.phonon
-rw-r----- 1 yyc users 232K Nov  5 06:15 _PhonDOS.castep
-rw-r--r-- 1 yyc users 3.3M Nov  5 08:56 aclimax
```

```
[yyc@analysis-node02 manualreduce]$ ls -lhtr
-rw-rwx---+ 1 yyc users 2.2M Nov  5 12:34 VIS_20557_5K_for_0.9hr.nxs
-rw-rwx---+ 1 yyc users 2.2M Nov  5 13:28 VIS_20559_50K_for_0.9hr.nxs
-rw-rwx---+ 1 yyc users 2.2M Nov  5 14:23 VIS_20561_75K_for_0.9hr.nxs
-rw-rwx---+ 1 yyc users 2.2M Nov  5 15:56 VIS_20563_100K_for_0.9hr.nxs
-rw-rwx---+ 1 yyc users 2.2M Nov  5 17:21 VIS_20565_125K_for_0.9hr.nxs
-rw-rwx---+ 1 yyc users 2.2M Nov  5 18:44 VIS_20567_150K_for_0.9hr.nxs
-rw-rwx---+ 1 yyc users 2.2M Nov  5 20:23 VIS_20570_175K_for_1.2hr.nxs
-rw-rwx---+ 1 yyc users 2.2M Nov  5 21:58 VIS_20572_200K_for_1.2hr.nxs
-rw-rwx---+ 1 yyc users 2.2M Nov  5 23:29 VIS_20574_225K_for_1.2hr.nxs
-rw-rwx---+ 1 yyc users 2.2M Nov  6 01:00 VIS_20576_250K_for_1.2hr.nxs
-rw-rwx---+ 1 yyc users 2.2M Nov  6 02:28 VIS_20578_275K_for_1.2hr.nxs
-rw-rwx---+ 1 yyc users 2.2M Nov  6 03:57 VIS_20580_300K_for_1.2hr.nxs
```

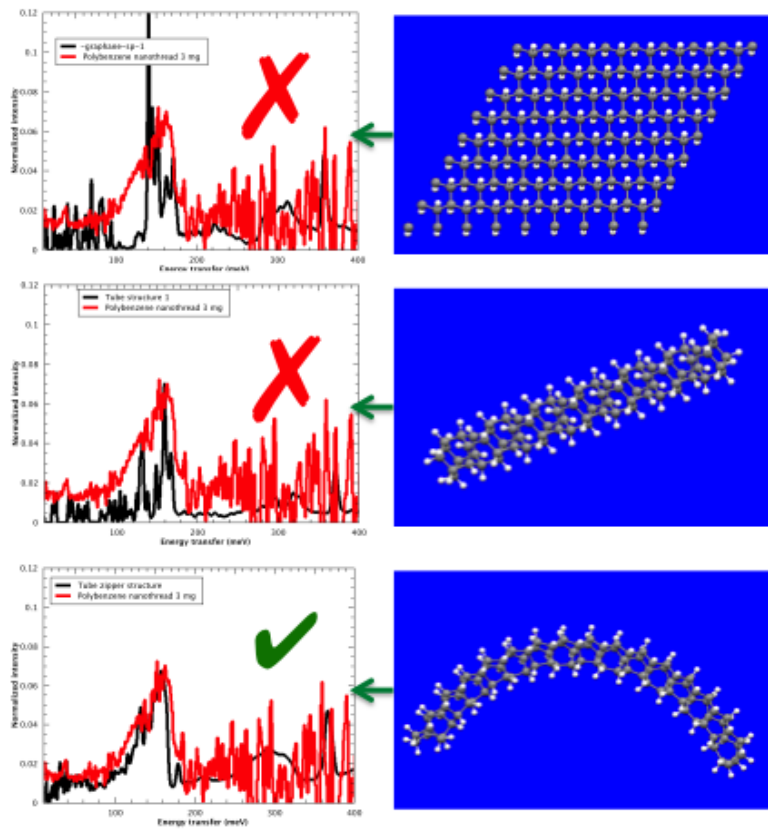
Simulation was started at the beginning of the experiment. By the time when experimental data were collected, the calculation was already finished with theoretical predication available to be compared with experiment. This eventually led to a critical decision made by the user (see next slide).

# VirtuES helped users to make decisions on-the-fly

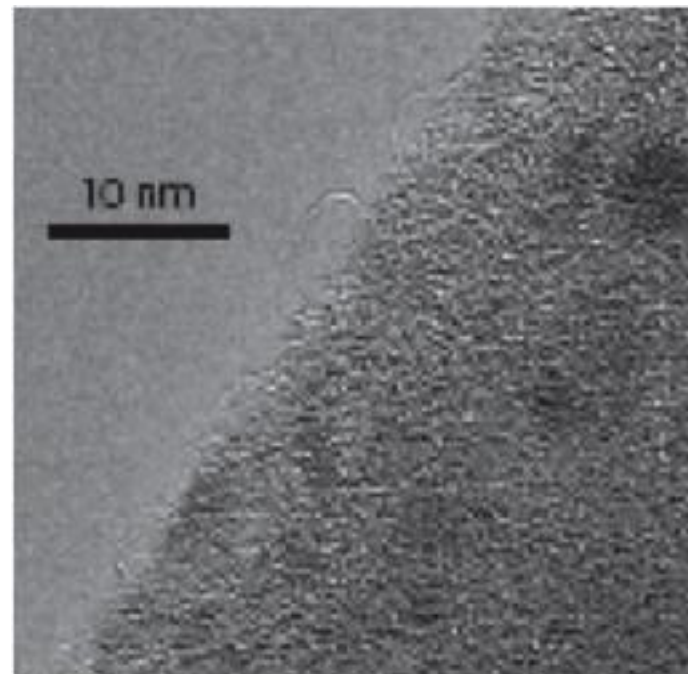


# VISION: Unprecedented capabilities and opportunities

## 3 mg of nanothread sample



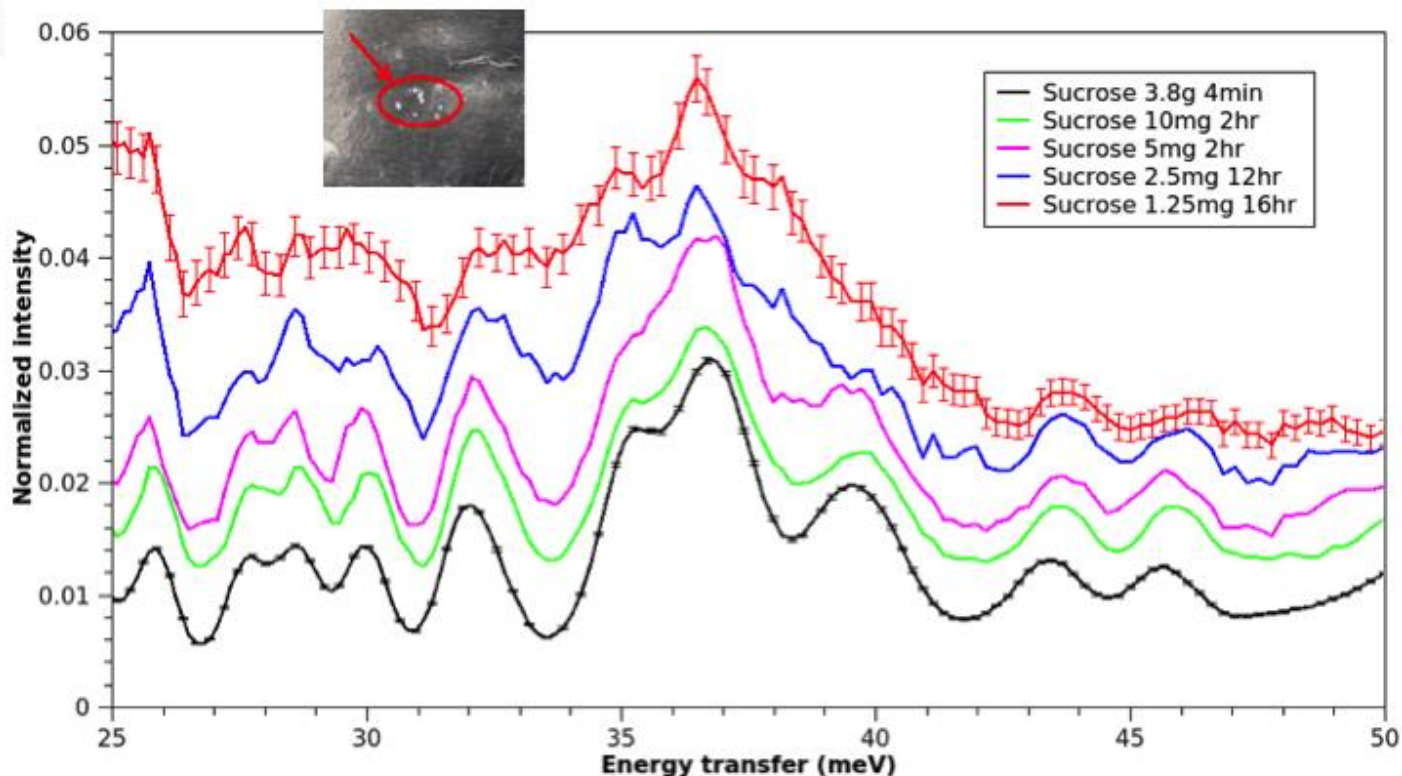
Comparison of the experimental data from VISION and a series of DFT calculations of hypothetical structures that contain  $sp^3$  carbon and the correct stoichiometry (C:H ratio 1:1) allows us to determine which structure corresponds to the measured spectra.



Collaboration with Malcolm Guthrie, John Badding, Vin Crespi. Original publication on carbon nanothreads: *Nature Materials*, **14**, 43 (2014)

# High sensitivity: milligrams of samples

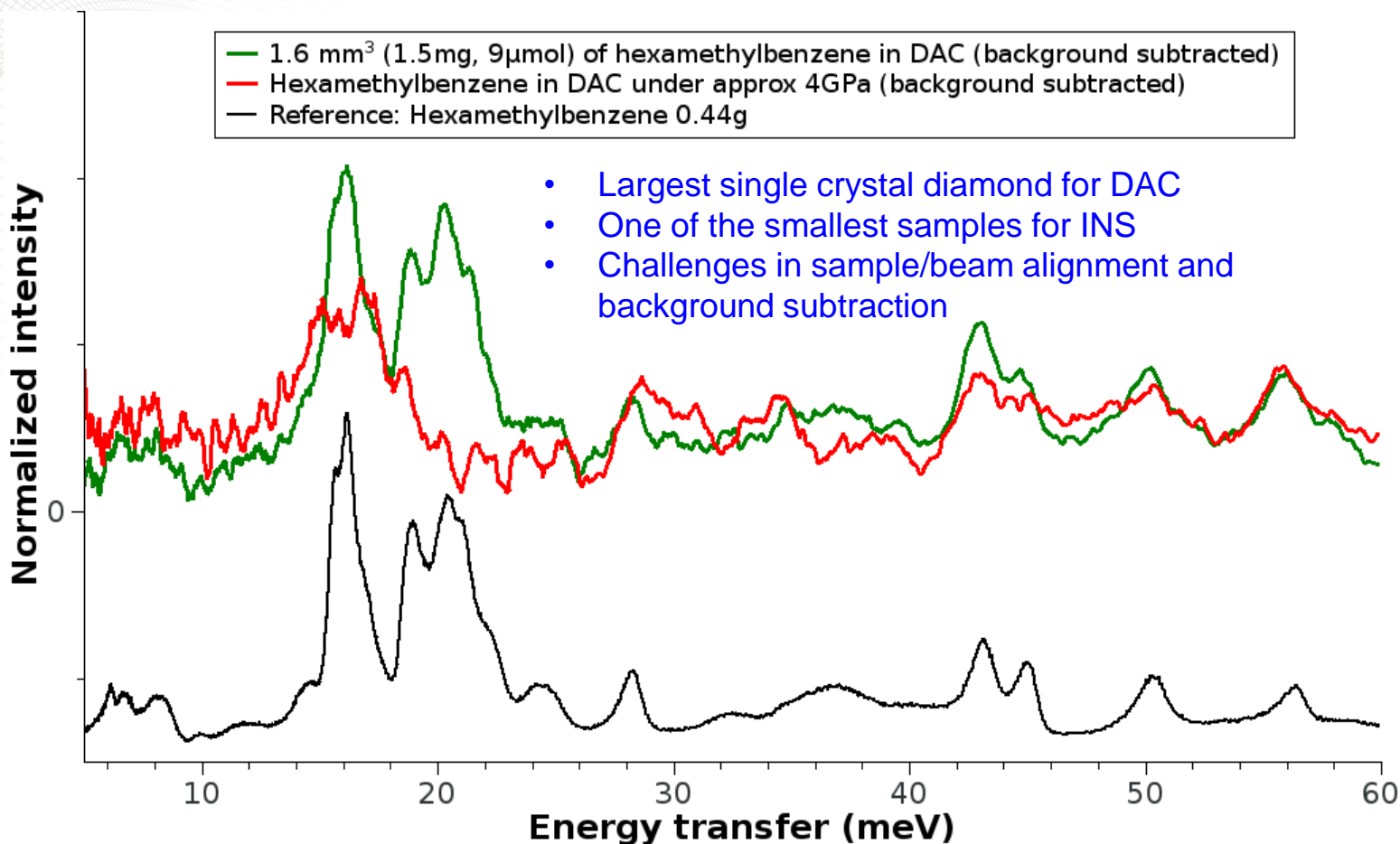
## 1.25 mg of table sugar



Extraordinary sensitivity, this is the smallest amount of sample ever measured using INS. Diamond anvil cells will be used in VISION

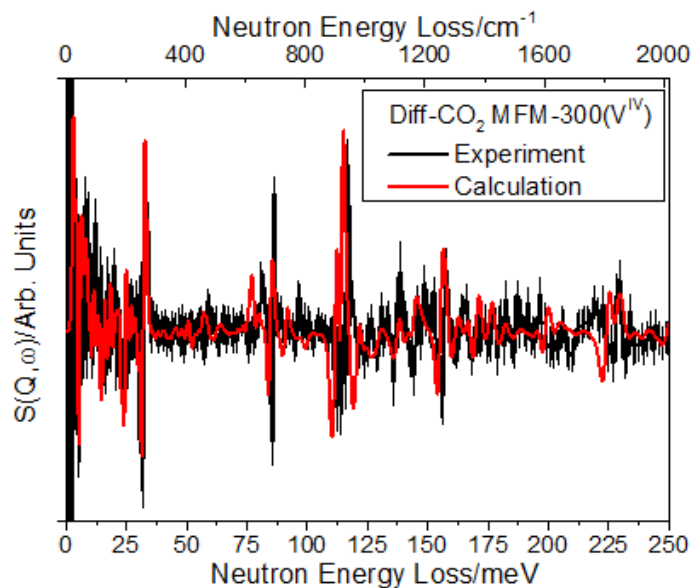
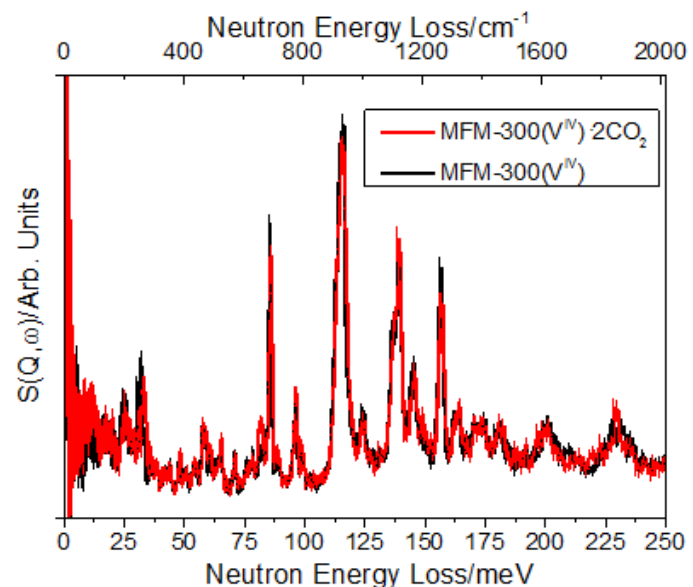
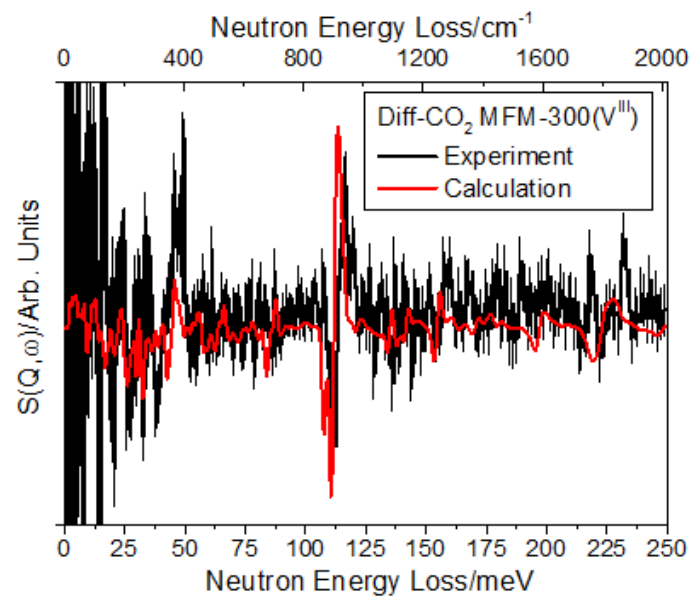
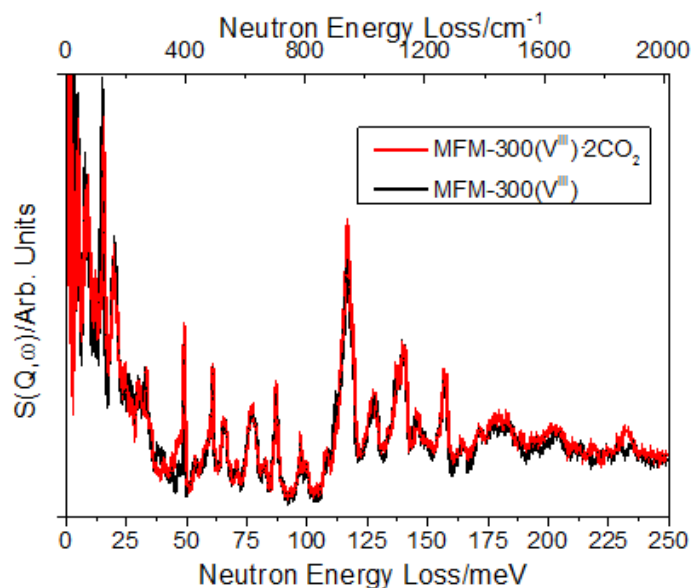
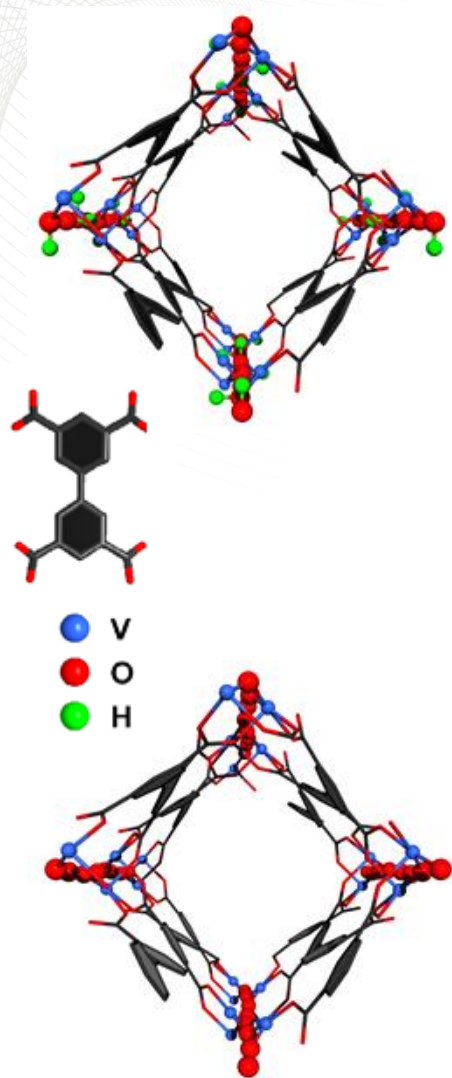


# A successful proof-of-principle test at VISION: using diamond anvil cell (DAC) for high pressure INS experiments



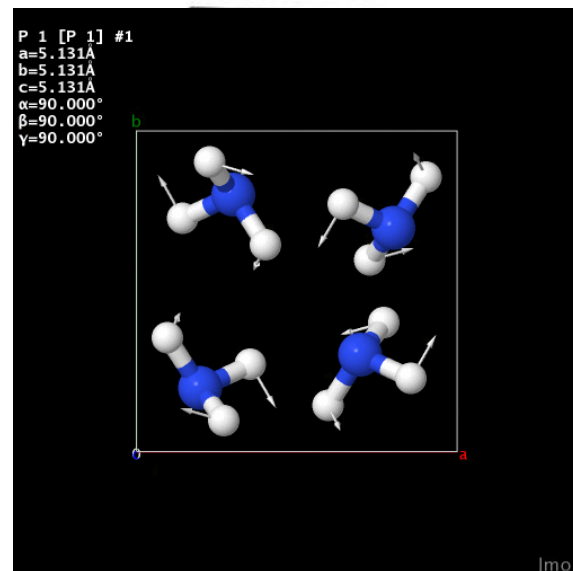
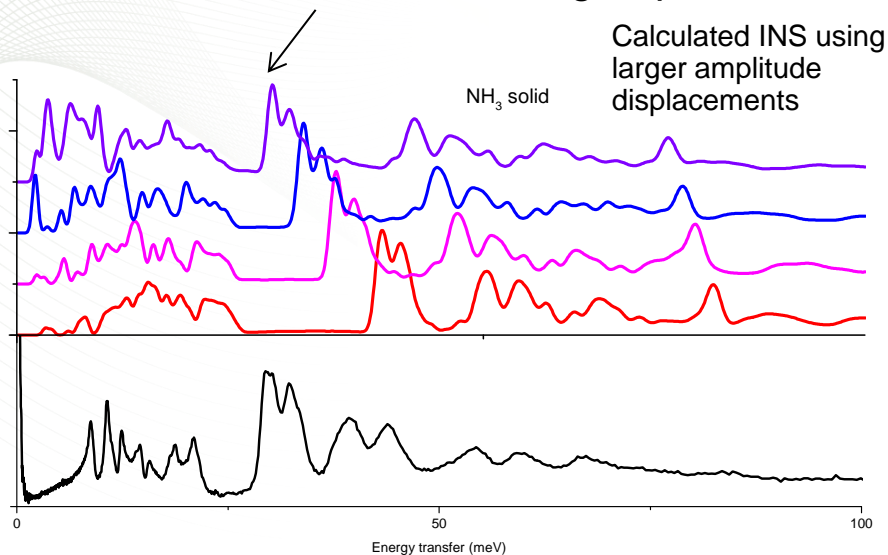
- INS spectrum from 1.6mm<sup>3</sup> (1.5mg, 9μmol) sample loaded in the DAC was successfully extracted, with significant details retained.
- Approx. 4GPa pressure was applied, leading to major changes in the spectrum.
- The unprecedented capability will open the door to many new areas using INS to study materials dynamical behavior under high pressure.

# NOTT-V MOF and CO<sub>2</sub> adsorption

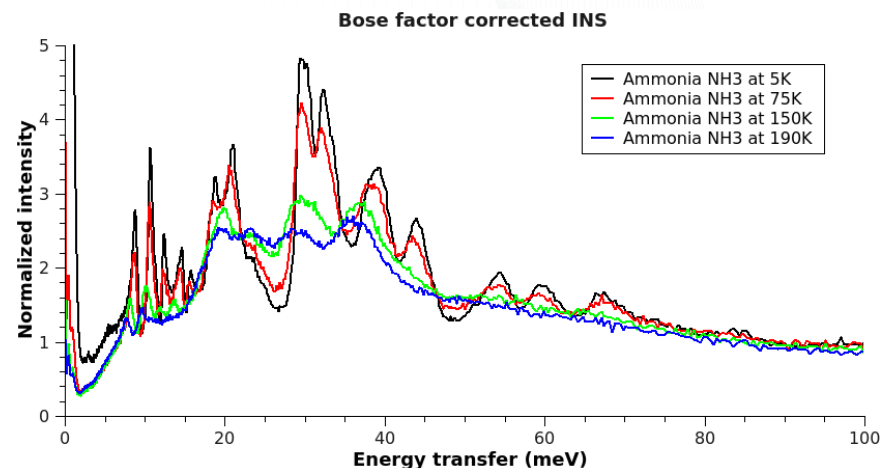
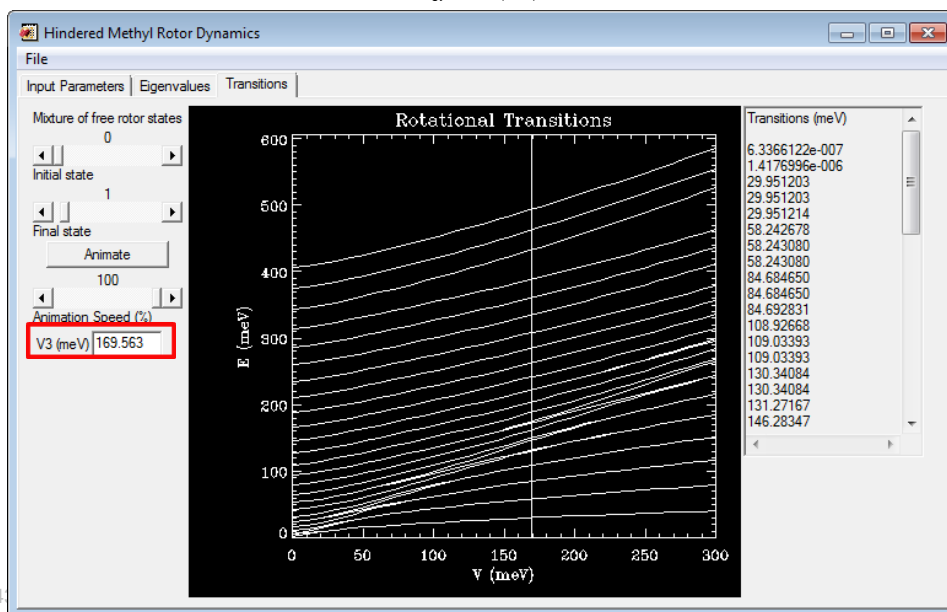


# Hydrogen in a “simple” molecular solid: Beyond DFT and harmonic approximation

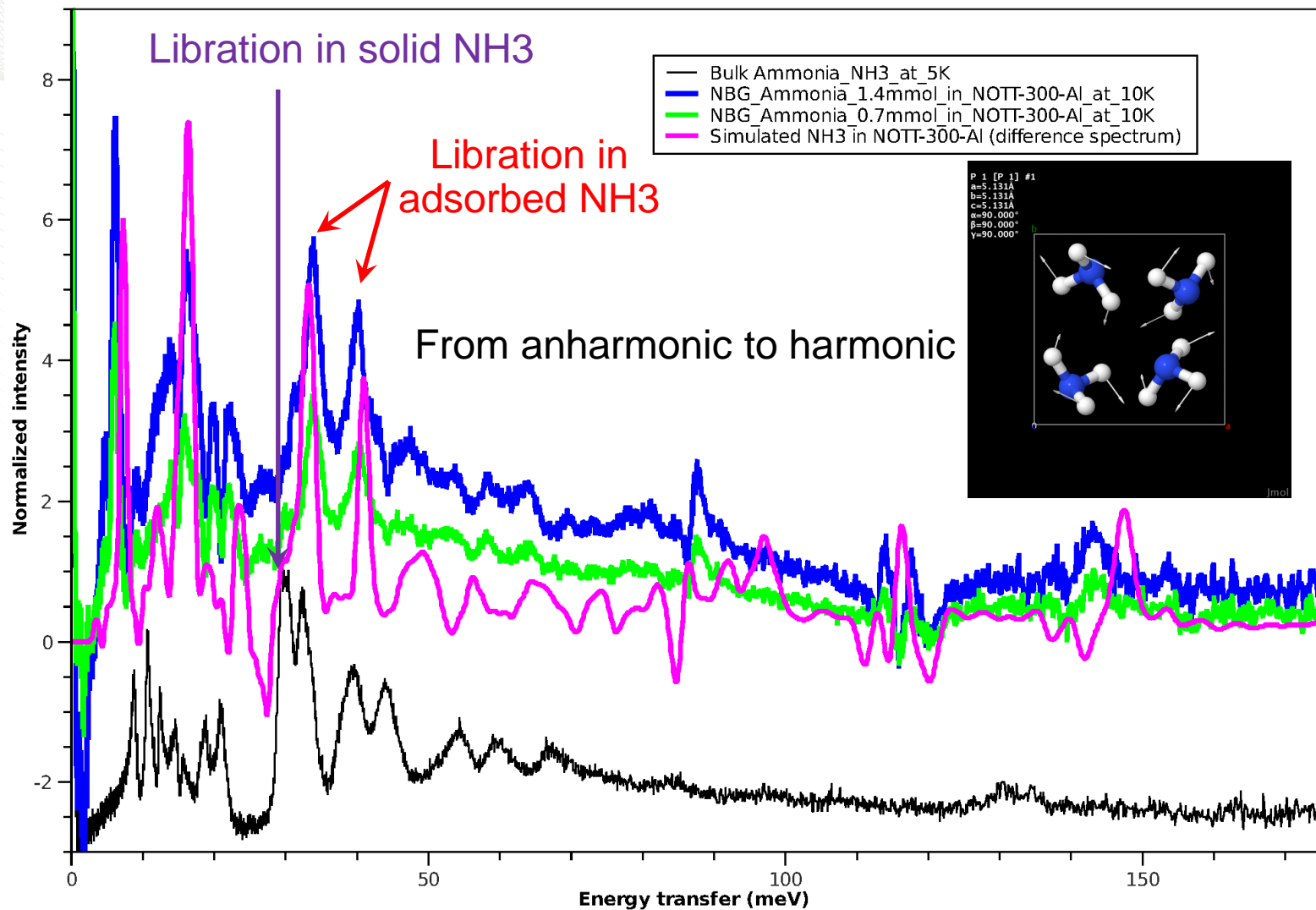
Libration/rotation of NH<sub>3</sub> group



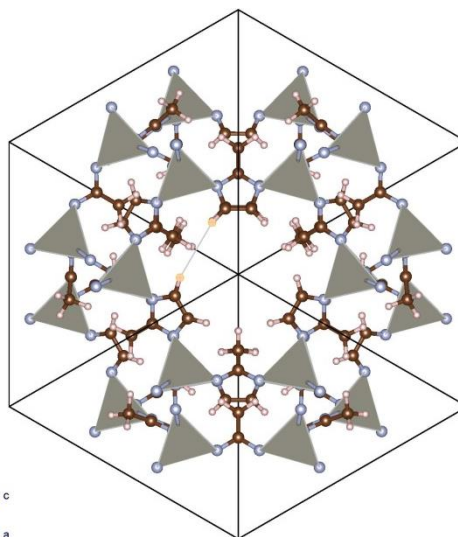
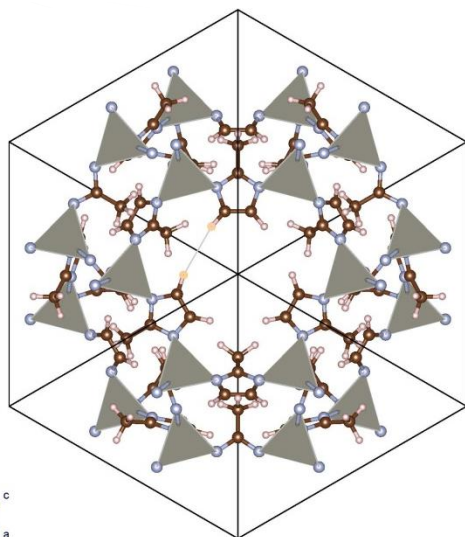
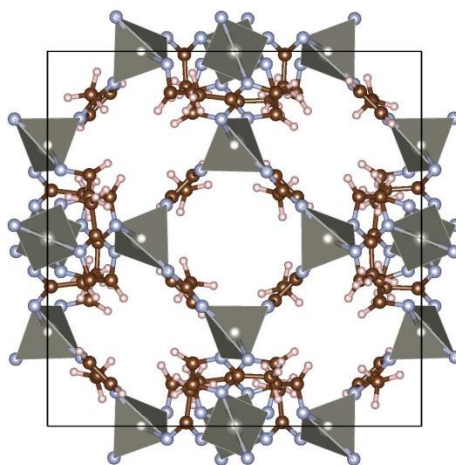
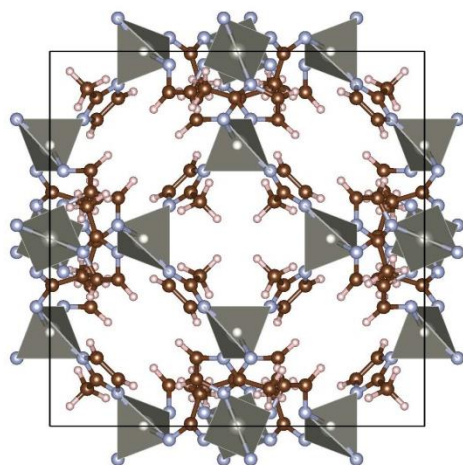
- DFT calculated energy barrier for rigid rotation of NH<sub>3</sub>: 180 meV
- Energy barrier solved from the rotor model : 170 meV



# Solid NH<sub>3</sub> vs NH<sub>3</sub> in MOF



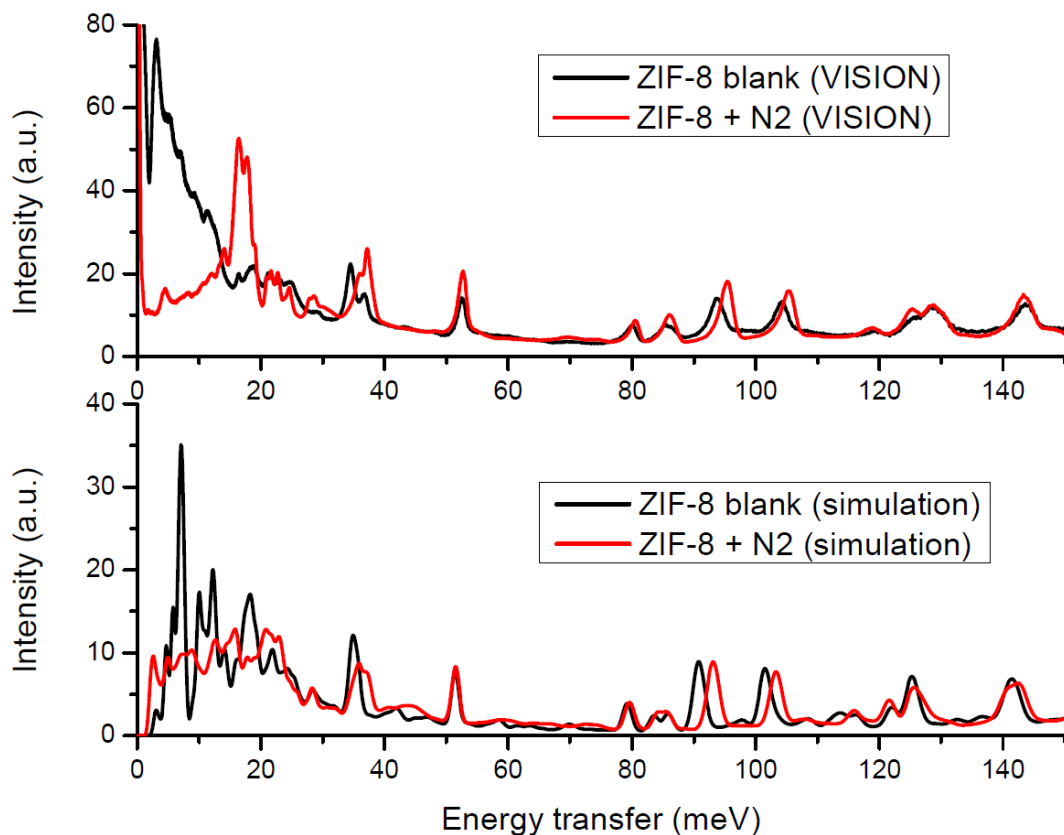
# Gate-opening in a metal-organic framework



Structure of blank ZIF-8 and ZIF-8 loaded with N<sub>2</sub>. The rotation of the methyl groups and the swinging of the imidazolate rings associated with the gate opening can be seen by comparing the marked areas

For clarity the N<sub>2</sub> molecules are not shown.

# Gate-opening in a metal-organic framework

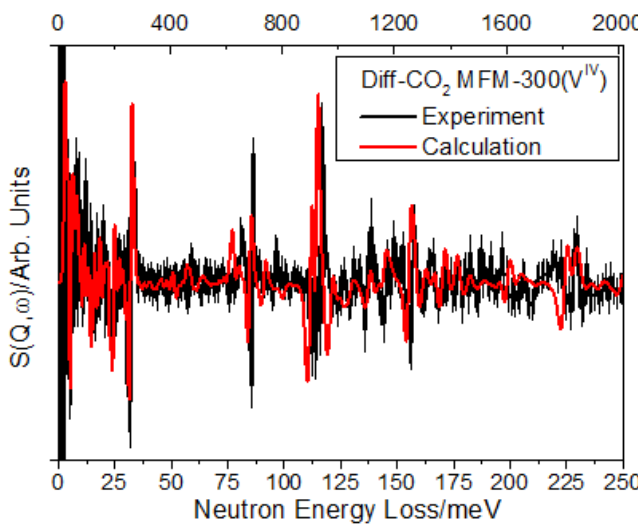
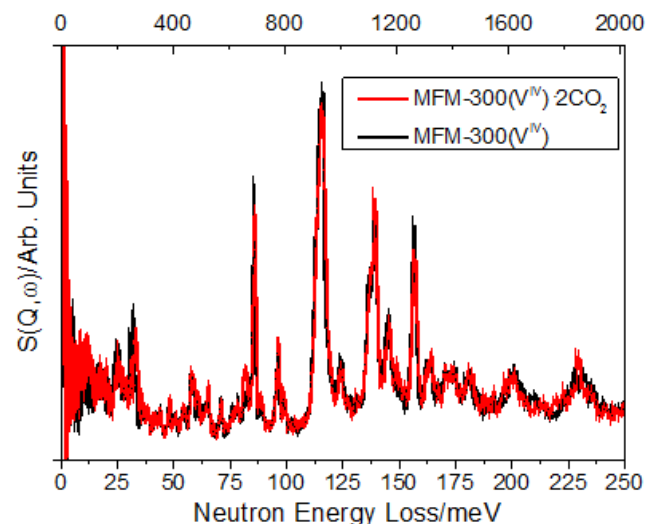
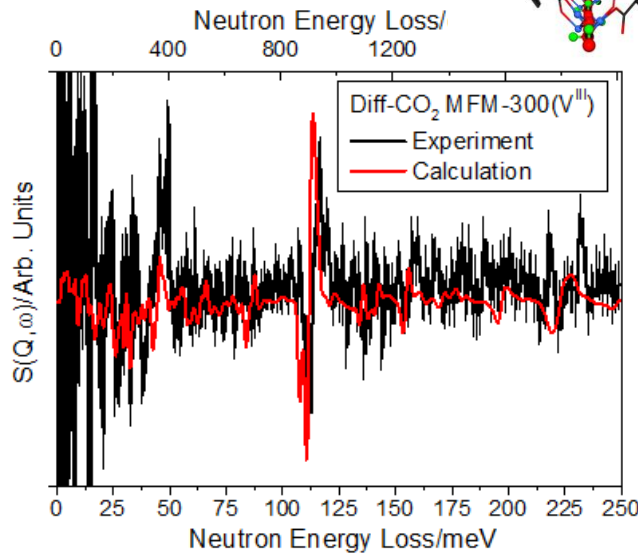
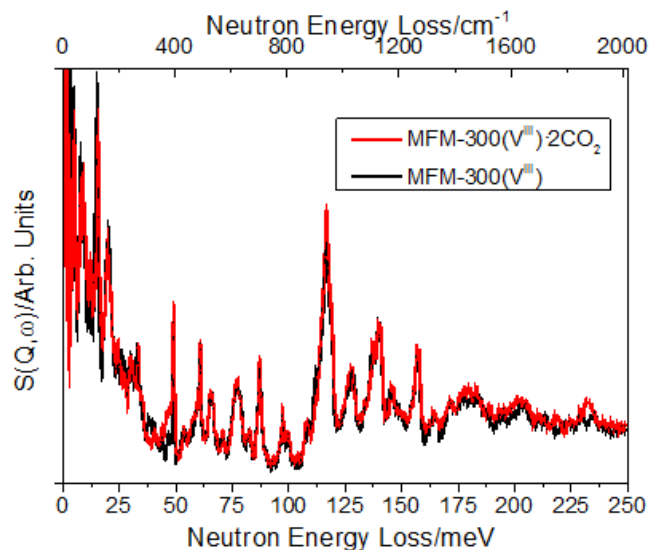
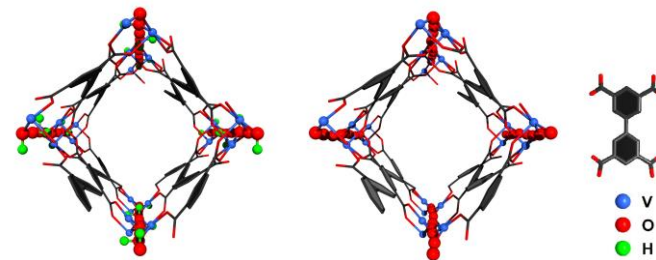


Measured (upper panel) and simulated (lower panel) INS spectra of blank ZIF-8 and ZIF-8+N<sub>2</sub>.

The strong peaks are mainly due to vibrational modes involving large displacement of hydrogen (in the methyl groups and the imidazolate rings)

Casco, M. E. et al. Gate-opening effect in ZIF-8: the first experimental proof using inelastic neutron scattering. *Chemical Communications*, v. 52, n. 18, p. 3639-3642, 2016

# INS signature of CO<sub>2</sub> capture

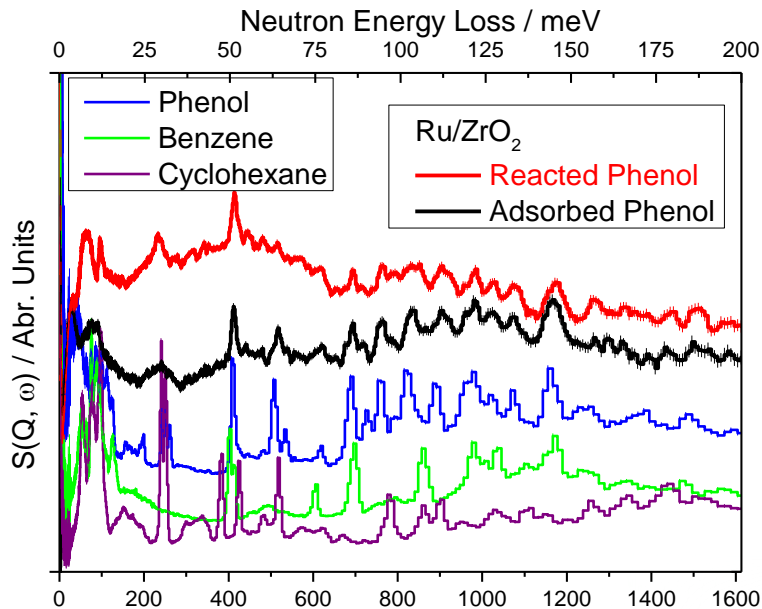
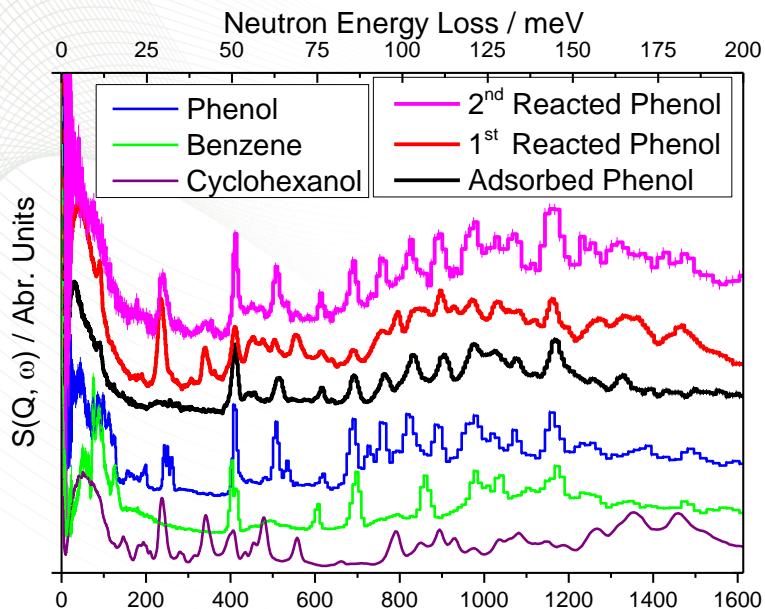


This study combining NPD, INS and modelling has unambiguously determined the CO<sub>2</sub> binding sites and structural dynamics for MFM-300(VIII) and MFM-300(VI). It is confirmed that the proton on the hydroxy group can not only attract and localise adsorbed CO<sub>2</sub> molecules via direct formation of hydrogen bonds, but also affects the macroscopic packing and arrangement of CO<sub>2</sub> molecules in the extended channel.

Z. Lu et al. "Modulating Supramolecular Binding of Carbon Dioxide in a Redox-

Active Porous Host" under review at Nature Communications

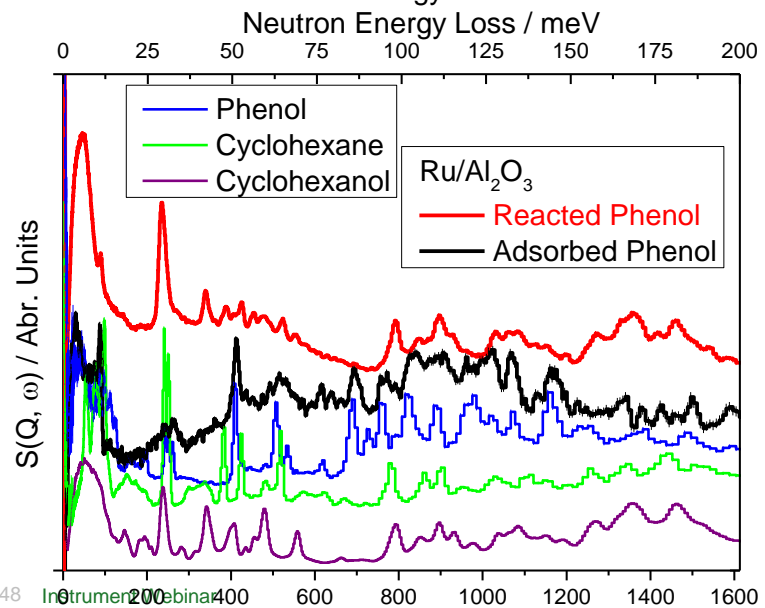
# Catalytic hydrodeoxygenation of phenol



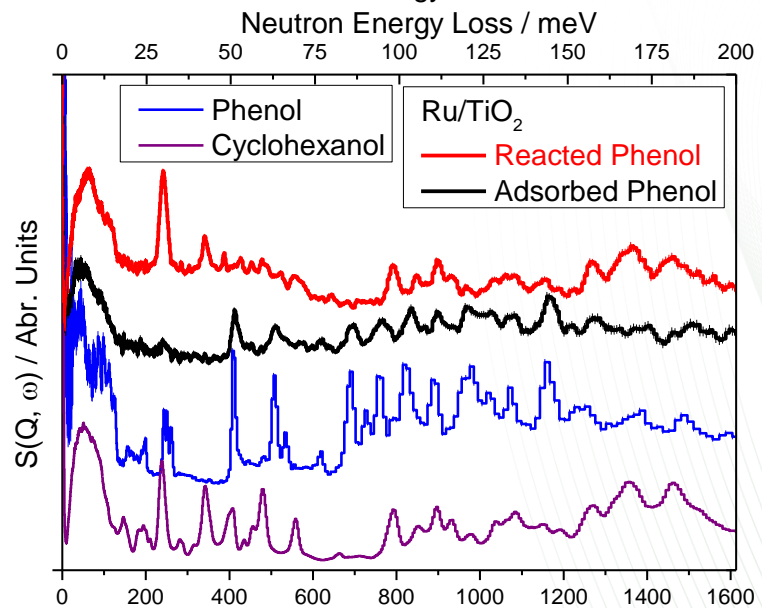
**Main products:**

**Ru/Nb<sub>2</sub>O<sub>5</sub>:**  
Benzene  
Cyclohexanol

**Ru/ZrO<sub>2</sub>:**  
Benzene  
Cyclohexane



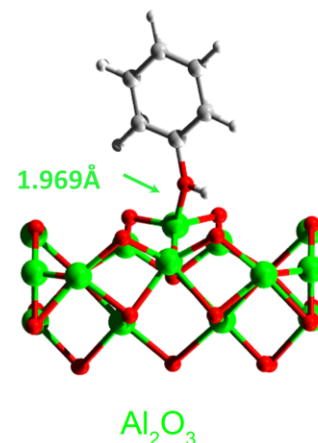
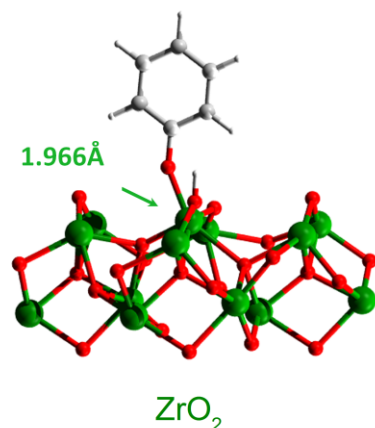
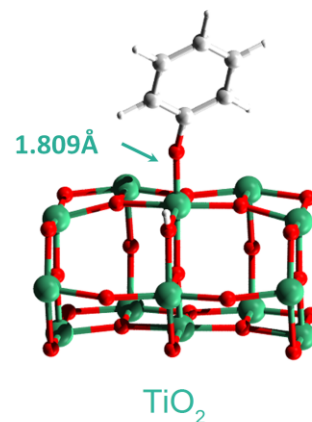
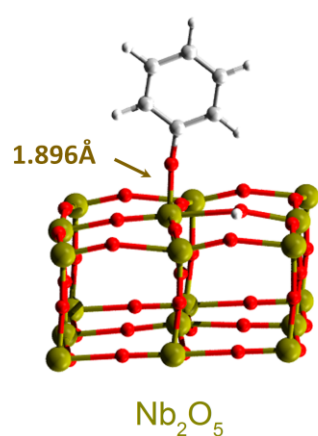
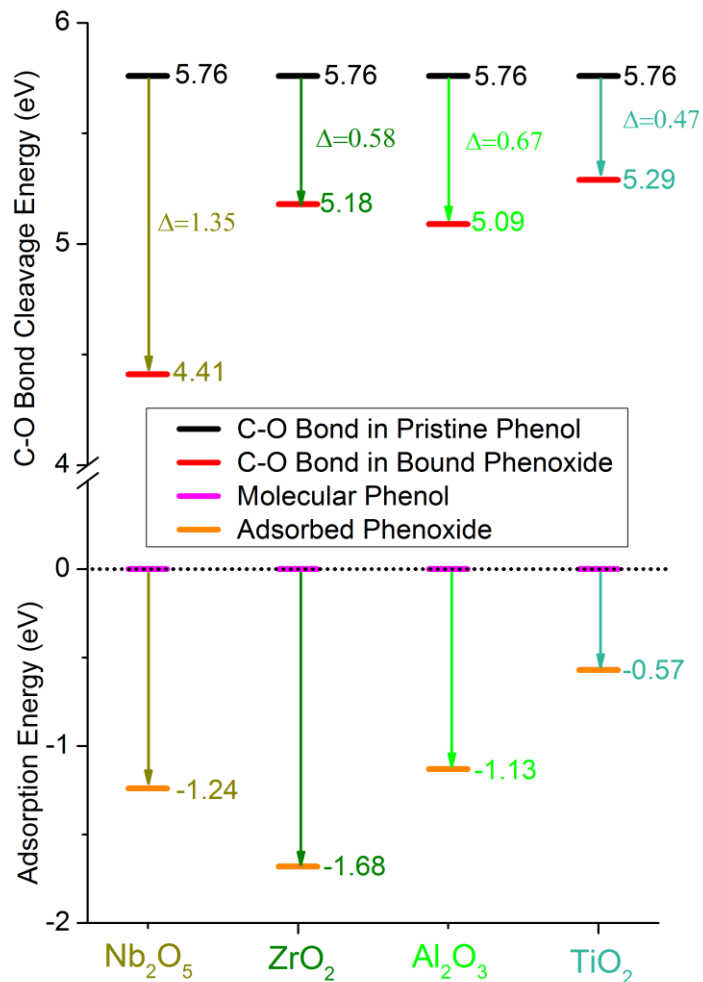
**Ru/Al<sub>2</sub>O<sub>3</sub>:**  
Cyclohexane  
Cyclohexanol



**Ru/TiO<sub>2</sub>:**  
Cyclohexanol



# Catalytic hydrodeoxygenation of phenol



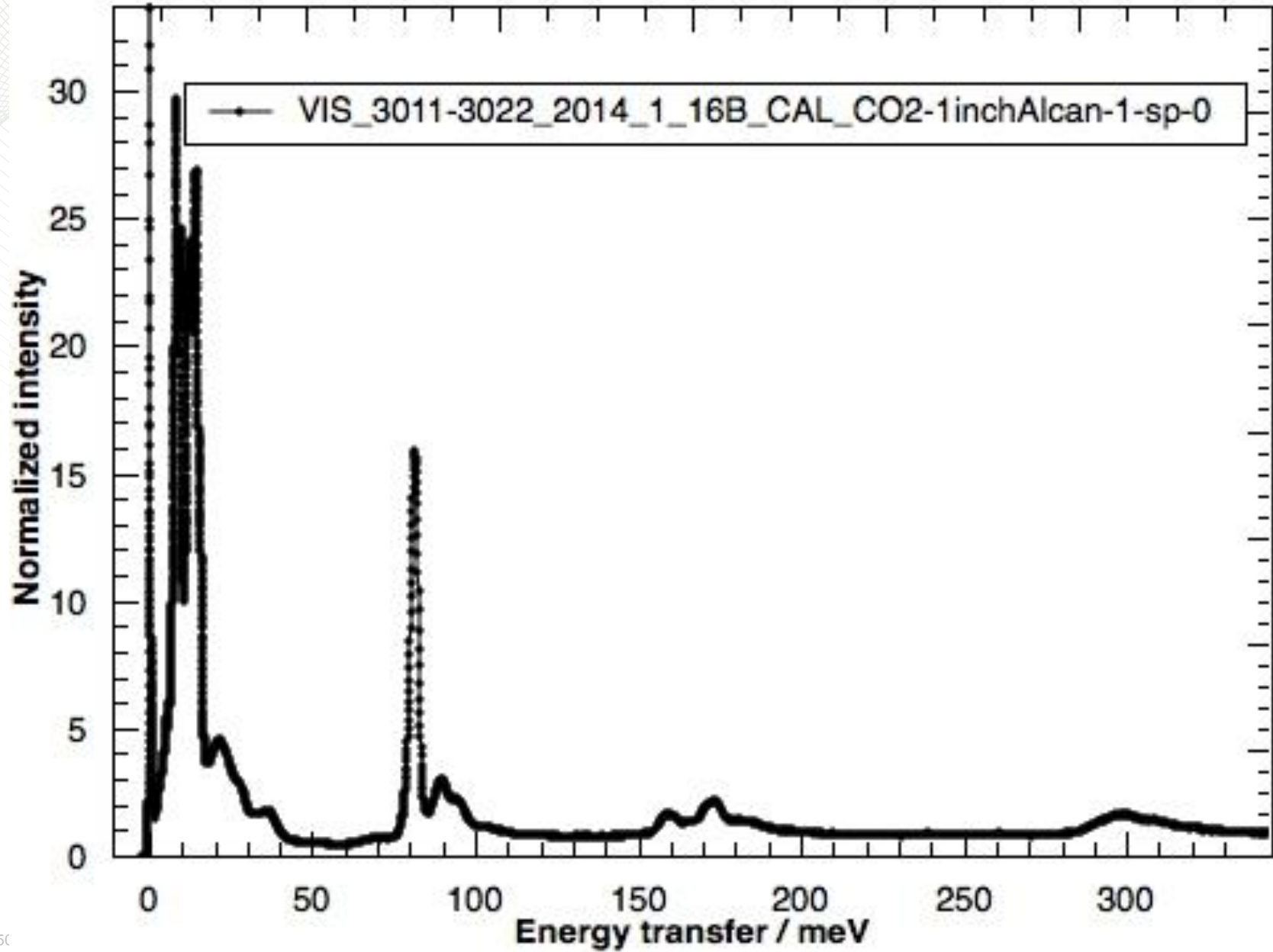
Shao, Y. et al. Selective production of arenes via direct lignin upgrading over a niobium-based catalyst. Nat. Commun. 8, 16104 (2017).

**OH removal:** strength of phenol chemisorption on the surface

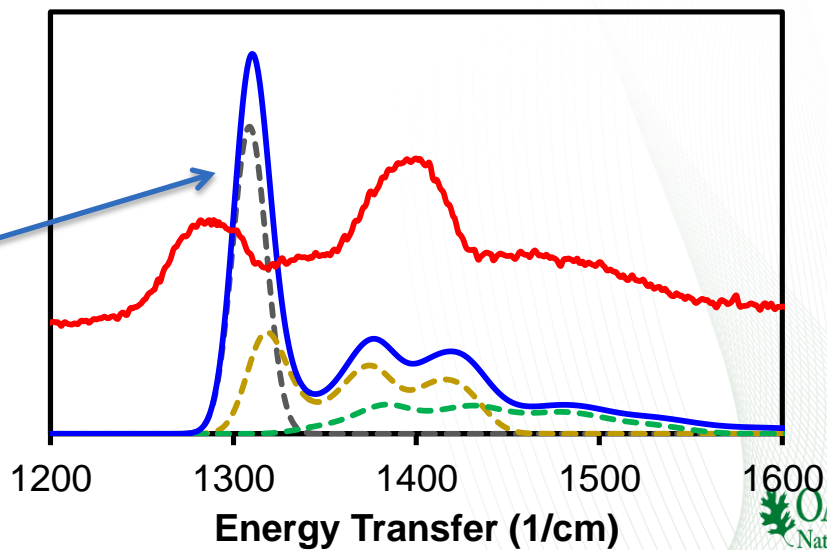
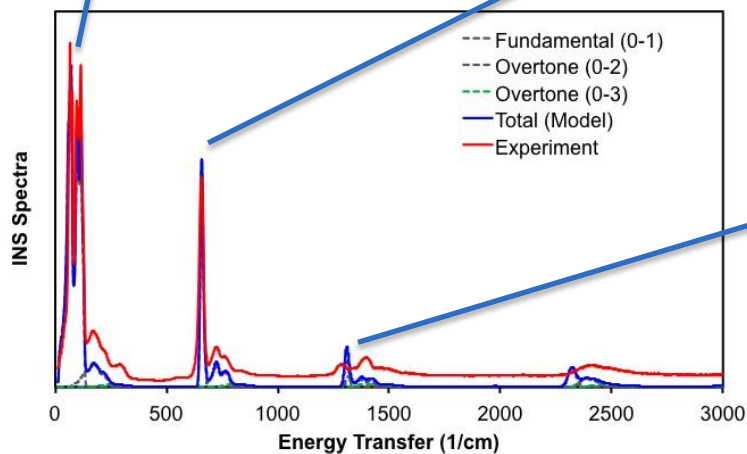
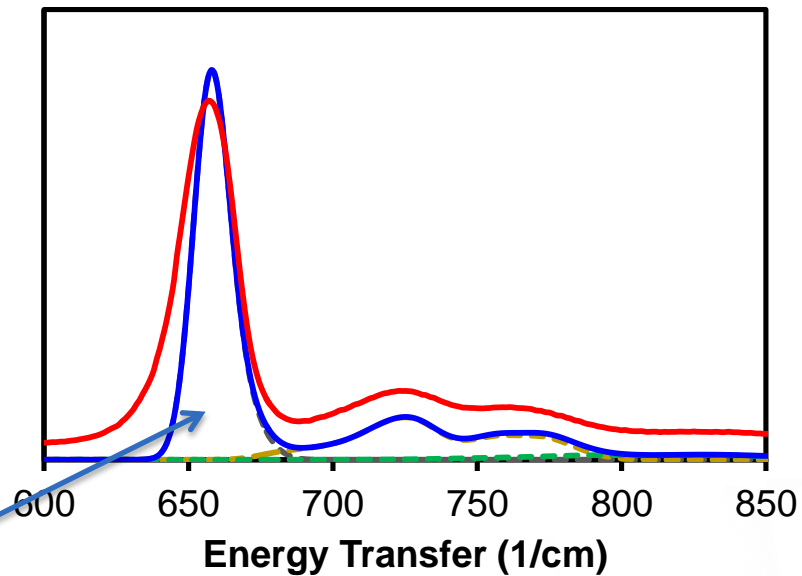
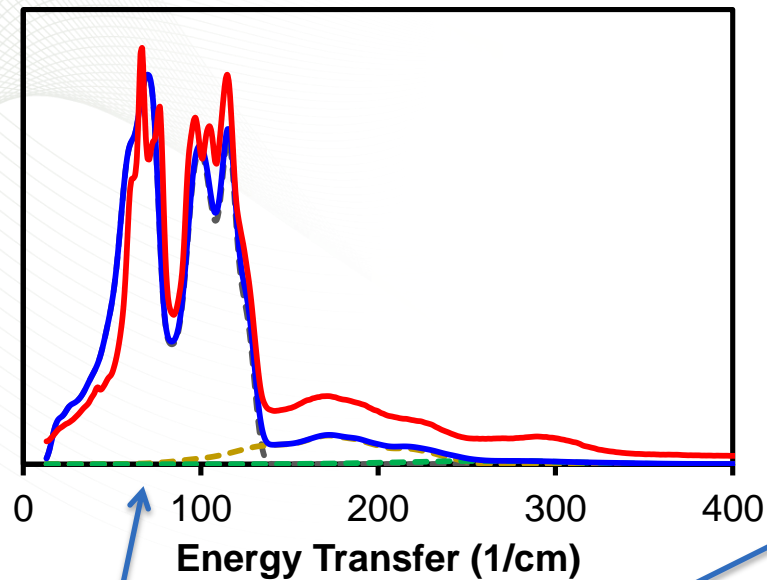
**Selectivity between the competitive processes of C<sub>aromatic</sub>-O bond cleavage and**

**C6-ring hydrogenation:** reduction in C-O bond cleavage energy

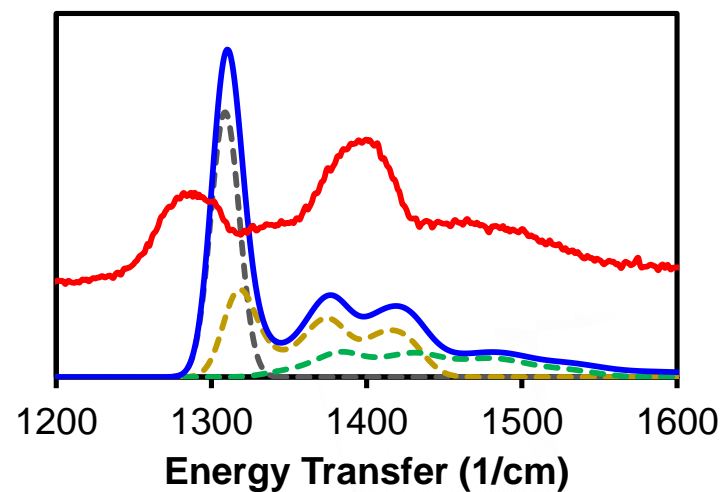
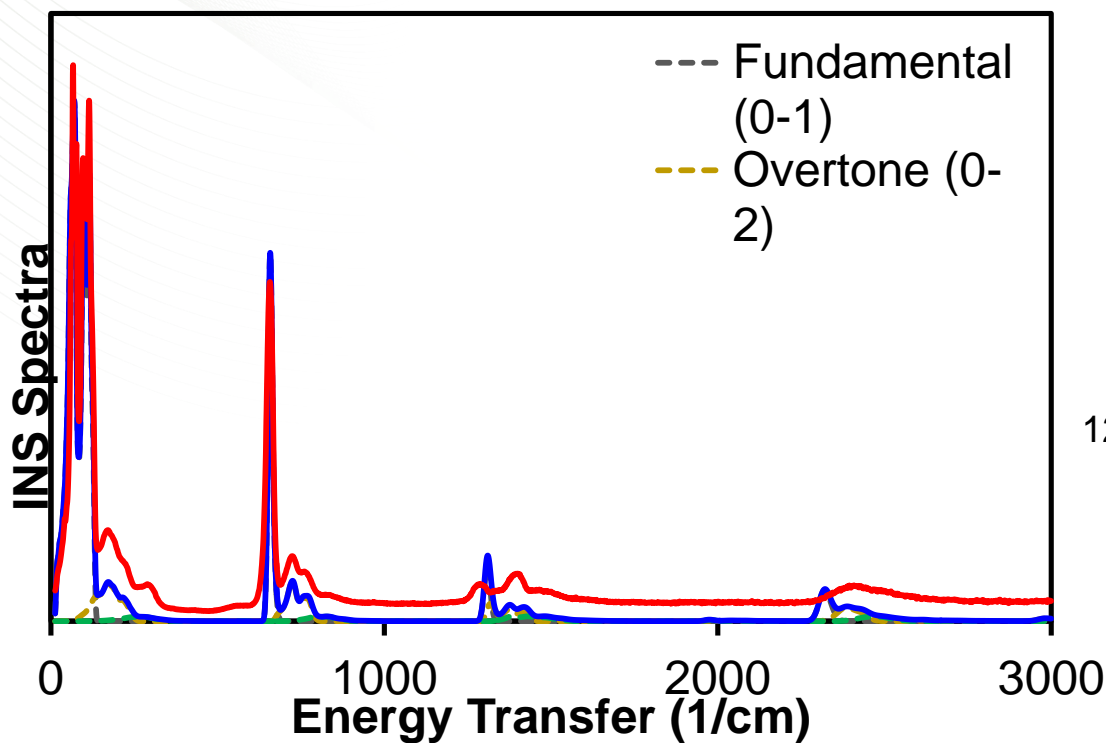
CO<sub>2</sub>



# CO<sub>2</sub> in the solid phase

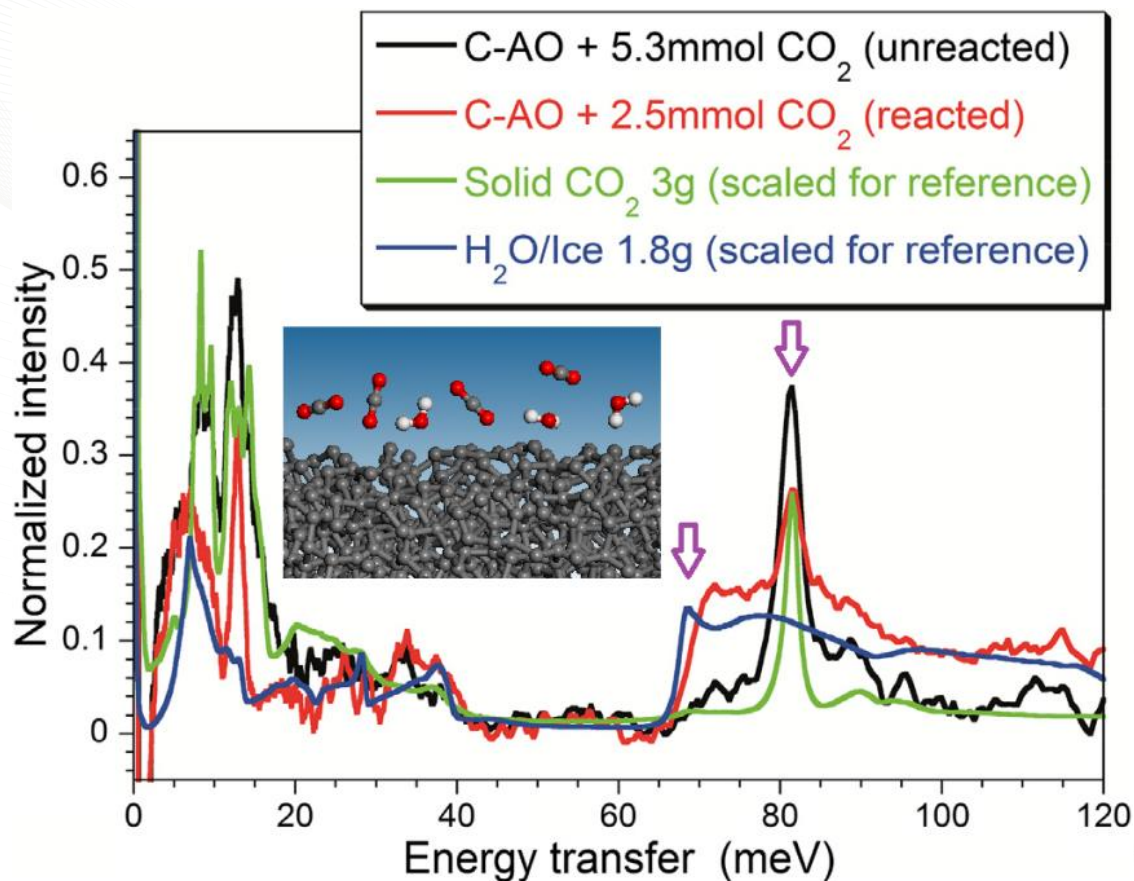


## Examples from VISION



- Overlap of the an overtone (of the translational modes) with a fundamental (the C-O symmetric stretching): the *Fermi resonance*
- First observation of CO<sub>2</sub> Fermi resonance *using INS*

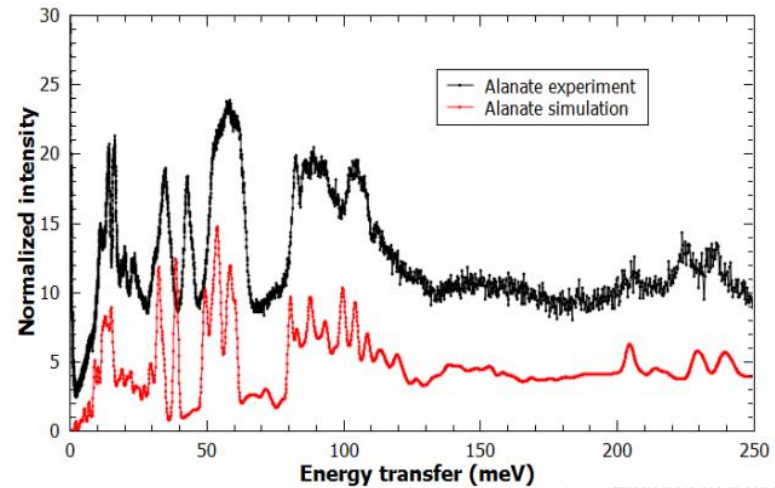
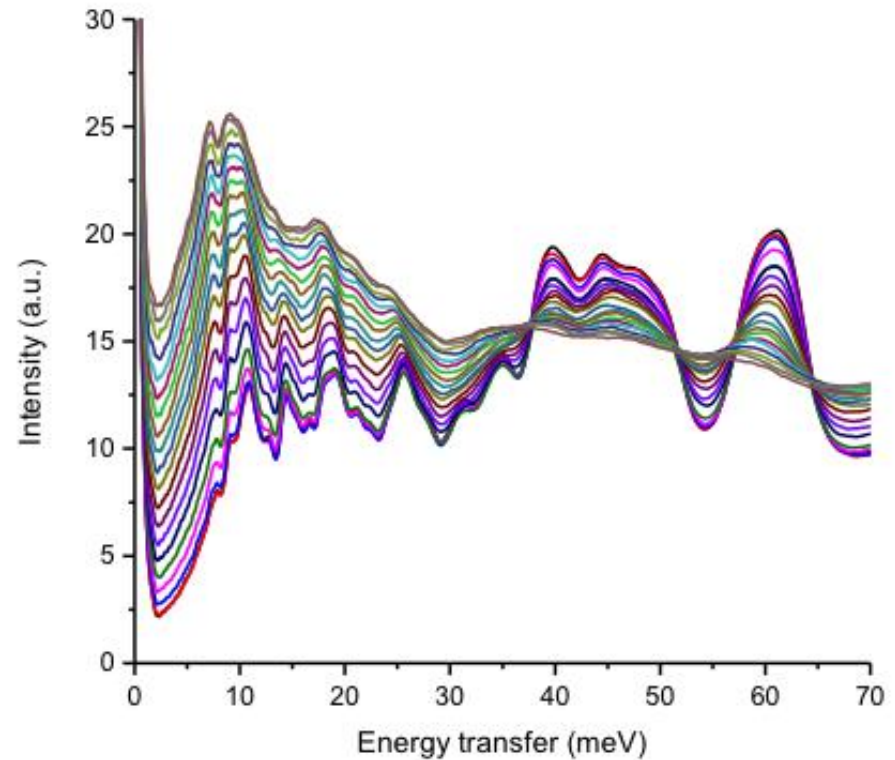
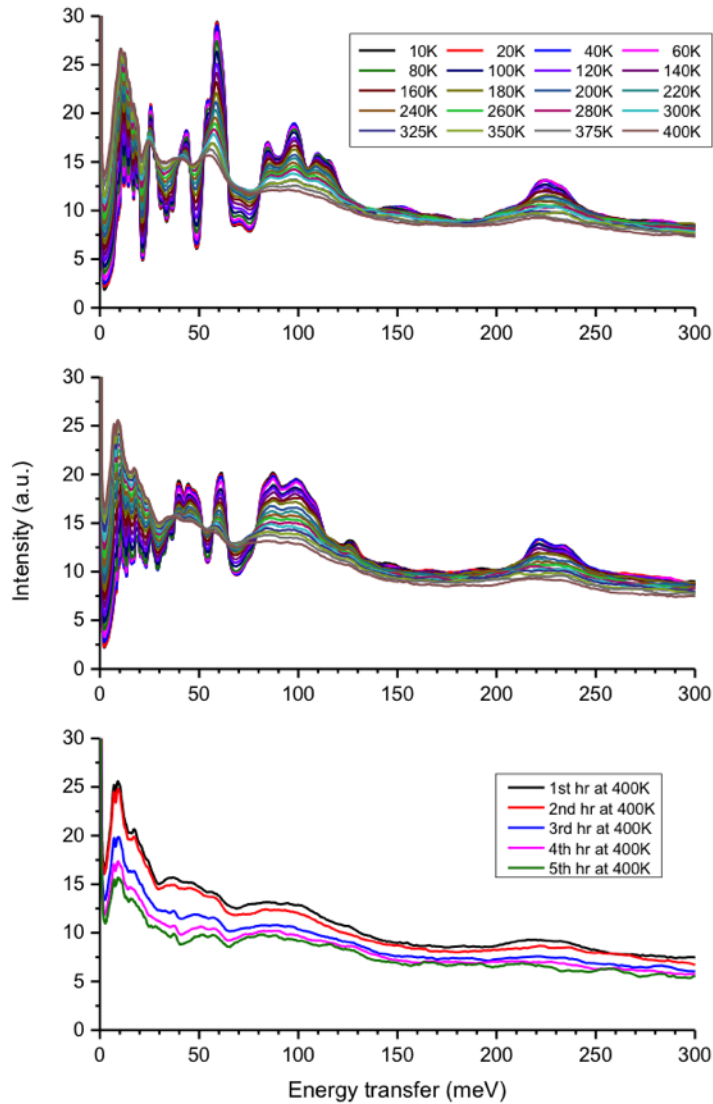
# Small amount of non-hydrogenous samples



The difference INS spectra before and after CO<sub>2</sub> dosing in C-AO (a nanoporous carbon sample), in comparison with the reference spectra for bulk solid CO<sub>2</sub> and H<sub>2</sub>O. Signal from the background and the blank C-AO has been subtracted.

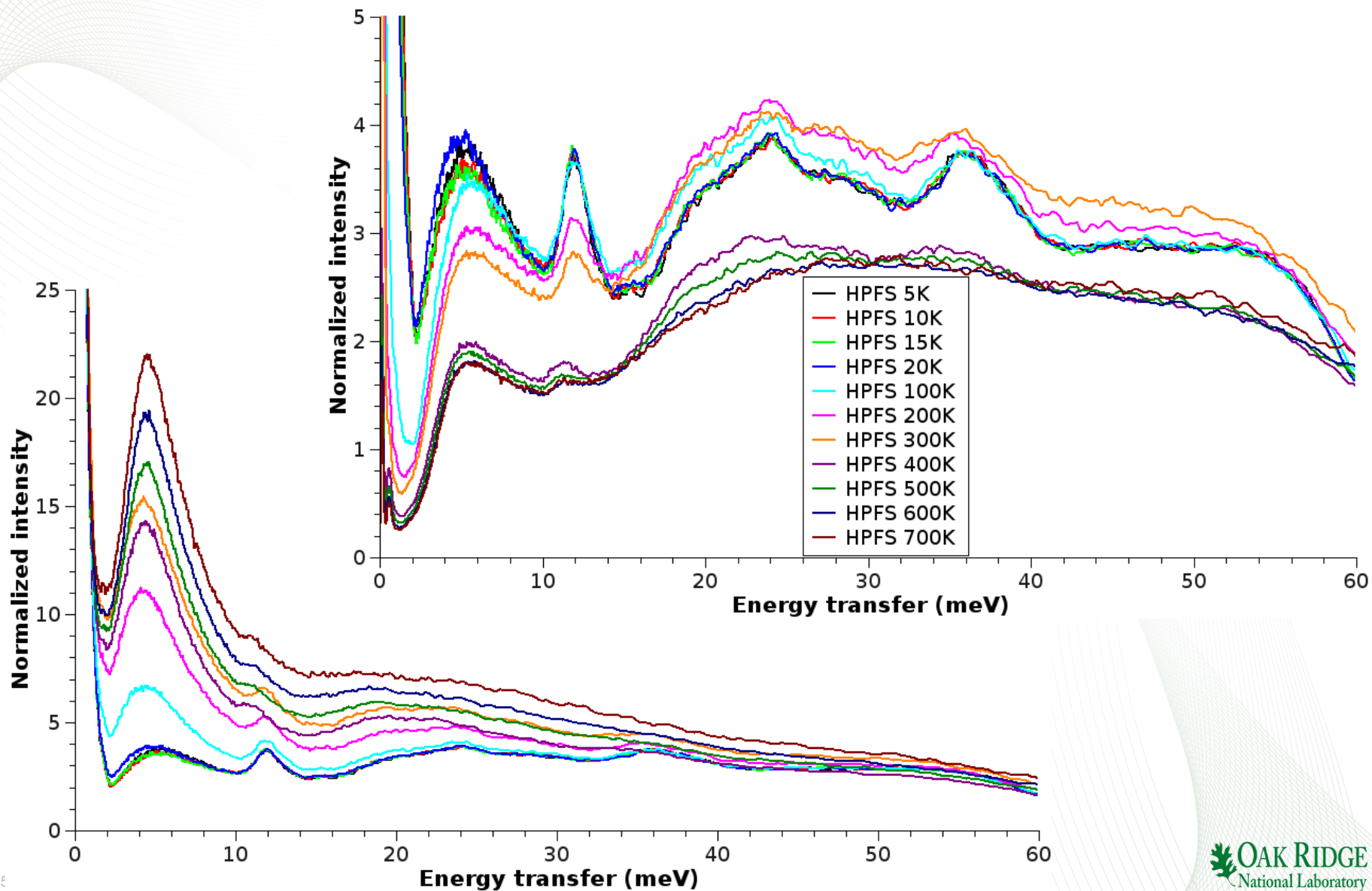
Very small amount of non-hydrogenous gas. In situ observation of surface reactions. Surface science, catalysis, gas capture and storage.

# Hydrogen in metal alanates

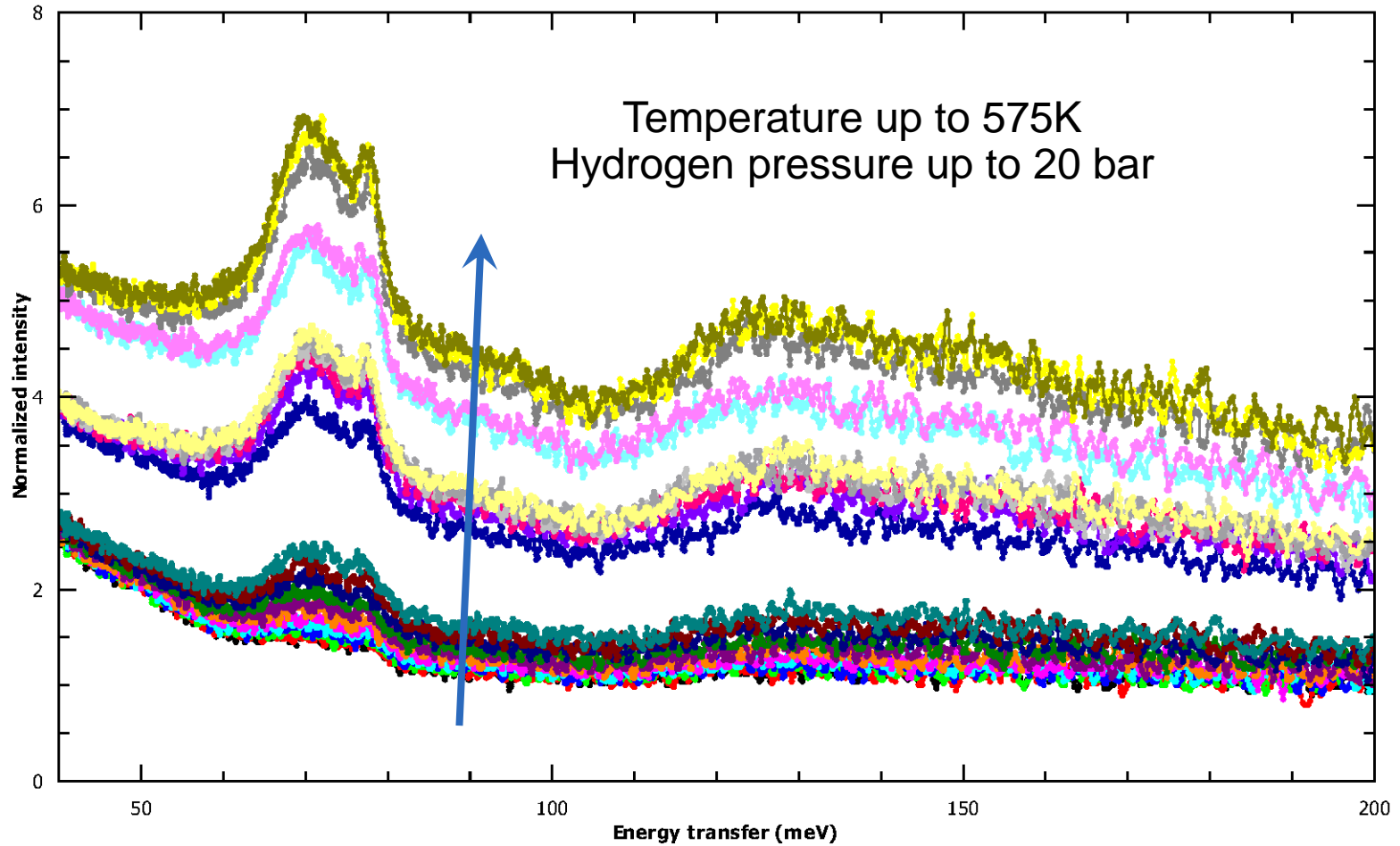


Thanks to Shin-ichi Orimo, Toyoto Sato,  
Shigeyuki Takagi and Keisuke Tomiyasu,  
Tohoku University, Japan.

# High temperature measurement up to 700K



# In situ observation of metal hydride formation





# Diffraction at VISION

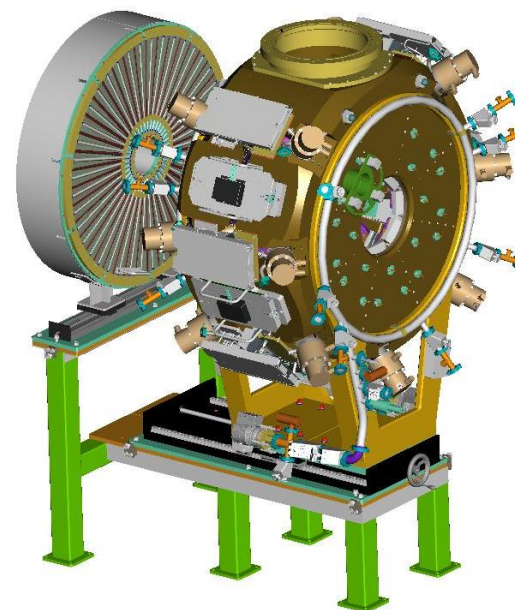
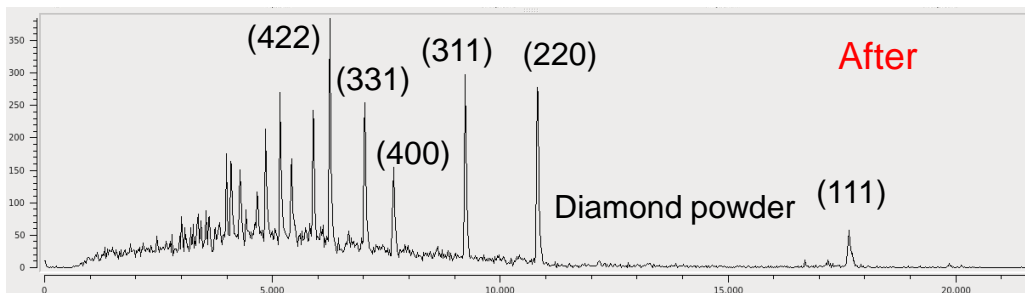
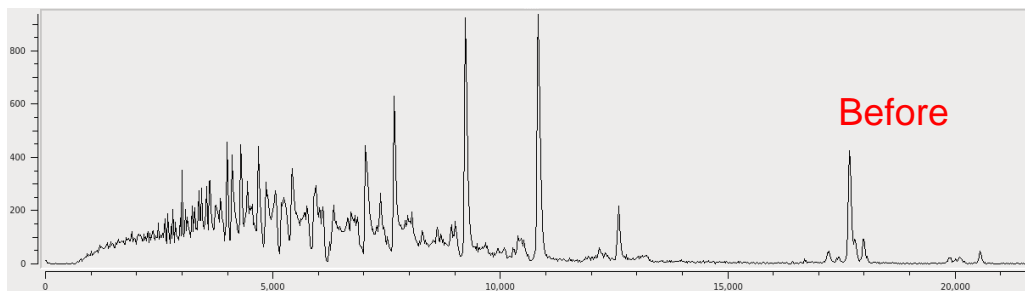
## Simultaneous diffraction and inelastic neutron scattering



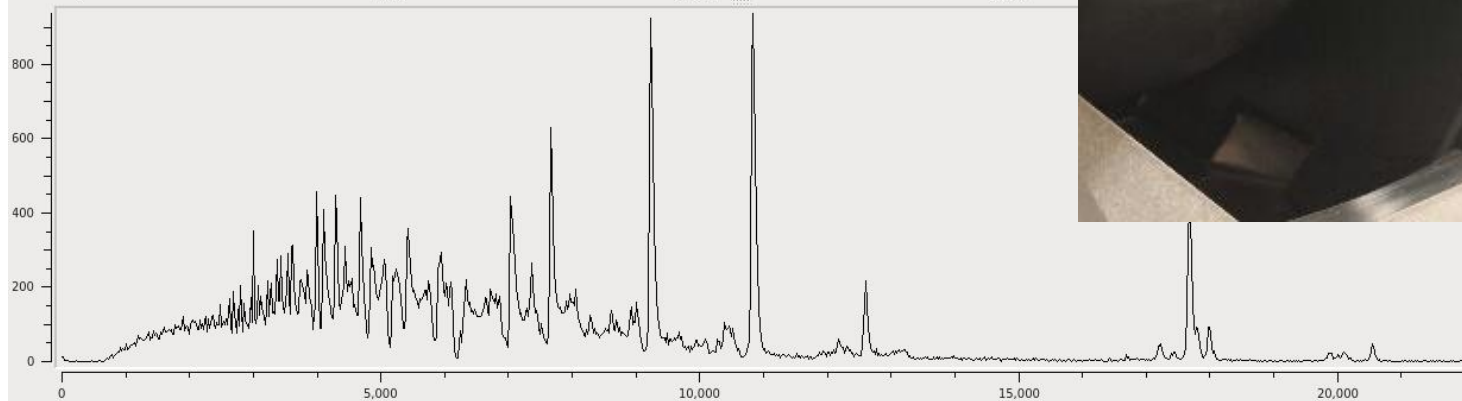
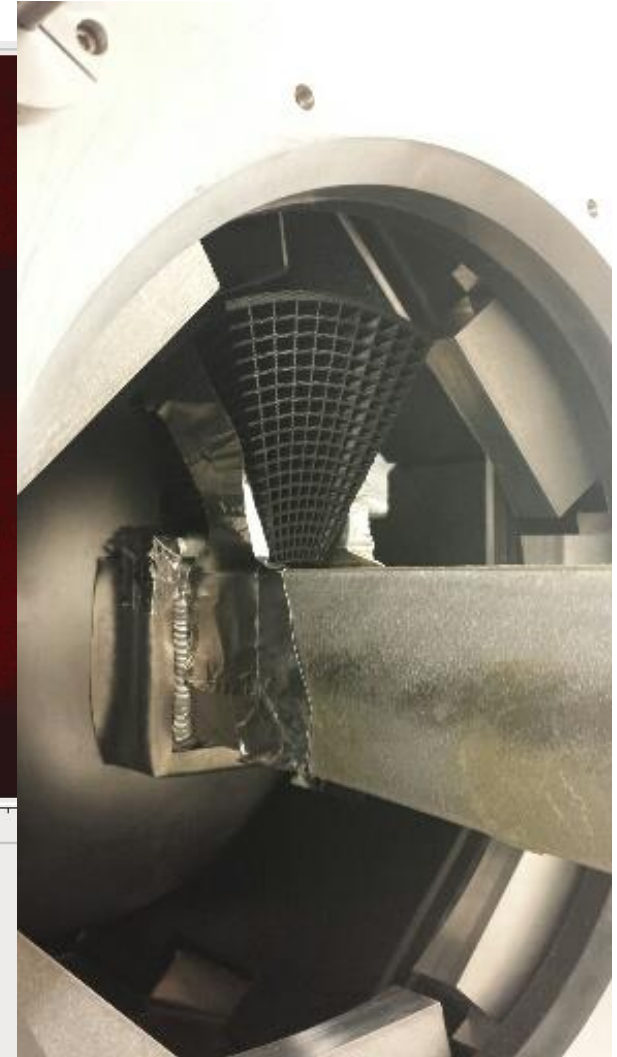
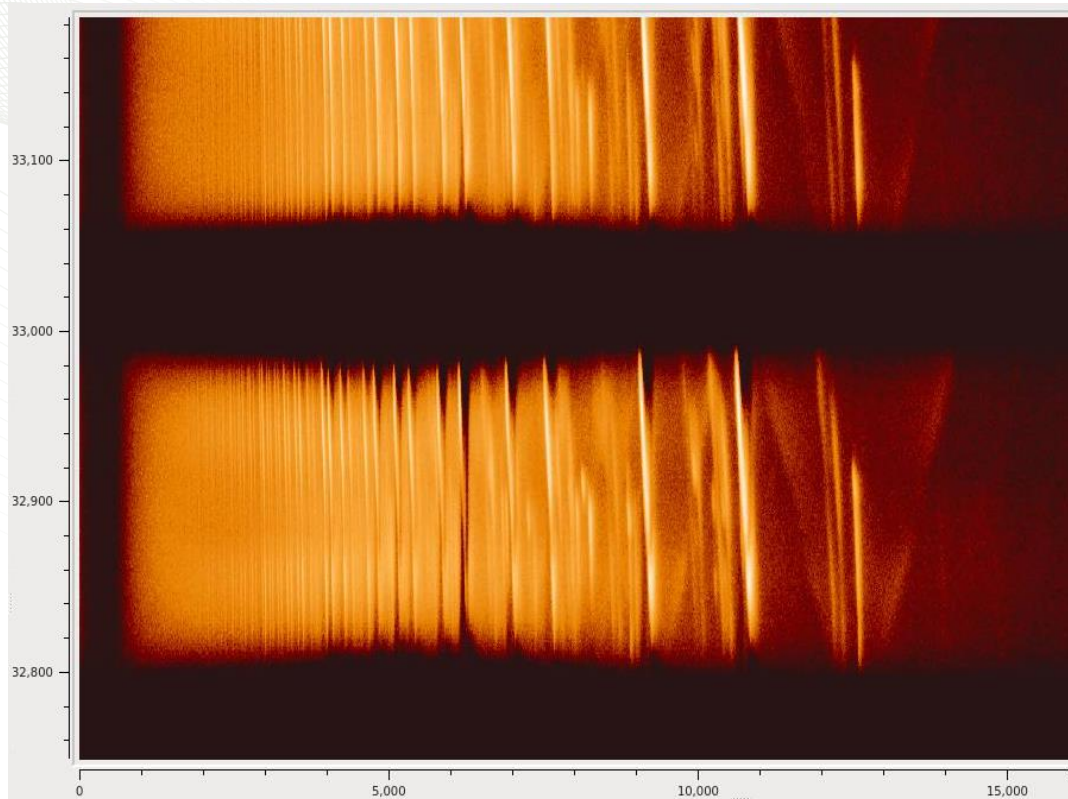
3D printed collimators have been tested for VISION to be used in the backscattering diffraction bank.

The reduction of the spurious peaks from the sample is very much noticeable.

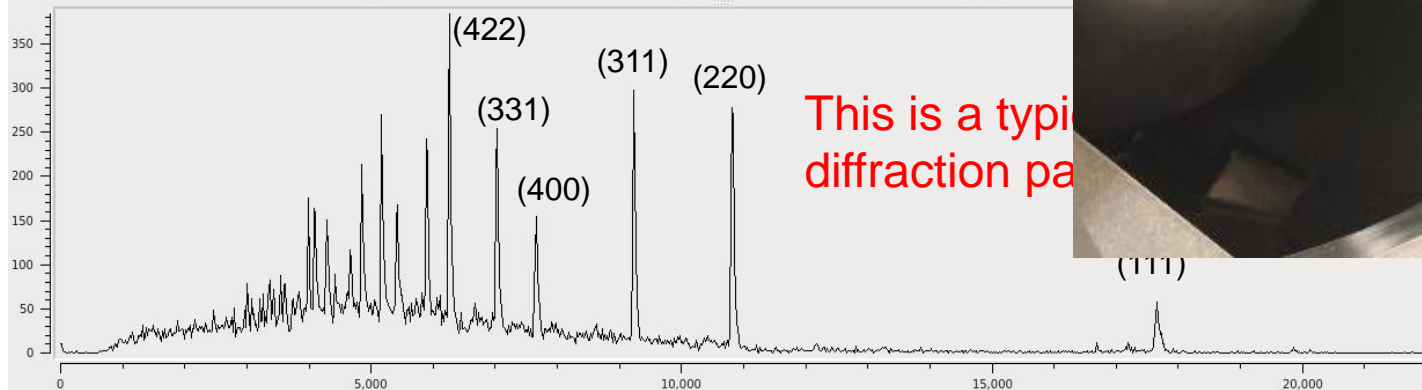
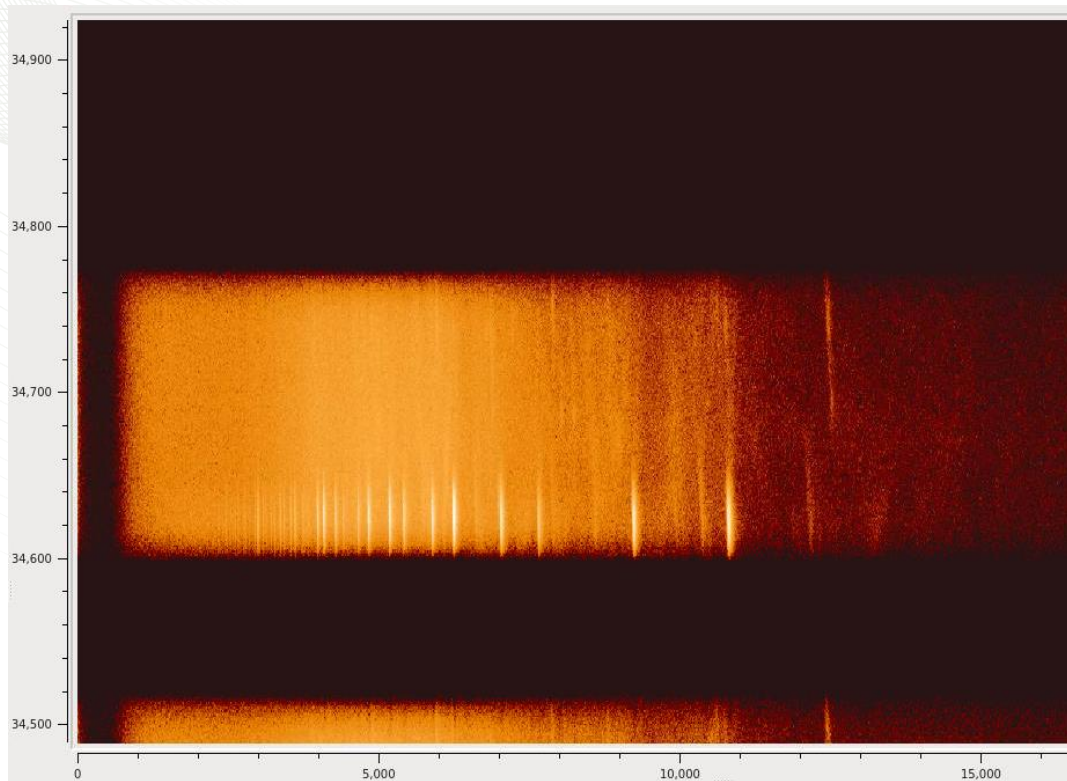
Data collected in histogram or event mode



# Bank 24, Tube 1 (NOT covered by collimator)



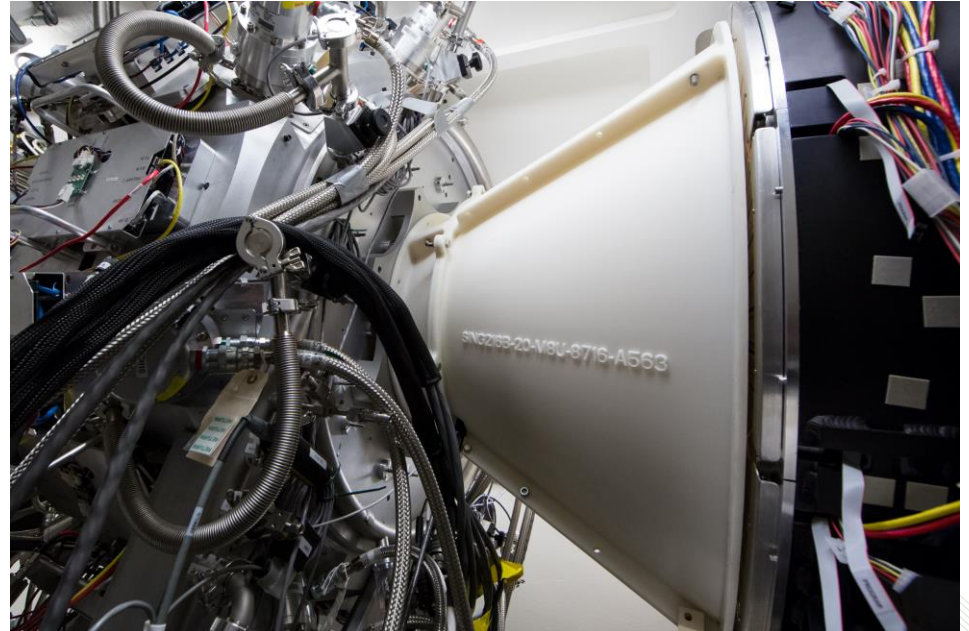
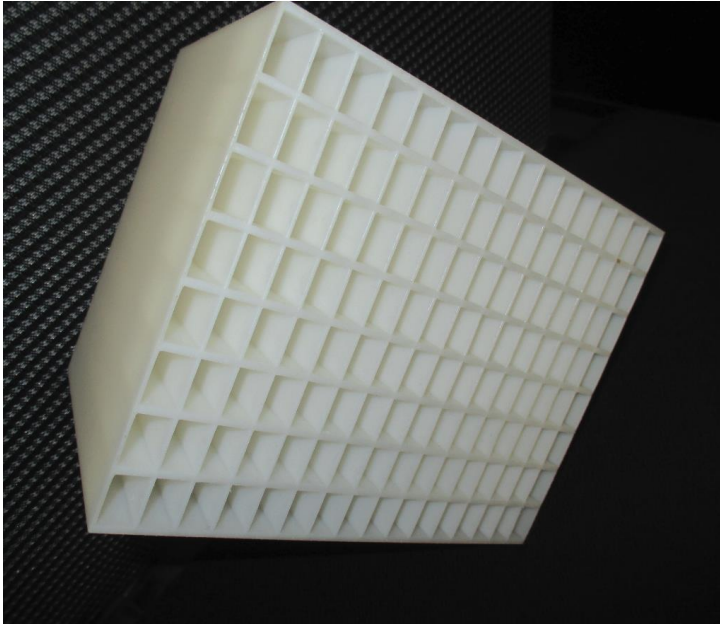
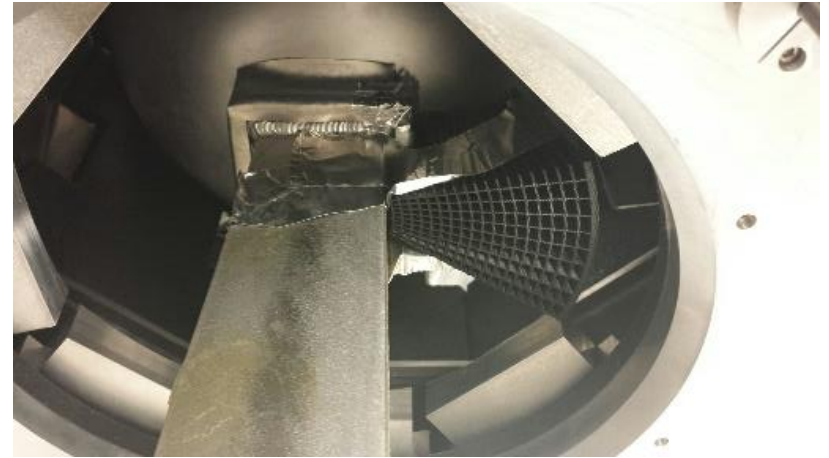
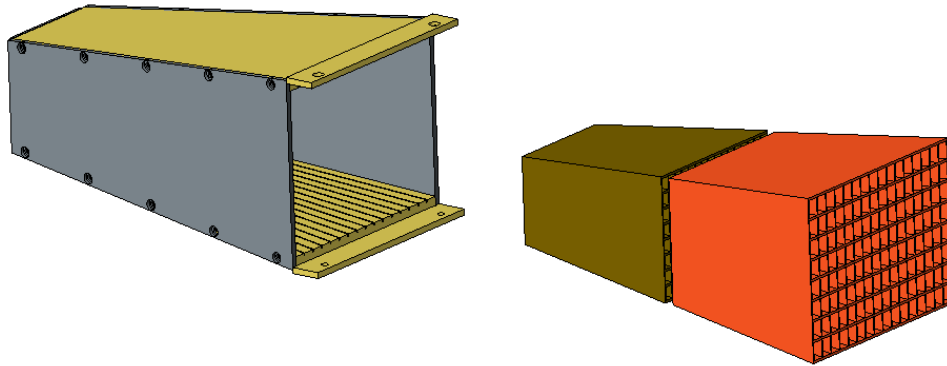
# Bank 24, Tube 8 (covered by collimator)



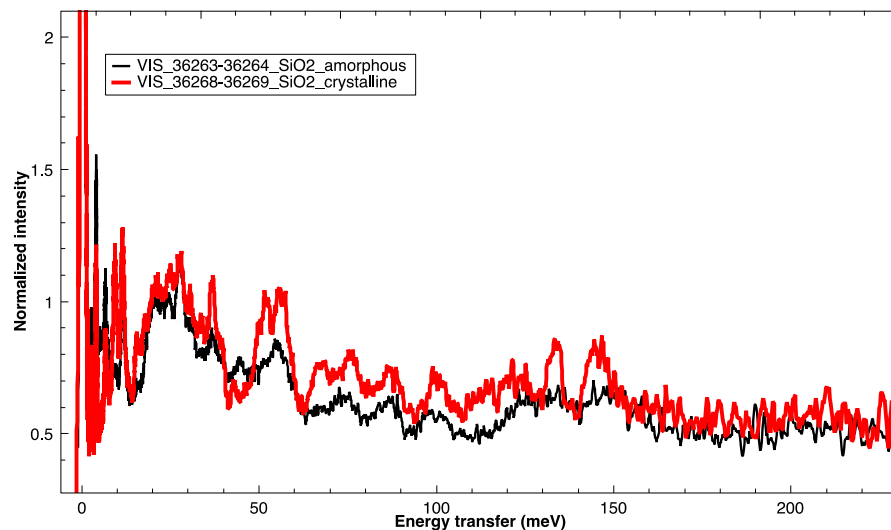
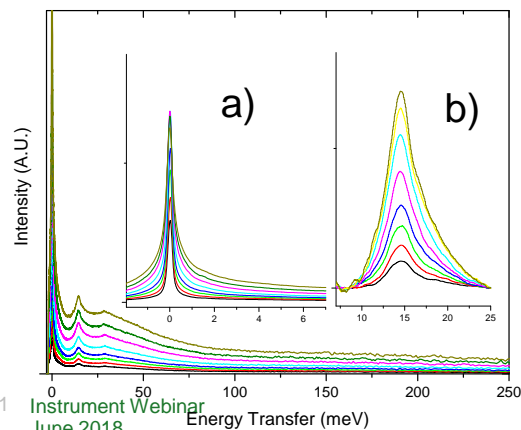
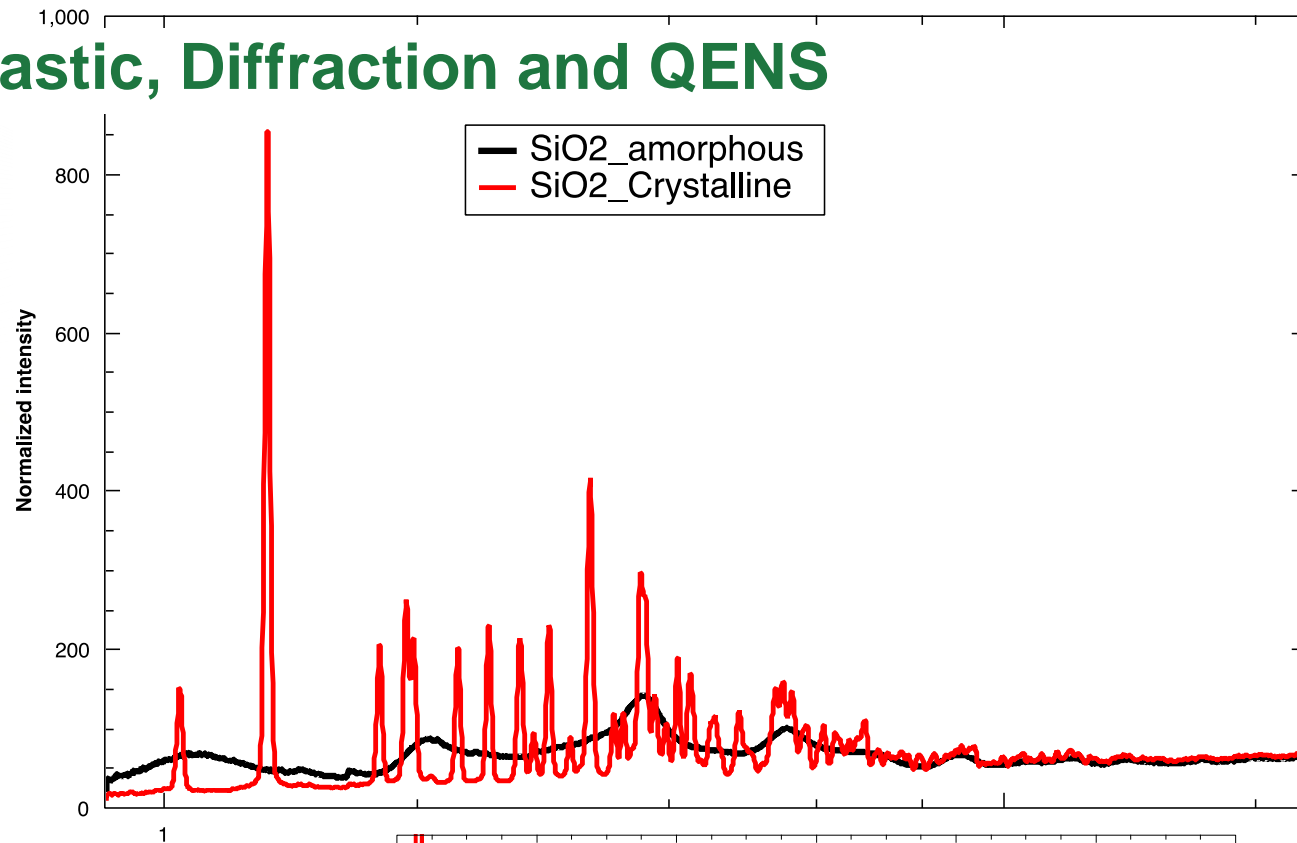
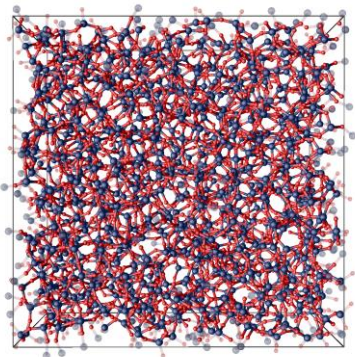
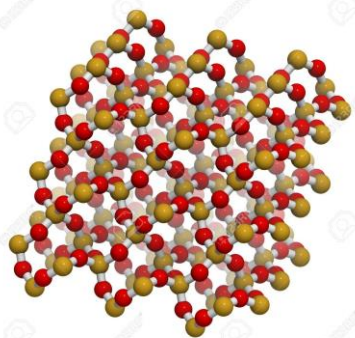
This is a typical diffraction pattern



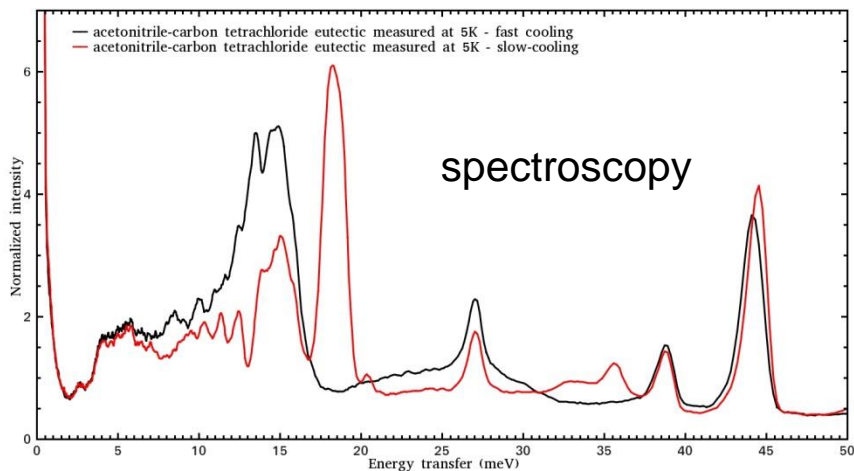
# VISION pioneered the use of 3D printed collimators



# VISION: Inelastic, Diffraction and QENS



# Structure and dynamics of liquids and solutions



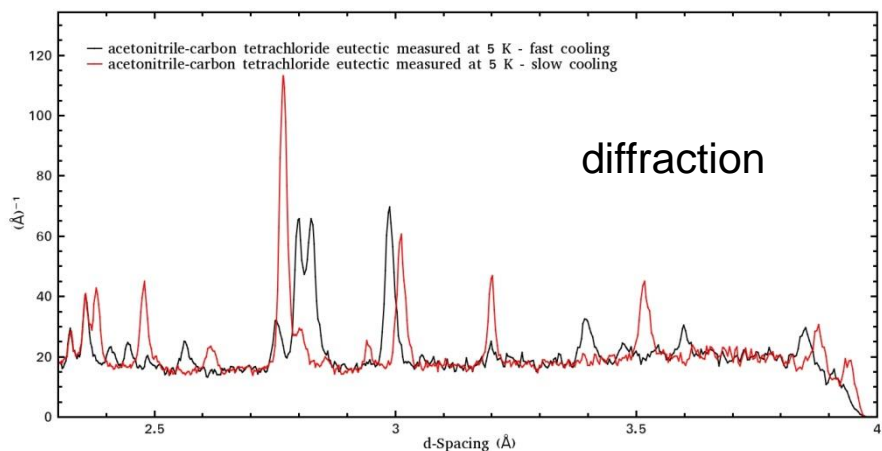
-  $\text{CD}_3\text{CN}-\text{CCl}_4$  (deep) eutectic system

-  $T_E = 210$  K;  $x(\text{CD}_3\text{CN})$  at eutectic composition is 0.75

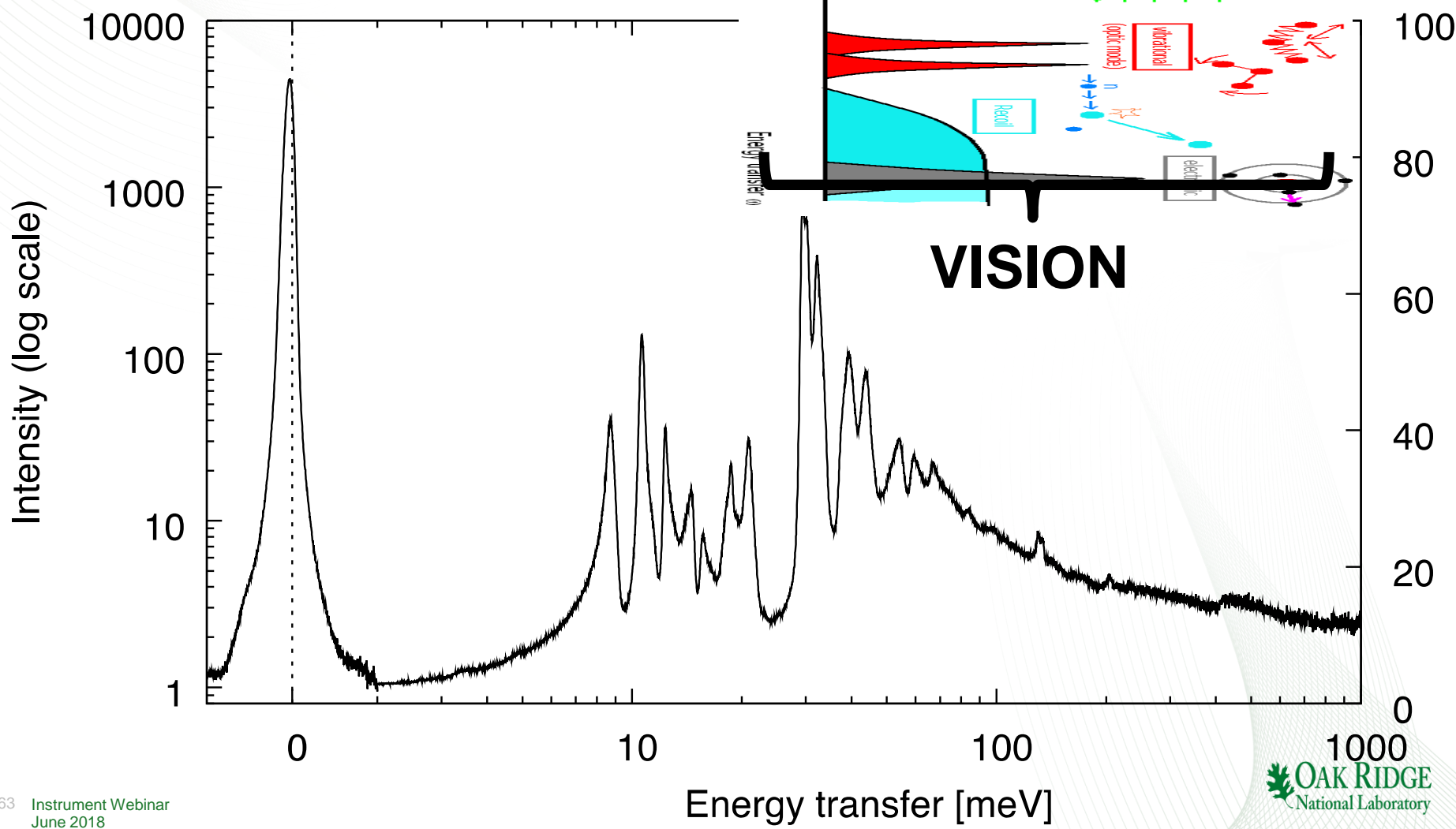
- no hydrogen bonding, but highly non-ideal system with  $\Delta H_{\text{excess}} = -800$  J/mol)

- eutectic structure differs when liquid is cooled quickly or slowly from room temperature

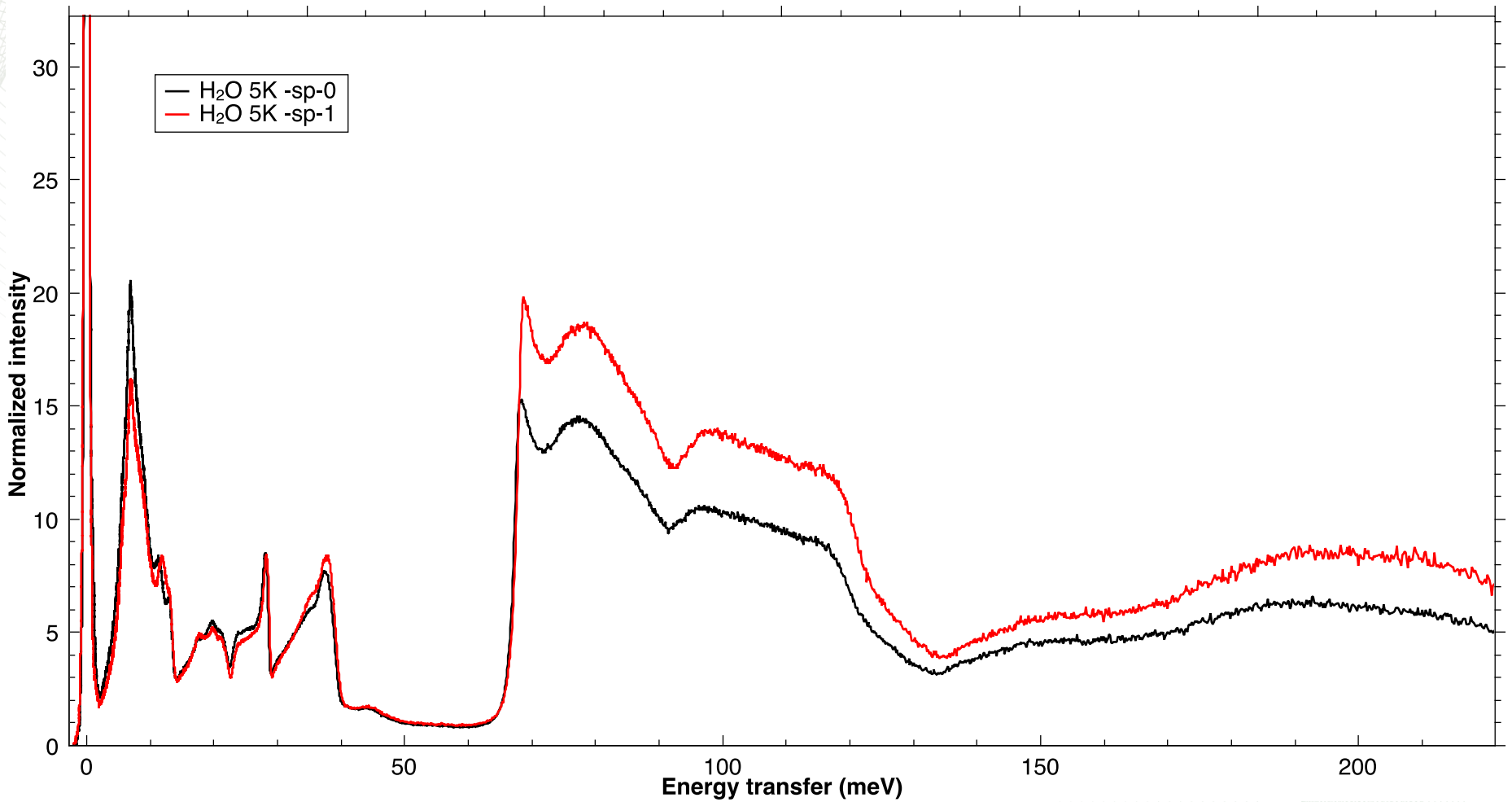
- simultaneous diffraction/spectroscopy is invaluable



# The energy spectra (according to VISION)

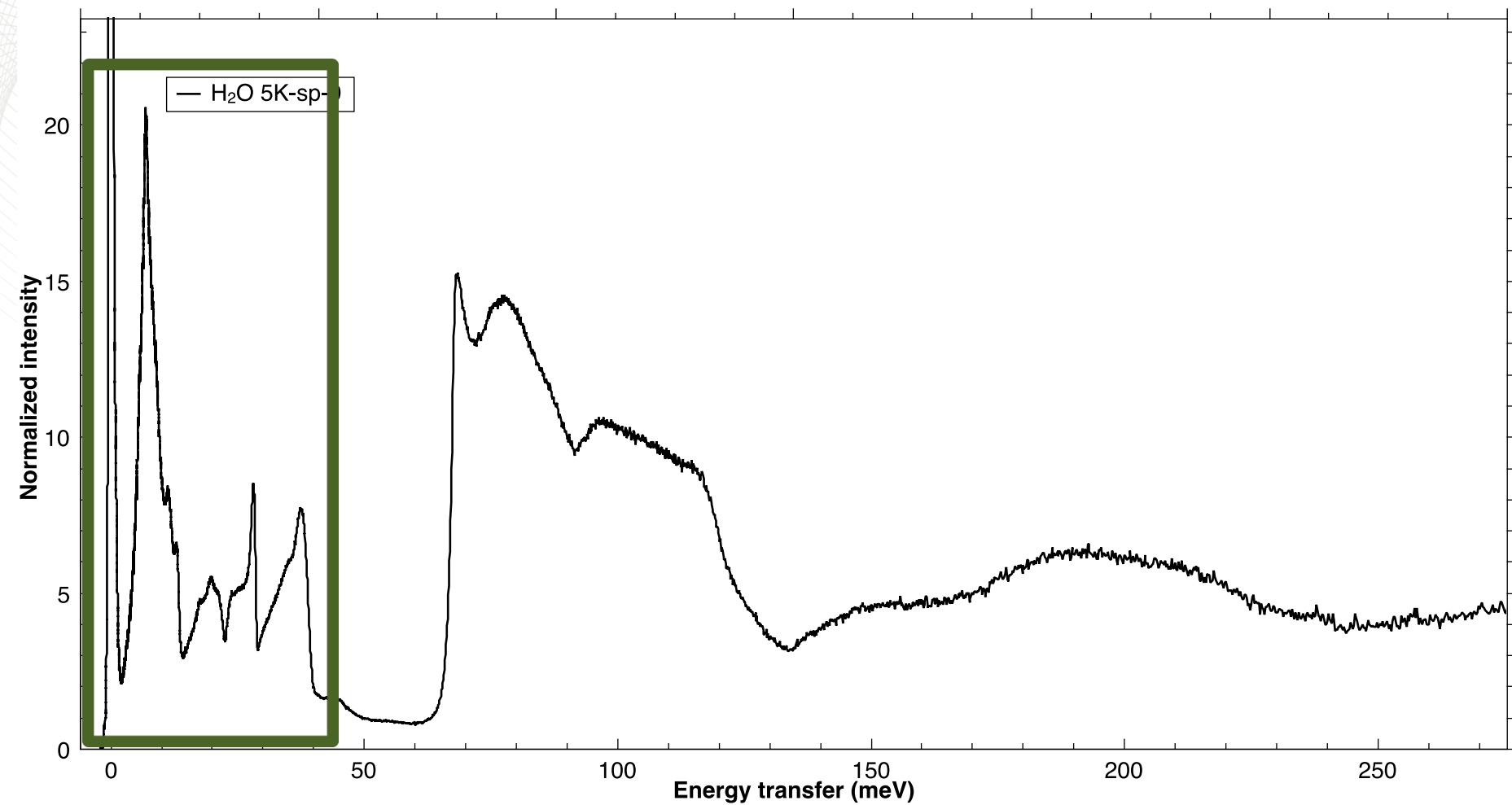


# Water (ice) in VISION

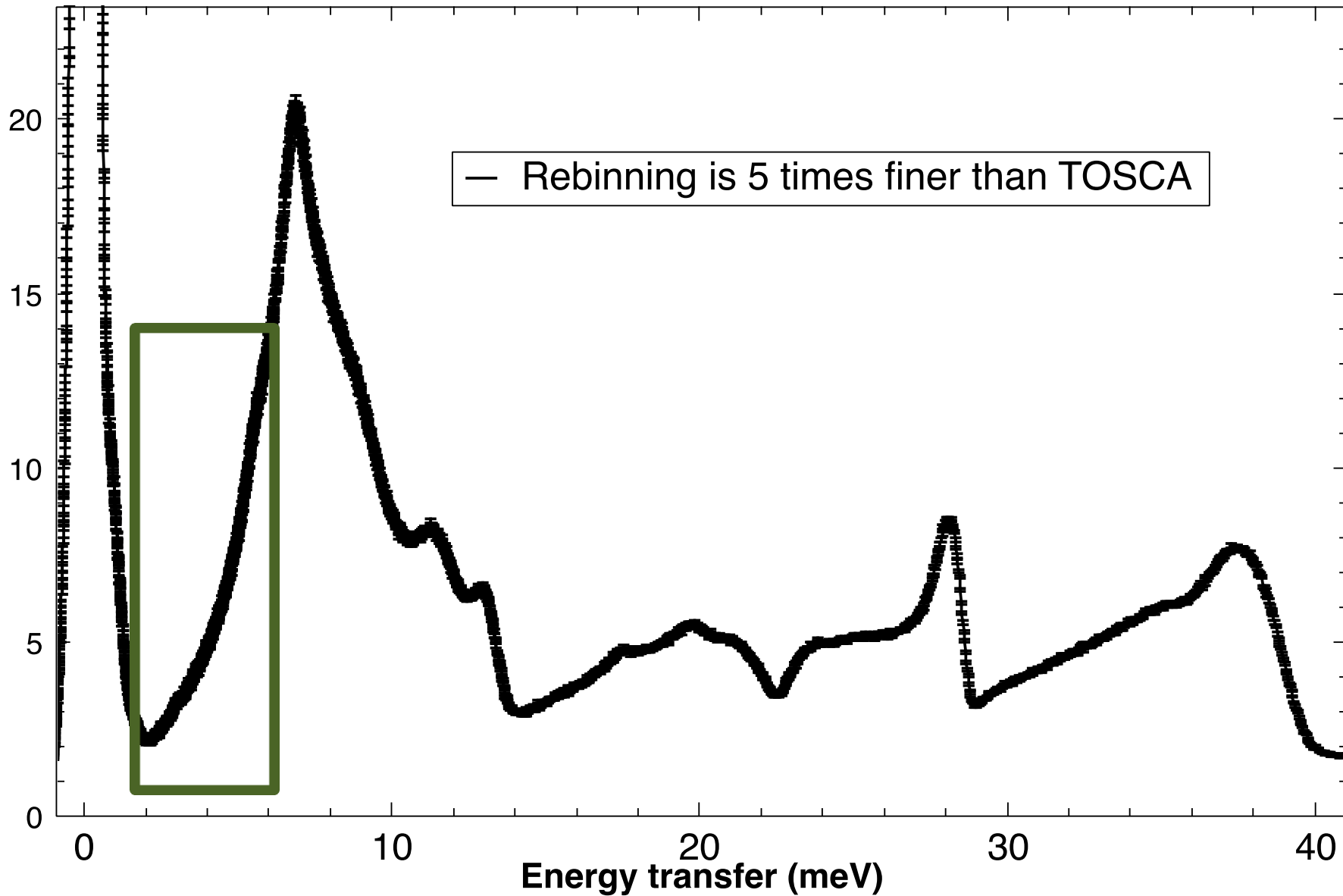




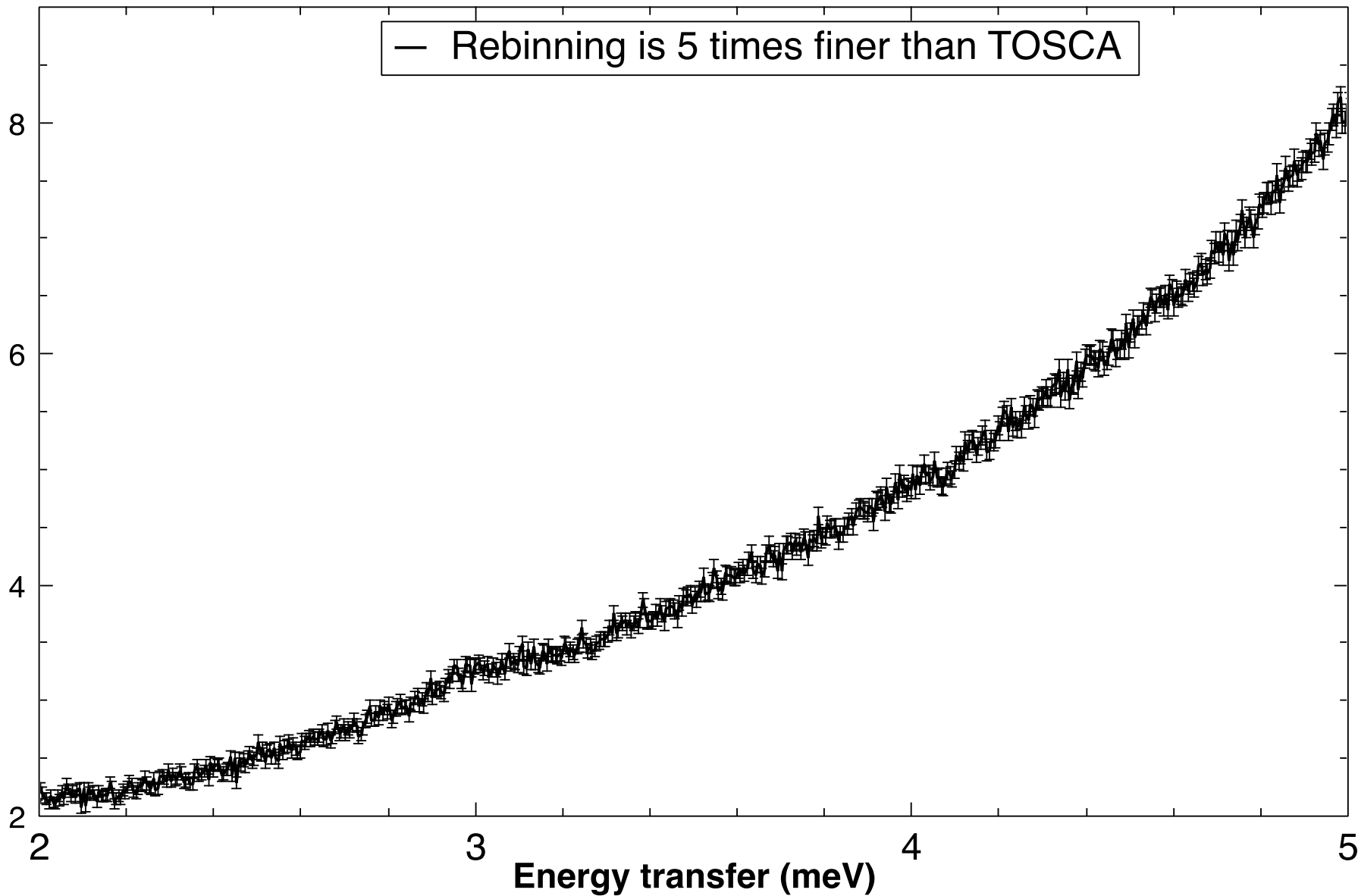
# Water (ice) in VISION



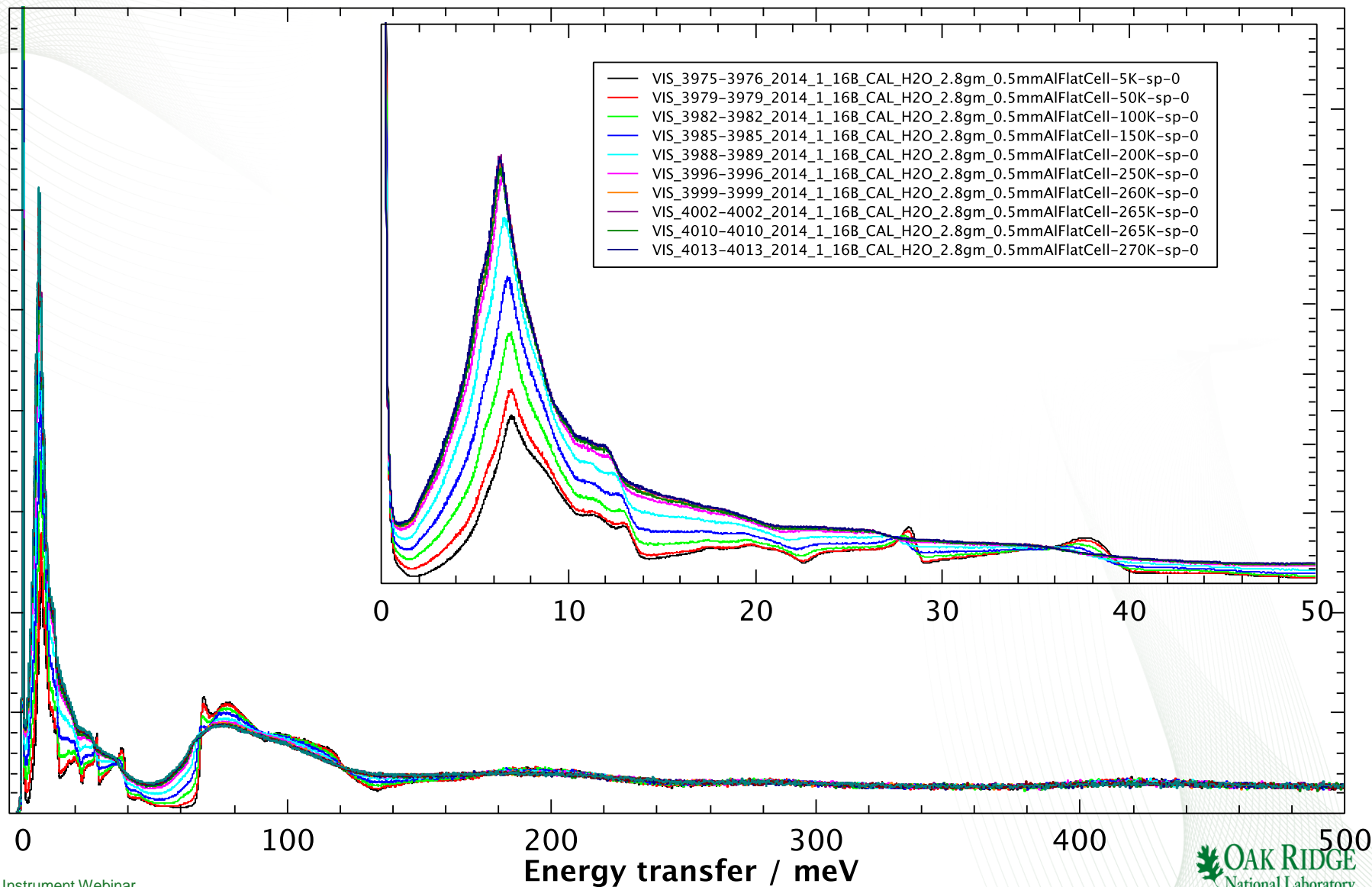
# Water (ice) in VISION Zoomed in



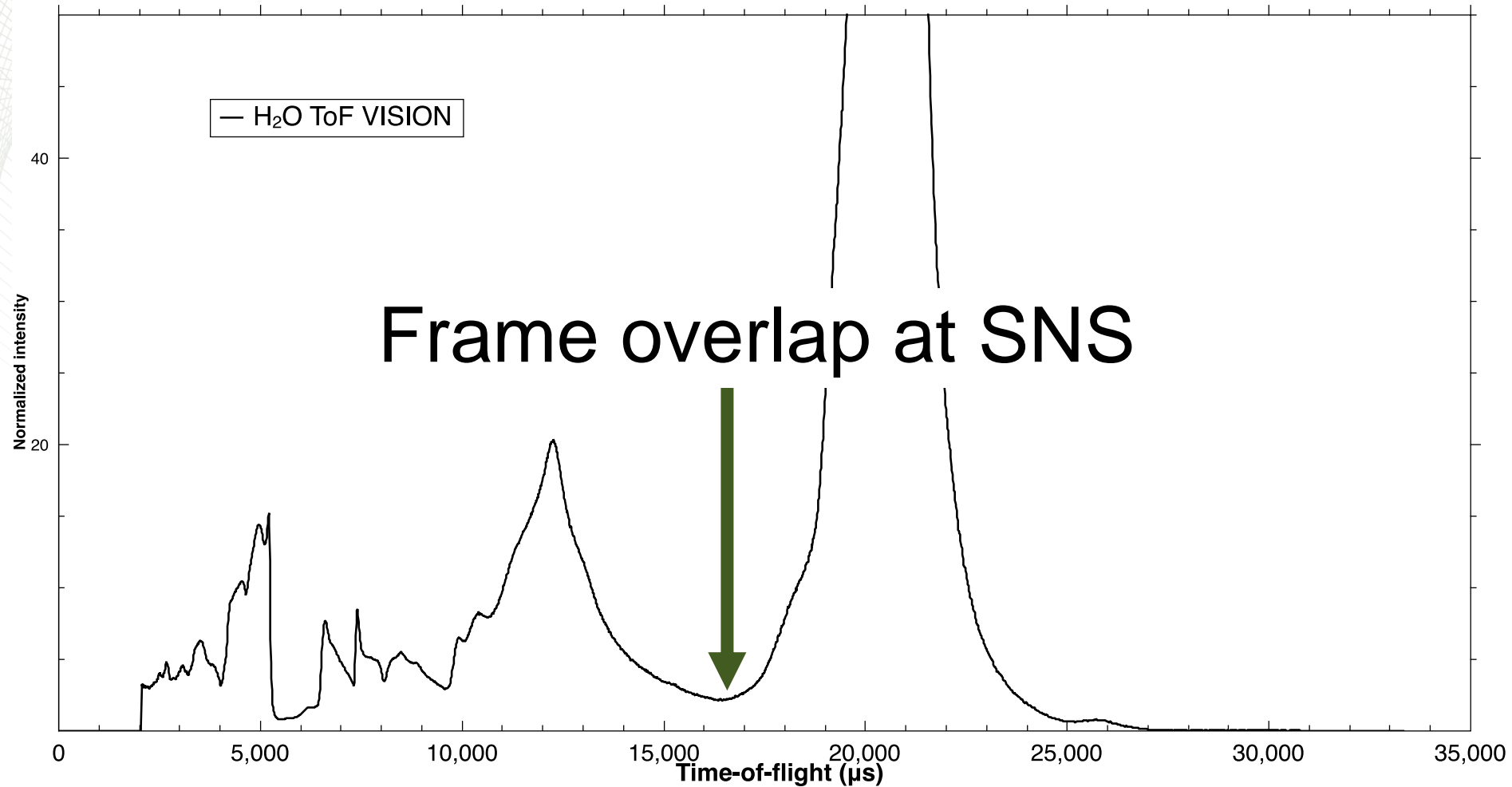
# Water (ice) in VISION Zoomed in



# Water in VISION (as function of temperature)



# Water (ice) in VISION converted in ToF

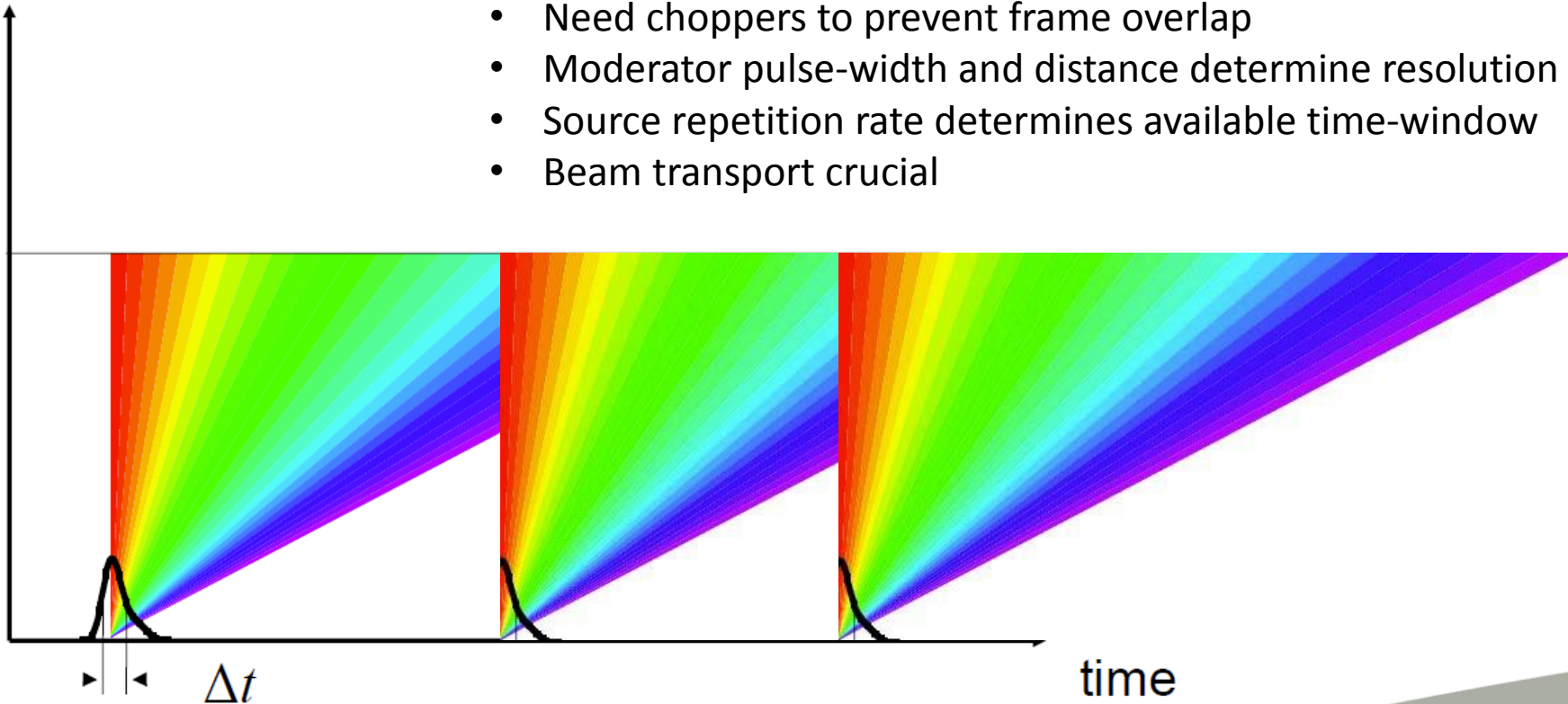


The  $T_0$  and frame overlap chopper are setup to run the instrument at 30Hz. This decision means that the intensity of the spectrum above 3.5 meV is reduced by 50%. But the benefits completely outweigh the reduction in flux.

# TOF method

distance

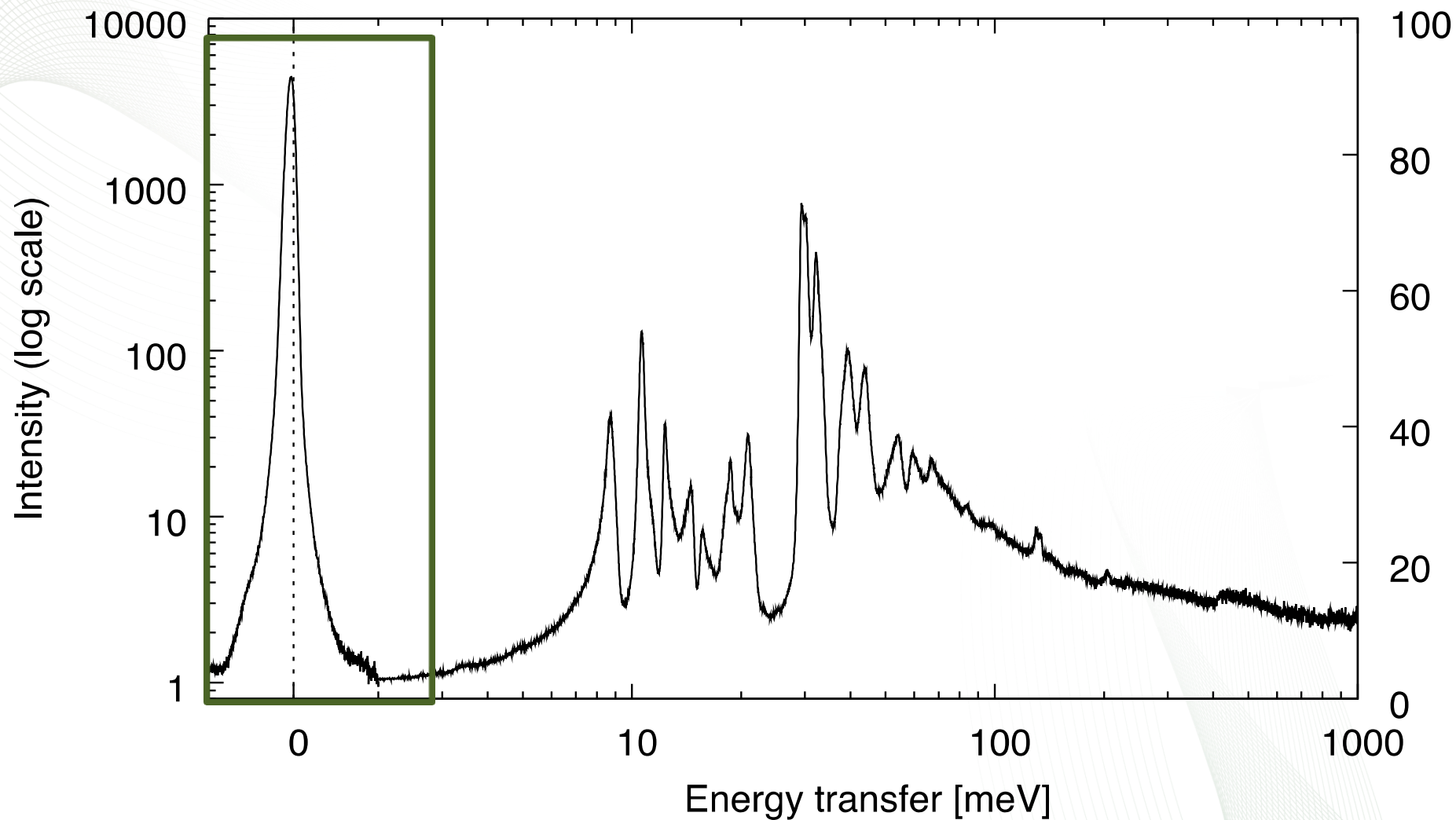
- Use distance to separate wavelengths
- Need choppers to prevent frame overlap
- Moderator pulse-width and distance determine resolution
- Source repetition rate determines available time-window
- Beam transport crucial



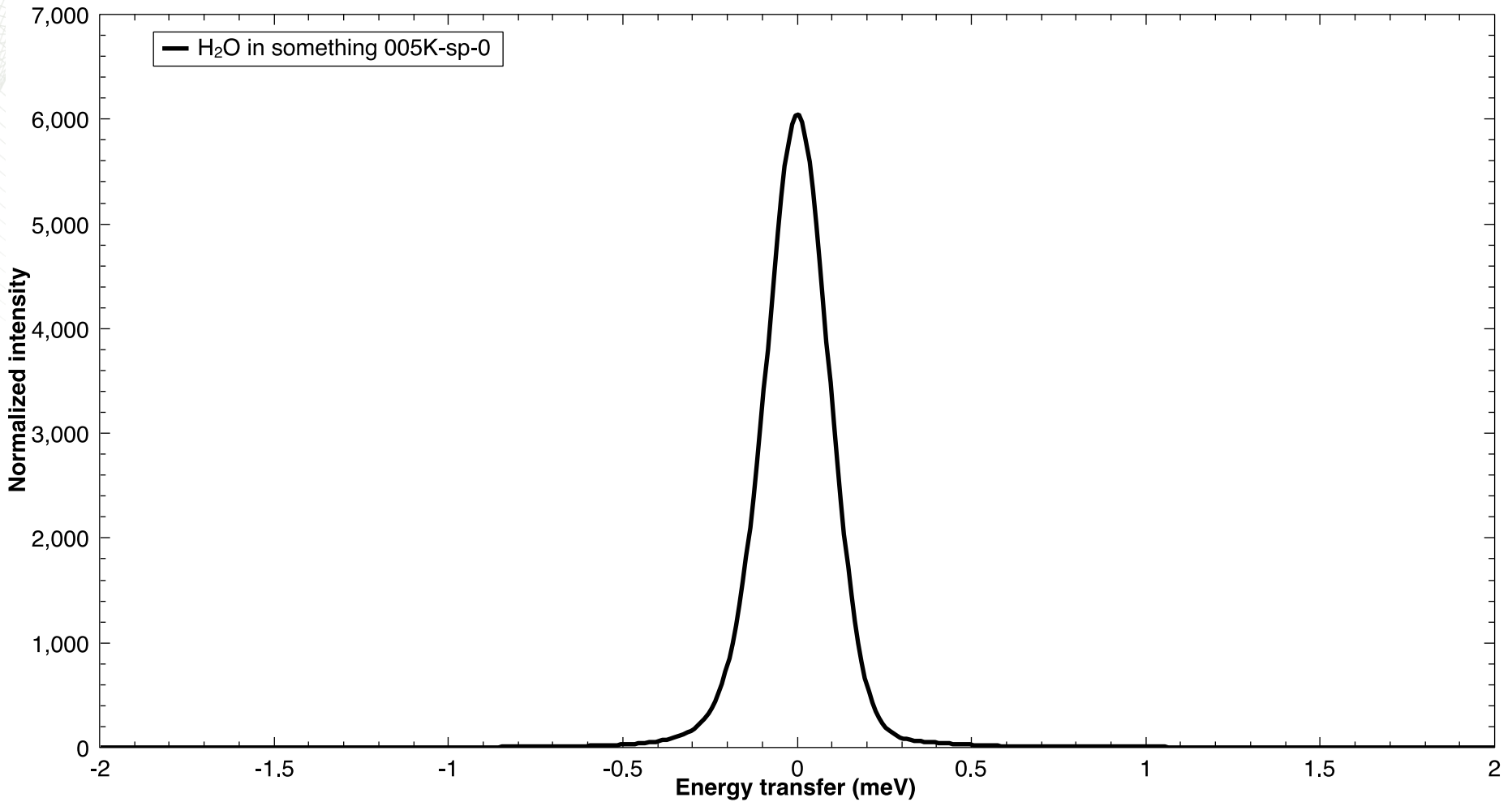
Science & Technology Facilities Council

ISIS

# What do I get around the elastic line?

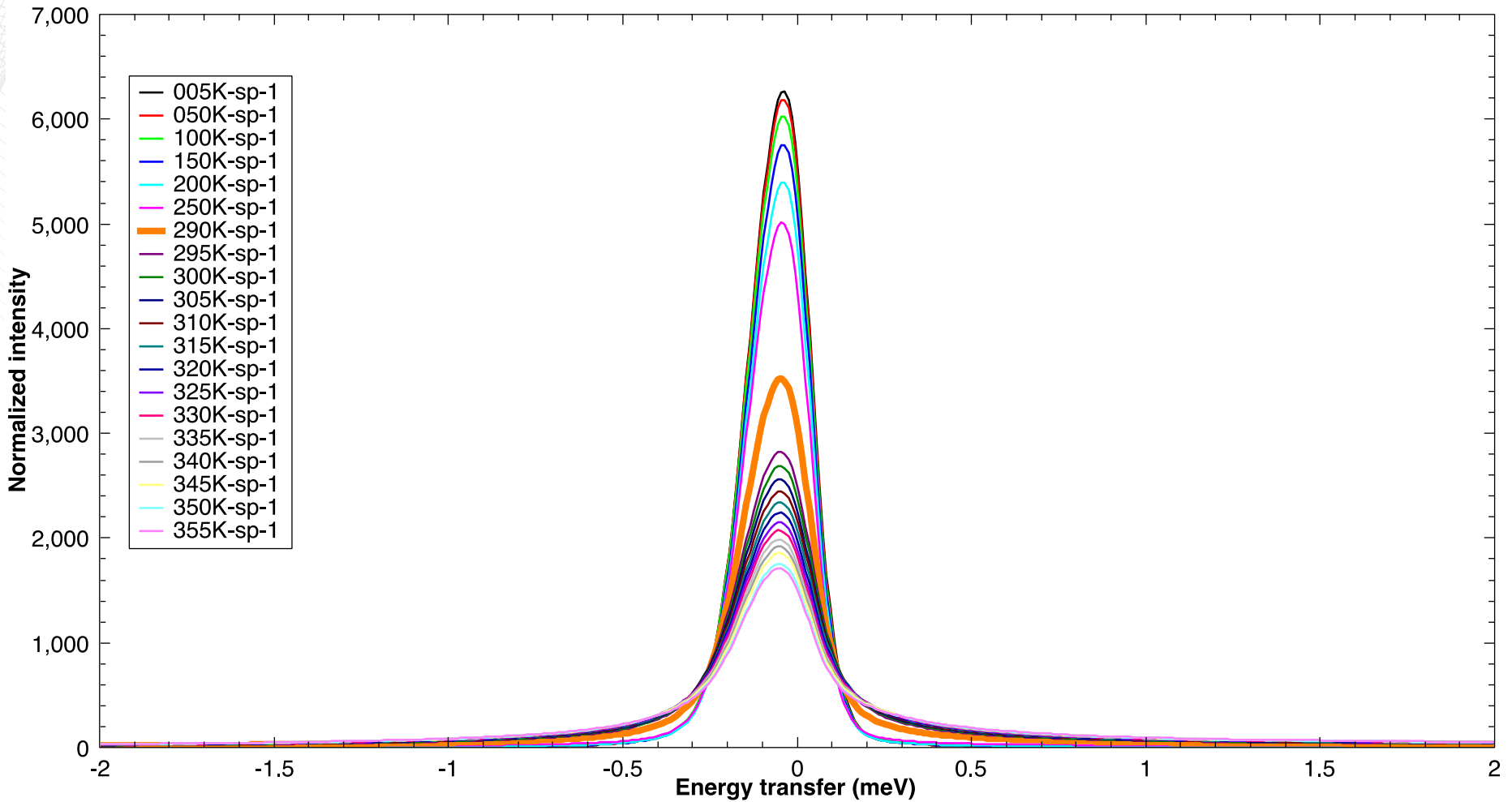


# What do I get around the elastic line?

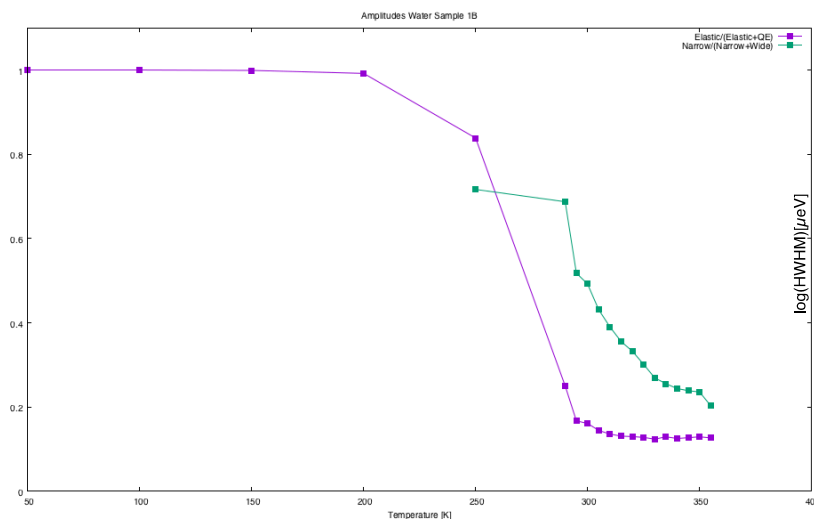
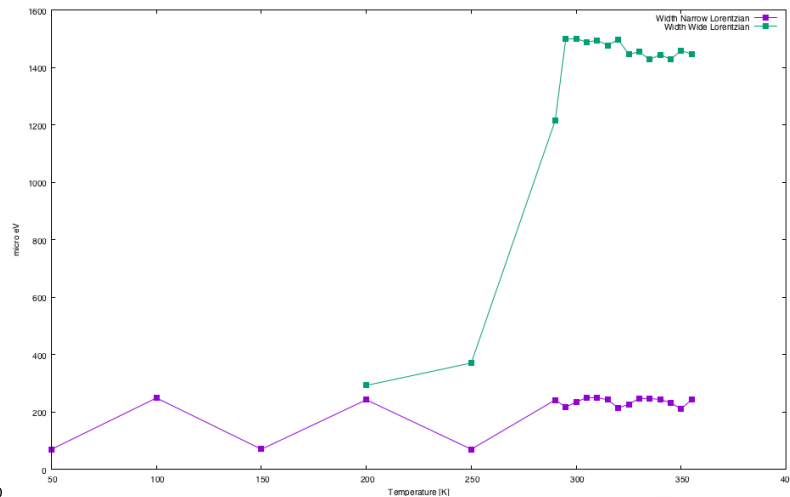
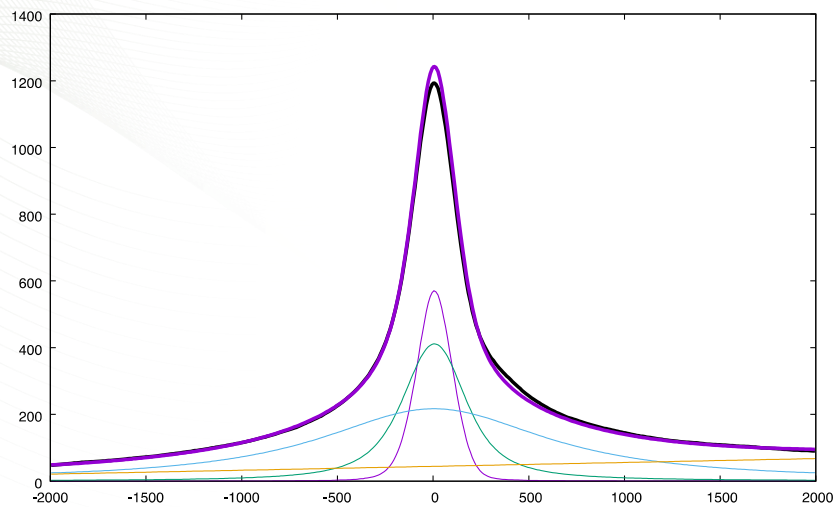




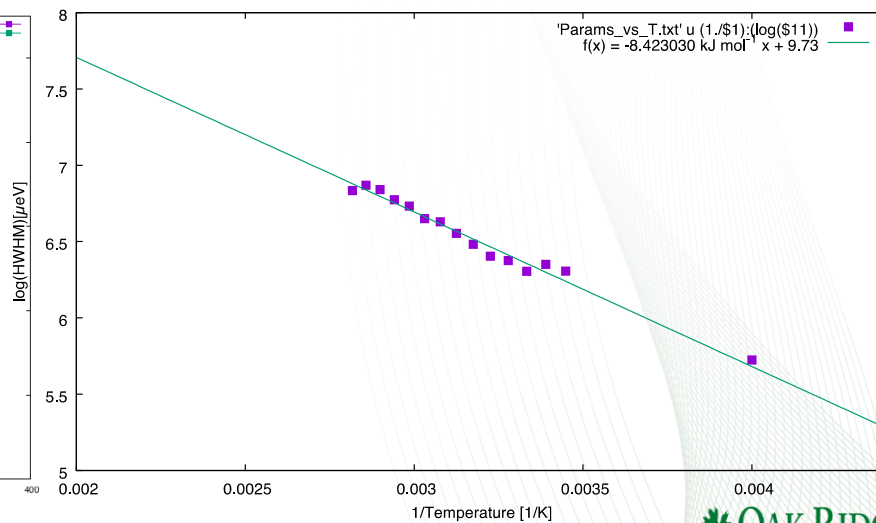
# What do I get around the elastic line?



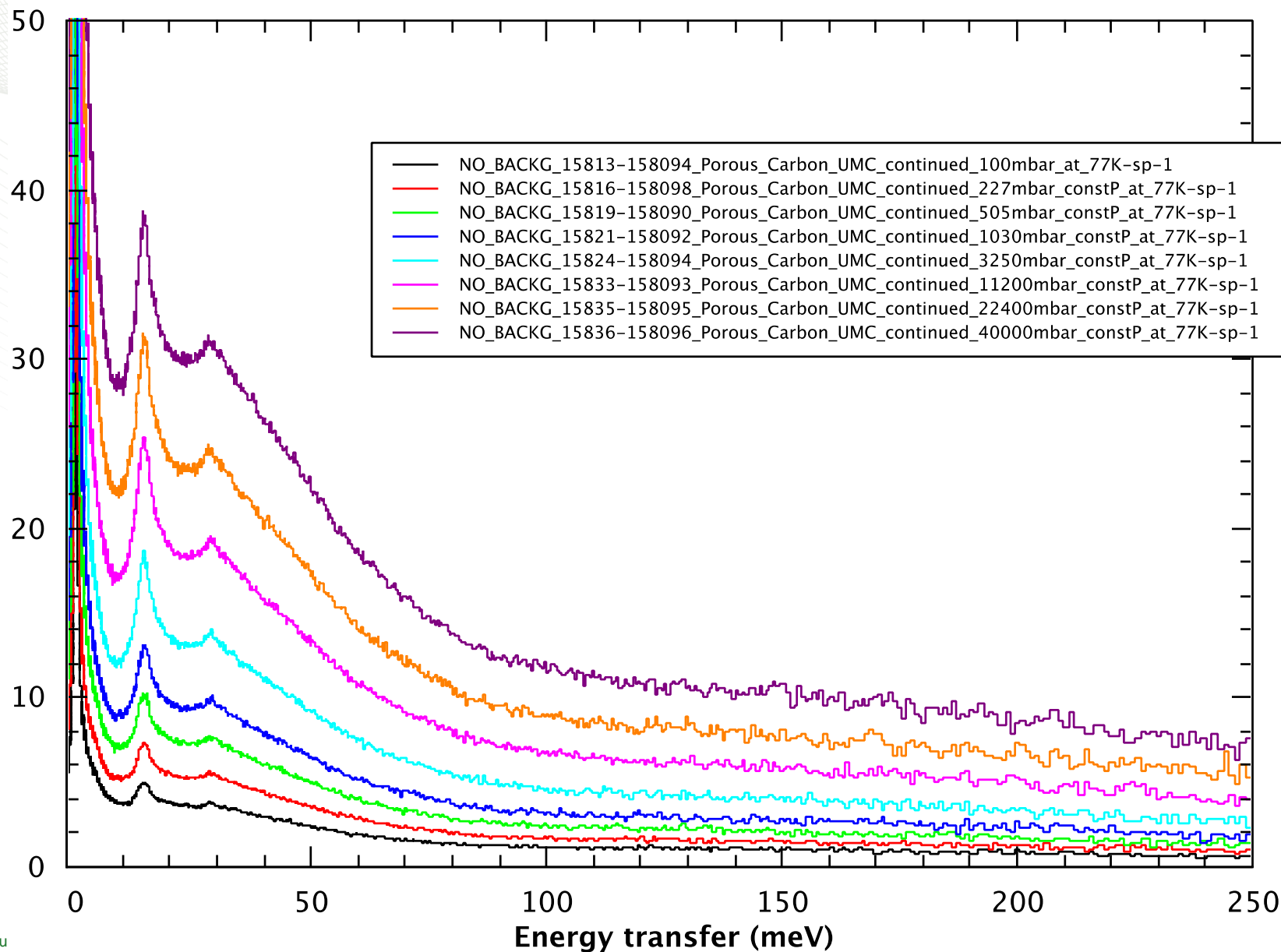
# What do I get around the elastic line?



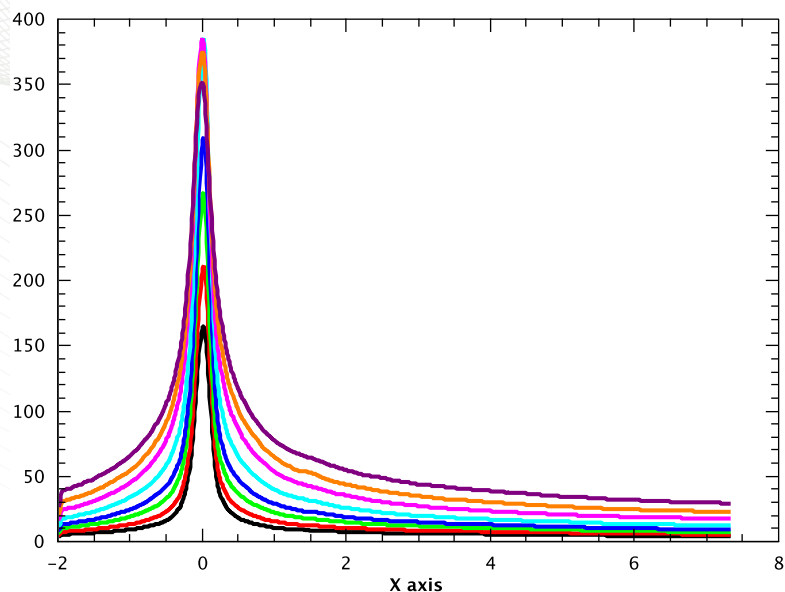
Arrhenius Plot Wide Sample 2B



# Molecular hydrogen in porous carbon



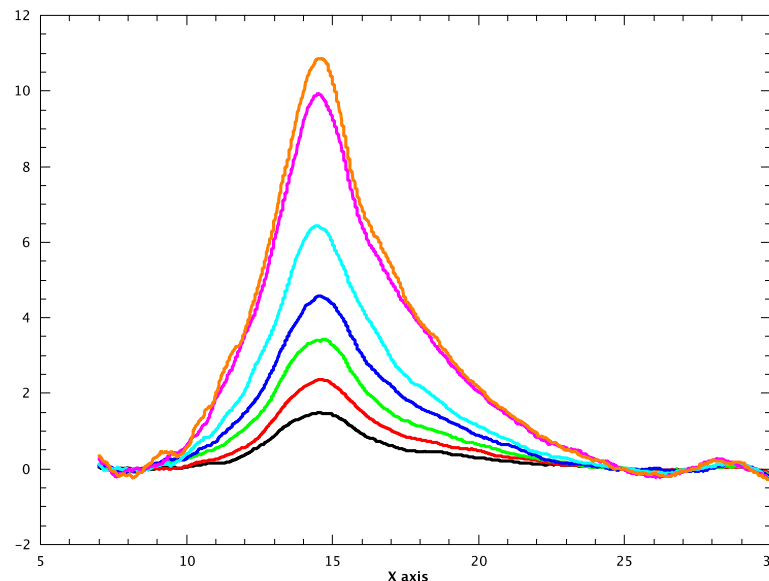
# Molecular hydrogen in porous carbon



Presence of the rotor line at 77K is indication of completely immobile molecular hydrogen in the pores. In the case of pure para-hydrogen (previous figure) the line disappears when the hydrogen melts.

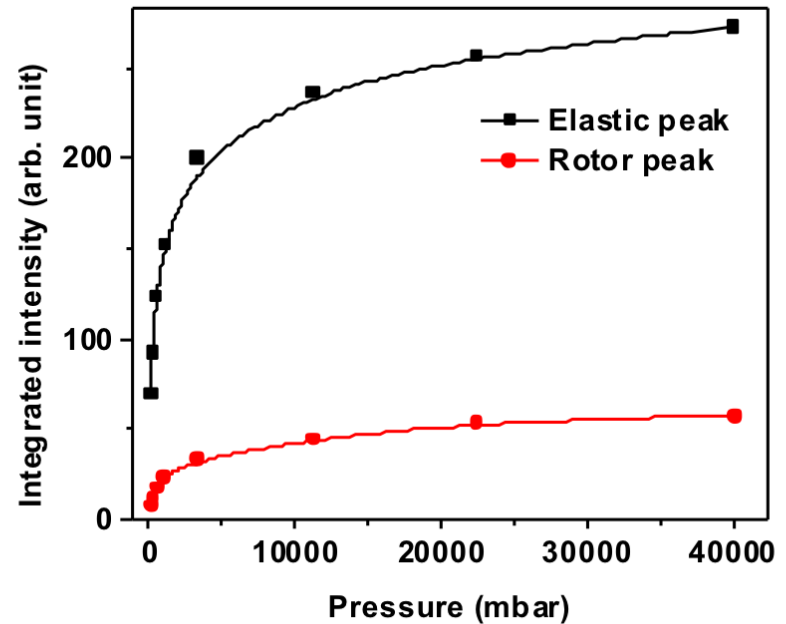
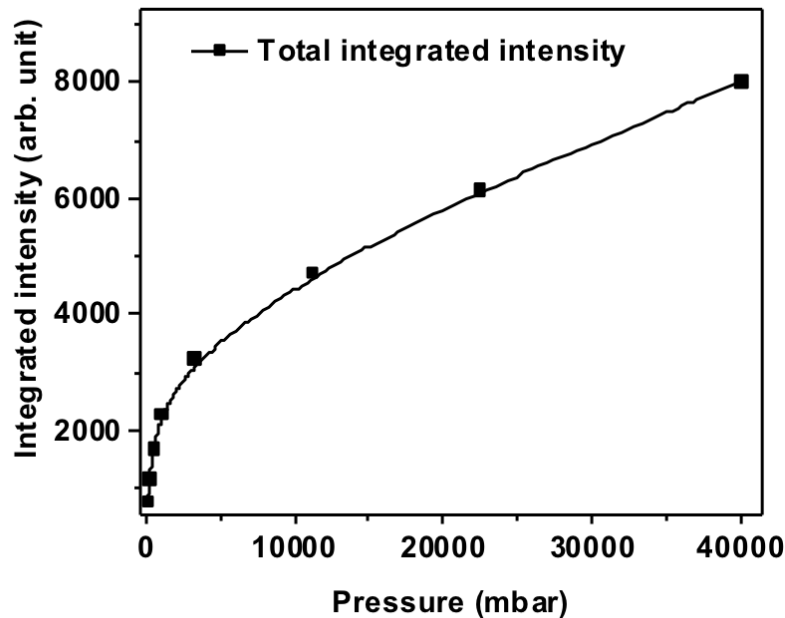
There is very little broadening of the rotor line, since the momentum transfer is larger than the corresponding one at the elastic line (dynamical trajectory of indirect geometry). The load keeps increasing even at 40 bar.

Presence of elastic line at 77K is indication of highly dense molecular hydrogen in the pores. The broadening of the elastic line is a consequence of the enhanced mobility of the molecules as the amount of hydrogen increases in the system. Larger pores, where hydrogen is less constrained have more mobility. In the gas the signal is extremely broad.

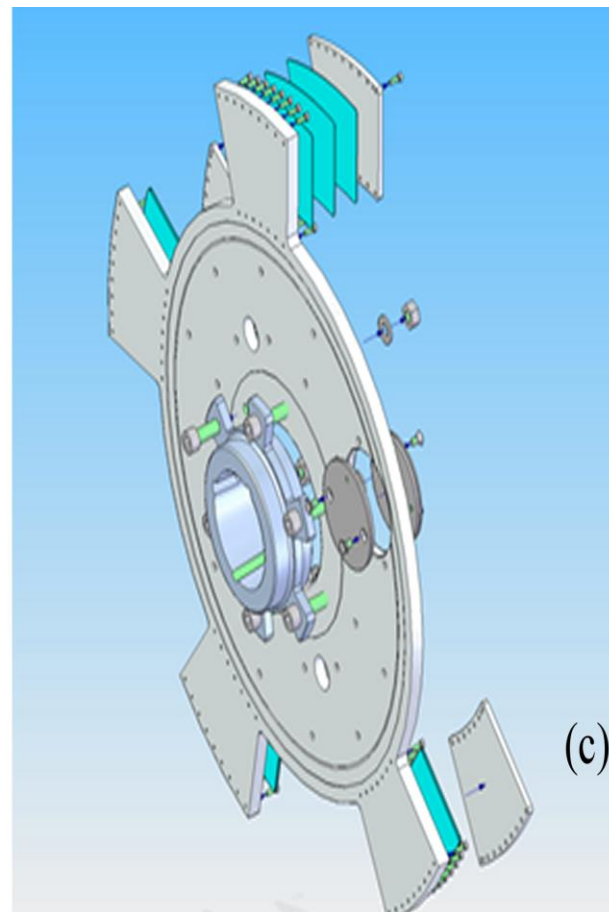
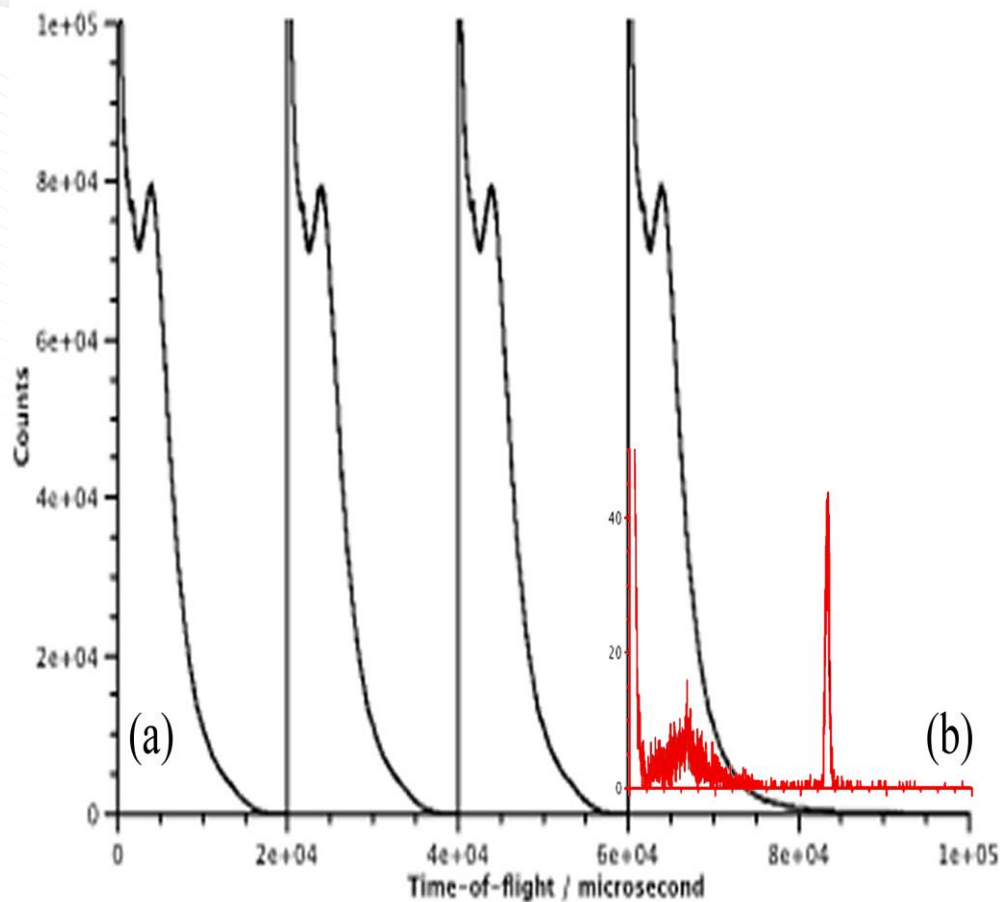


# Molecular hydrogen in porous carbon

1. The total integral of the spectral intensity is proportional to the amount of hydrogen in the system (left plot)
2. The integrated area under the elastic peak is proportional to the amount of hydrogen that is in a liquid like and solid like phase (right panel)
3. The integrated area under the rotor line is proportional to the amount of hydrogen in solid like phase (right panel)



# TOSCA Elastic line

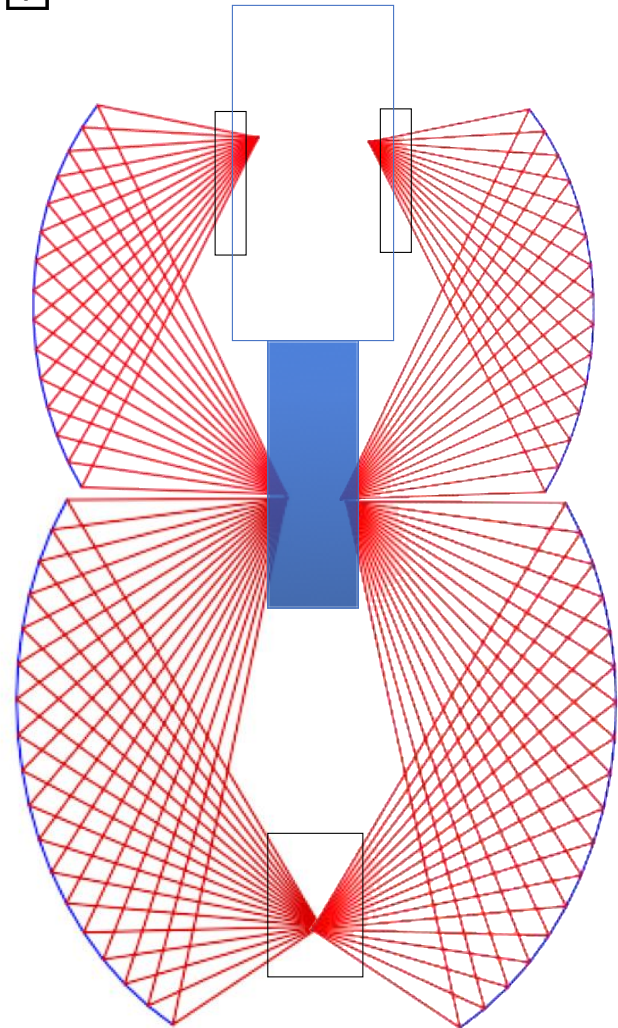


# SPHIINXS

## Spherical Indirect Neelastic Xtal Spectrometer (SPHIINXS)

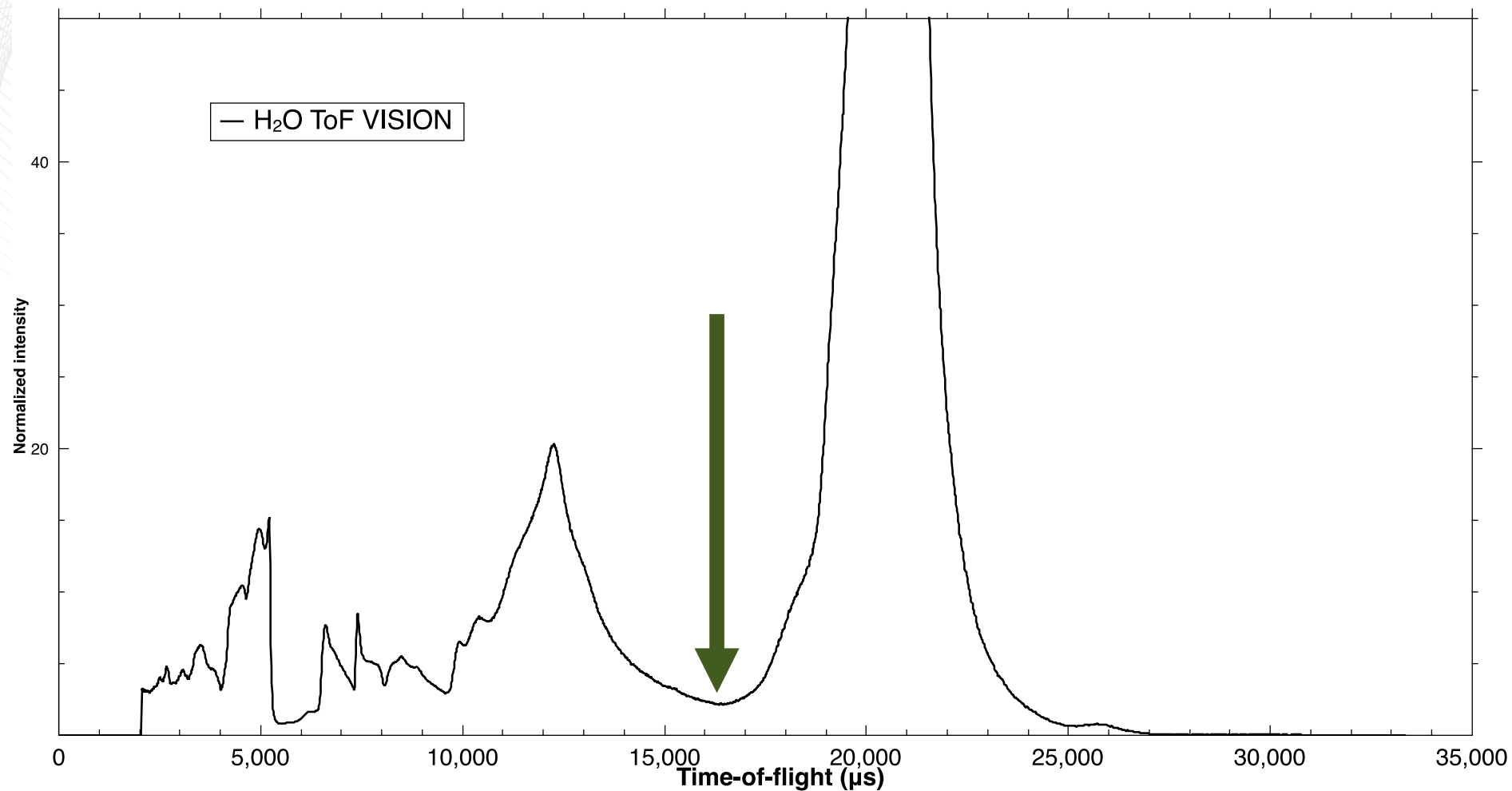
Expanding the VISION concept

- Graphite analysers on a parametric surface
- Range 3 meV to 1000 meV
- Large solid angle coverage  $\sim 10^3$  sr
- Focussing neutron guide (1 cm<sup>2</sup>)
- Powders and single crystal samples
- Energy resolution 1%  $\Delta\omega/\omega$  and  $\sim 70$   $\mu$ eV at the elastic line
- Positioned at 35-40 m from moderator



# Water (ice) in VISION converted in ToF

## Very simple “back of an envelope” calculation

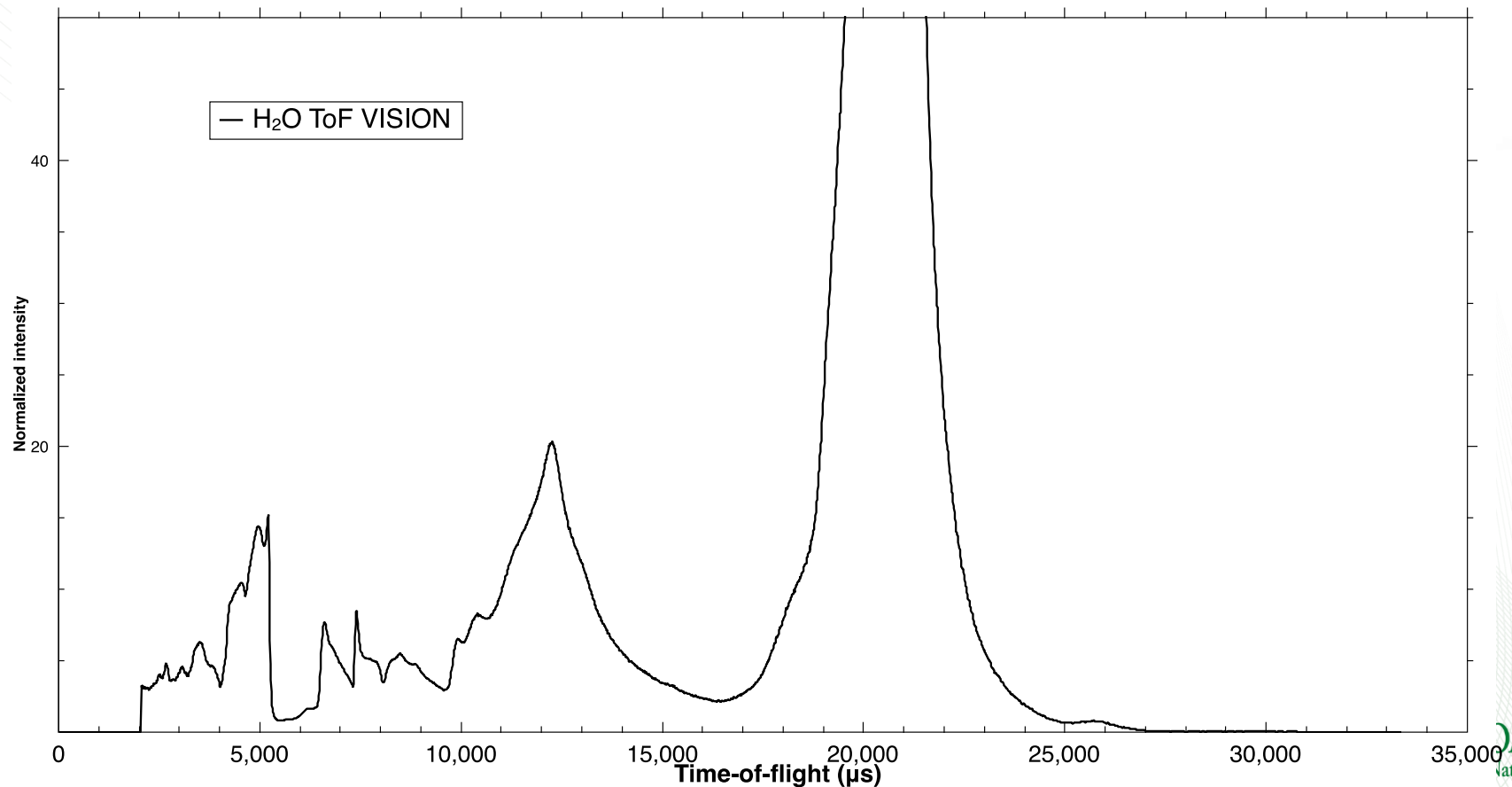




# Water (ice) in VESPA converted in ToF

Very simple “back of an envelope” calculation

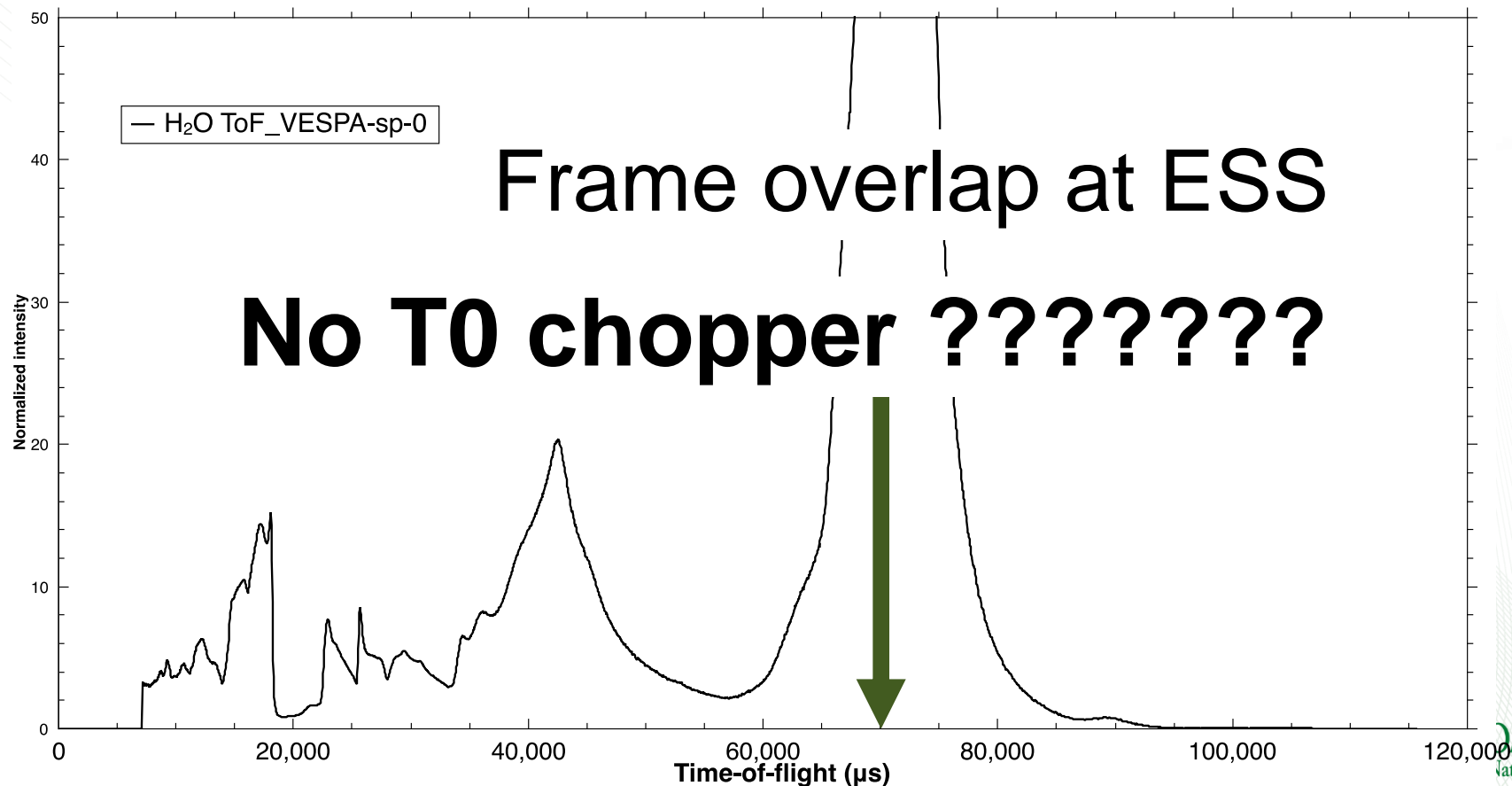
$$ToF_{VESPA} = \frac{L_{iVESPA}}{L_{iVISION}} ToF_{VISION}$$



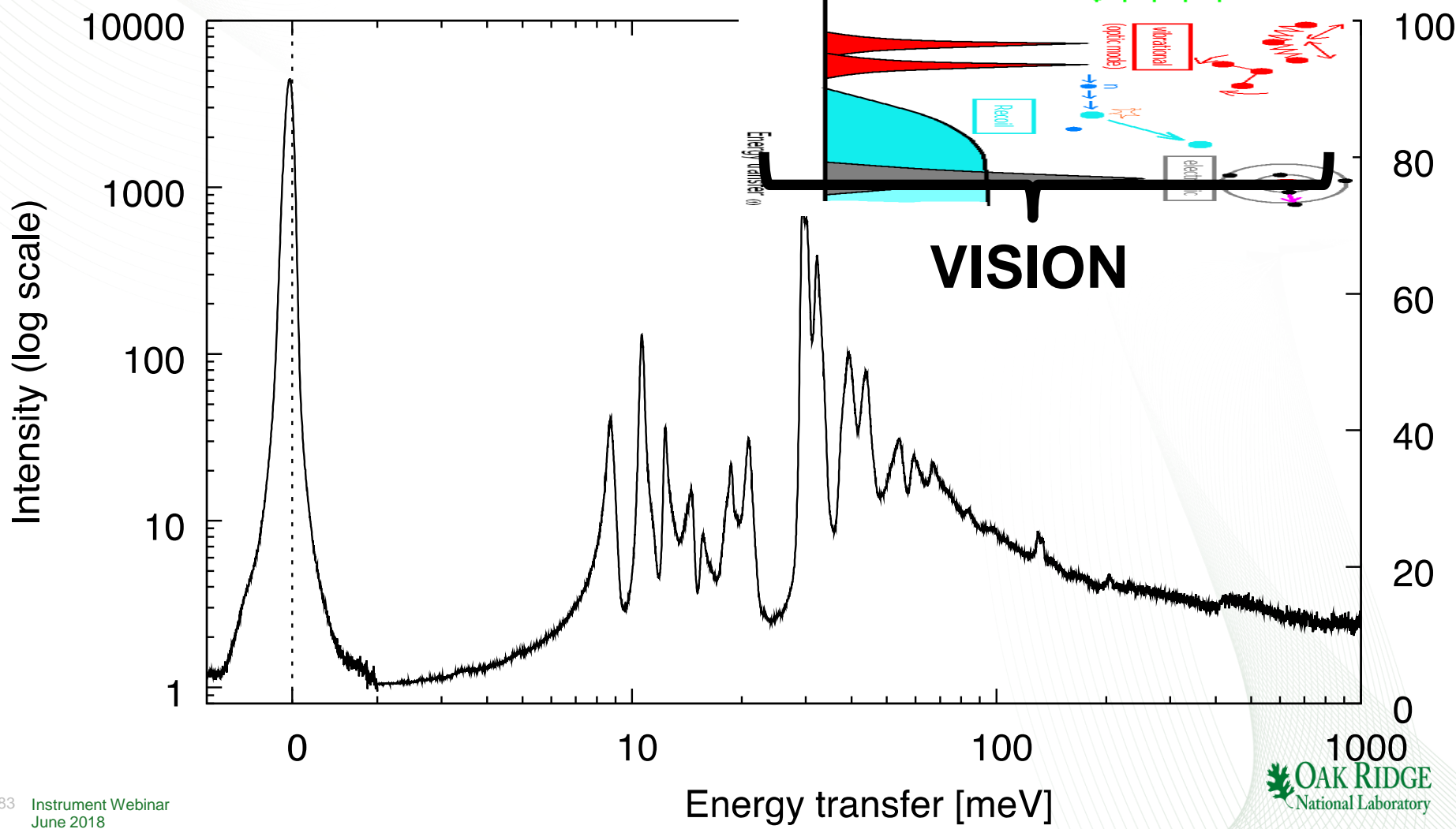
# Water (ice) in VESPA converted in ToF

Very simple “back of an envelope” calculation

$$ToF_{VESPA} = \frac{L_{iVESPA}}{L_{iVISION}} ToF_{VISION}$$

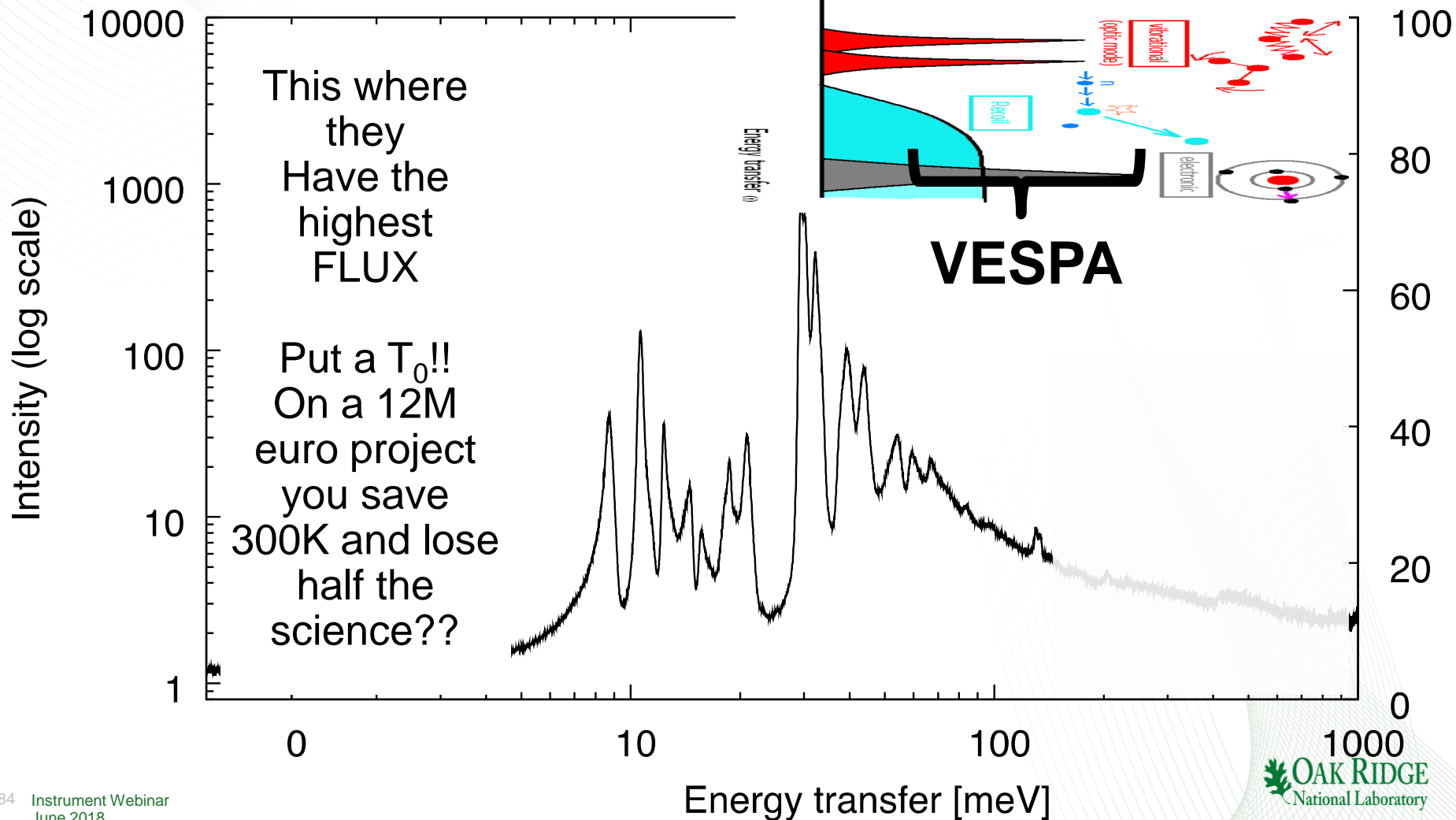


# The energy spectra (according to VISION)



# The energy spectra (according to VESPA)

No  $T_0$  chopper ????????



# VESPA vs VISION

## NO $T_0$ Chopper

**VISION Kills VESPA**  
Does it makes any sense at all to build VESPA?

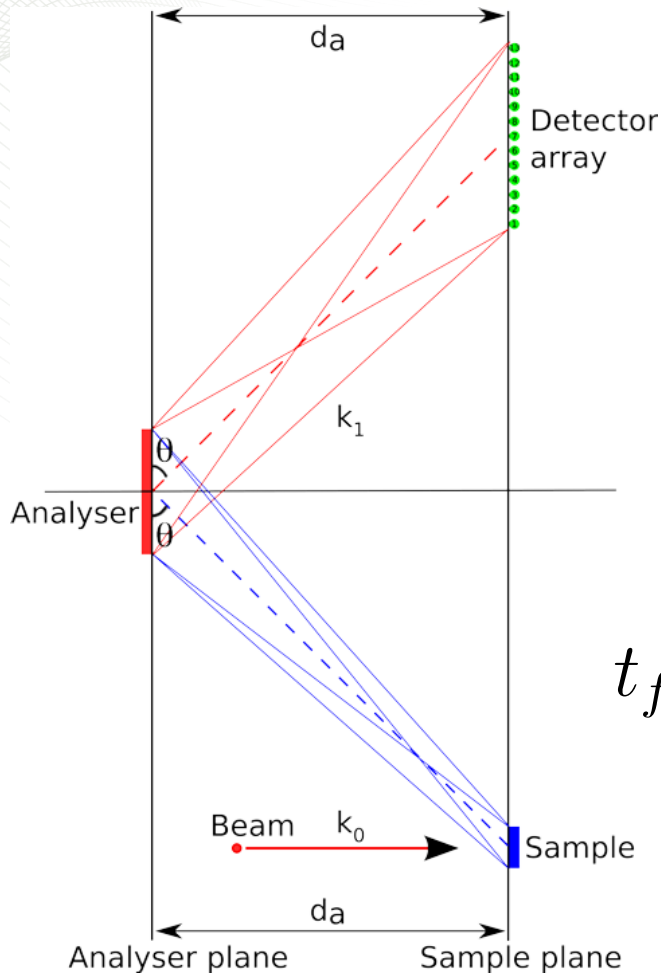
✓	✗	✓	✗	✓
---	---	---	---	---

## WITH $T_0$ Chopper

**VESPA beats VISION**  
in the QENS Regime  
**VESPA has world class capability**

✗	✓	✓	✓	✓
---	---	---	---	---

# Secondary analyzer

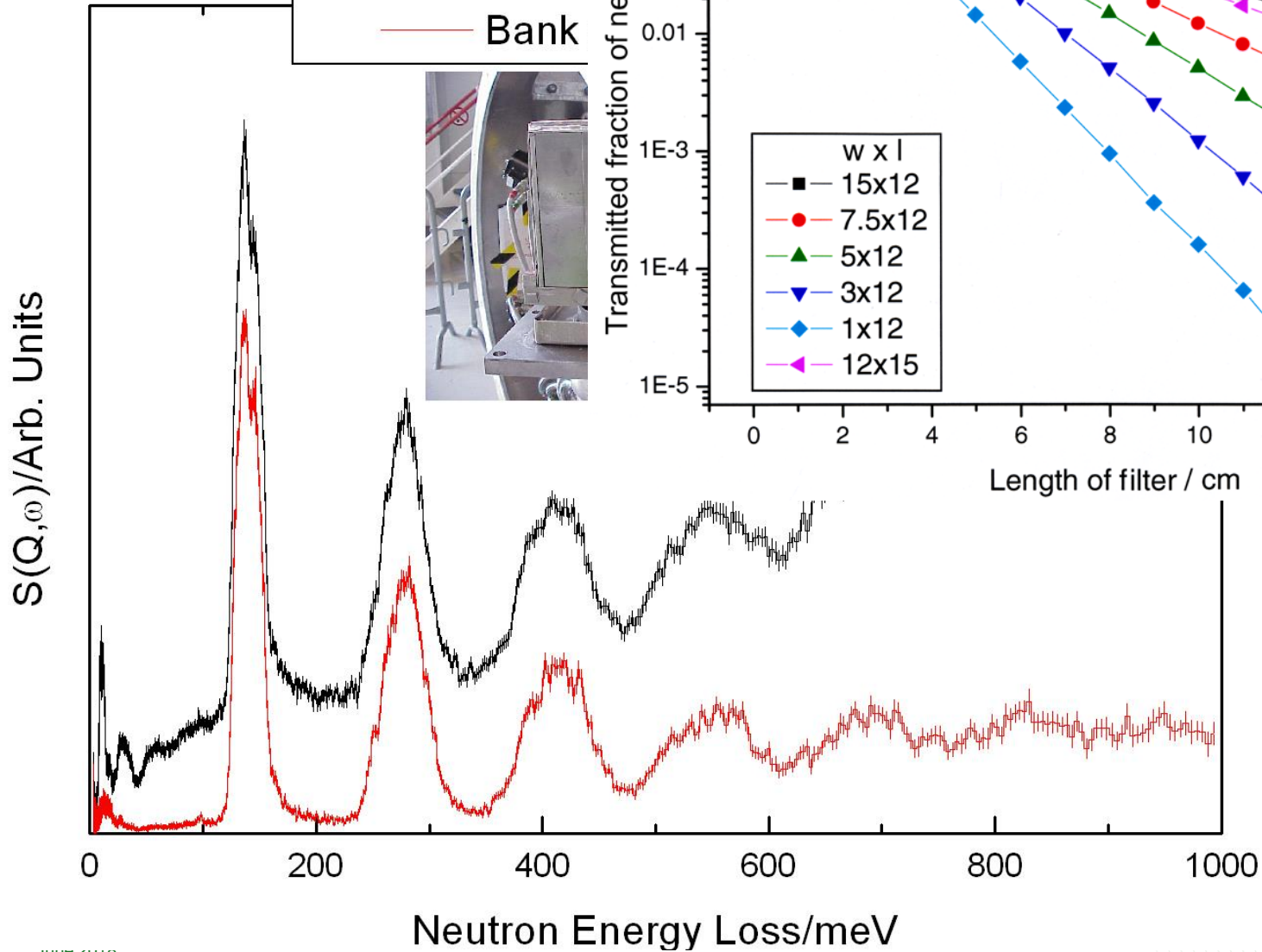


$$\lambda_f = 2d_{PG} \sin \theta$$

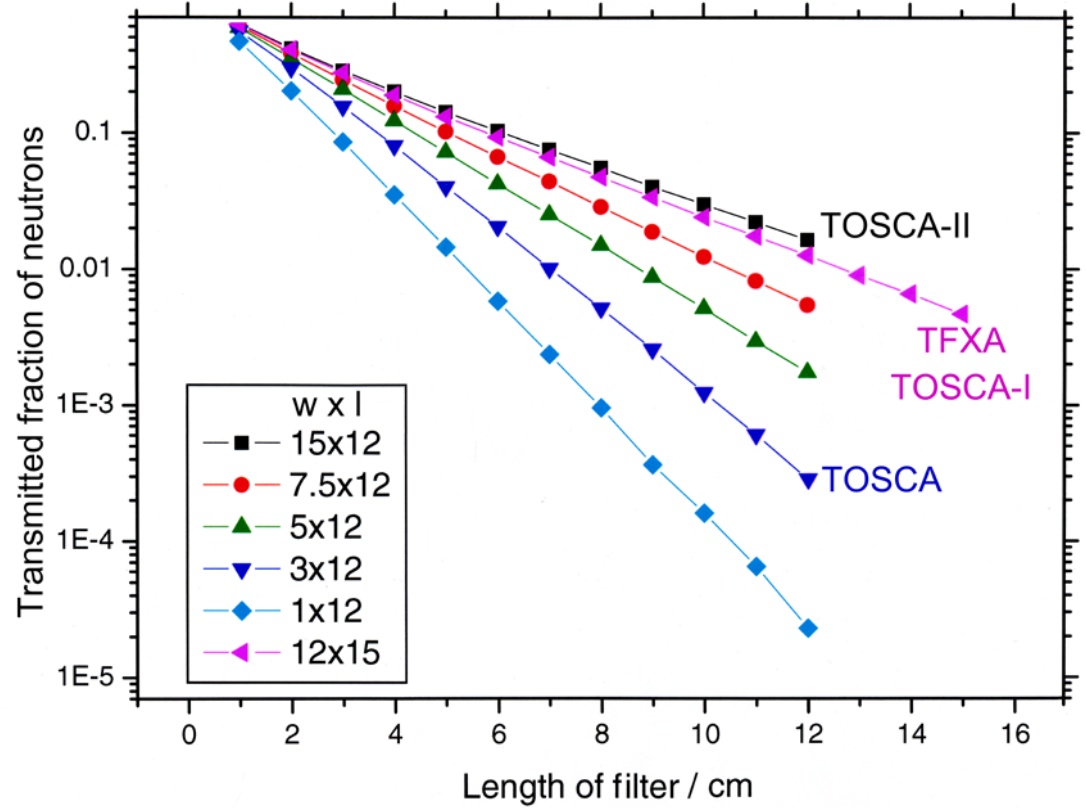
$$\frac{h}{m_n} \frac{t_f}{d_f} = 2d_{PG} \frac{2d_a}{d_f}$$

$$t_f = 4d_{PG}d_a \left( \frac{m_n}{h} \right) = \text{constant}$$

# Be Filter

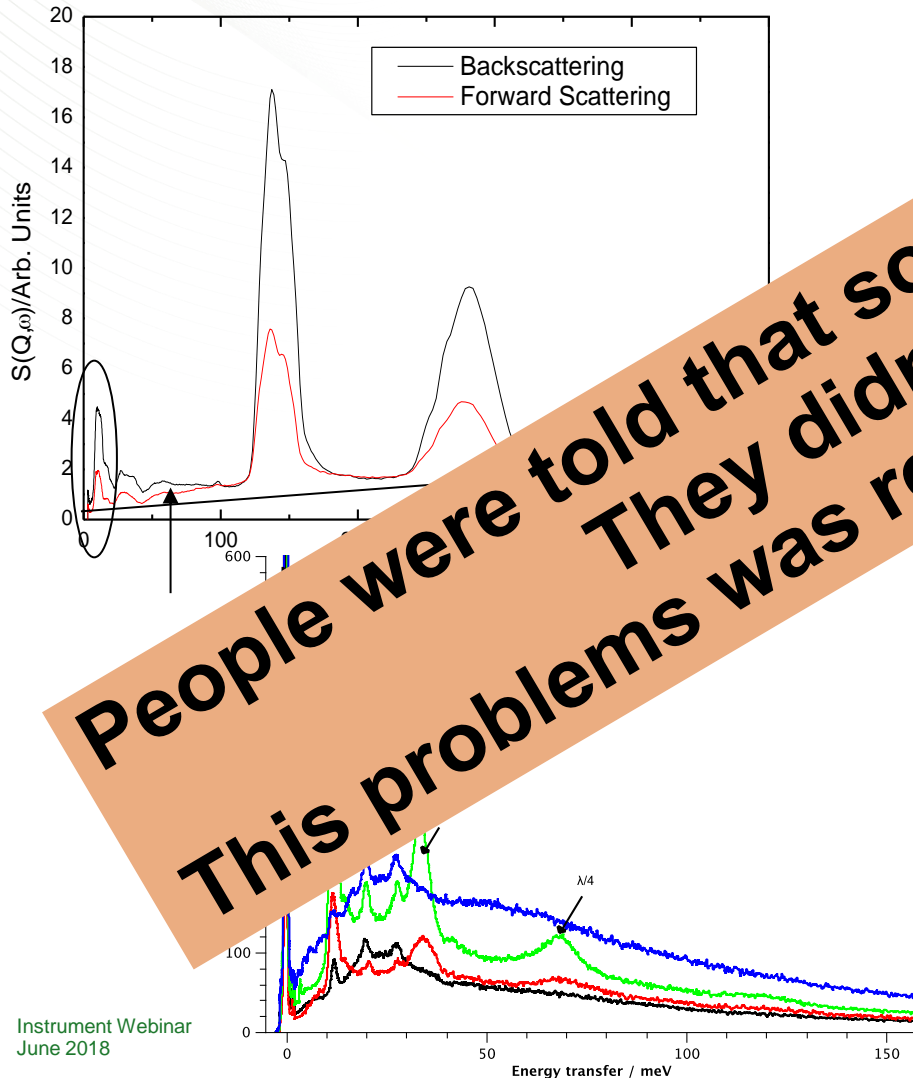


ZrH<sub>2</sub> Calibr  
— Bank  
— Bank

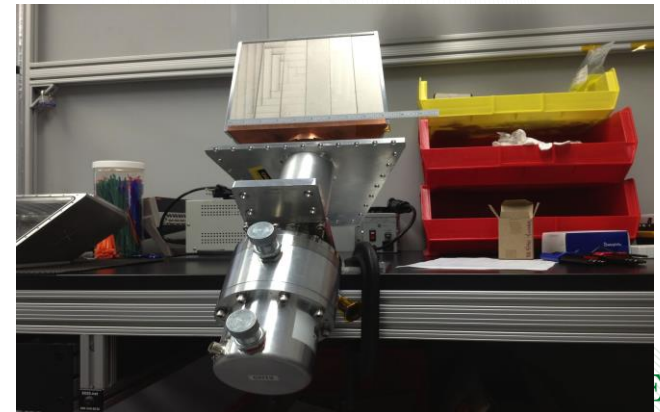


# REPEATING MISTAKES! TOSCA & VISION

- We have a non-existing feature at low energies and a big increase of the signal at high energy transfers.
- The effect is most noticeable in forward scattering



People were told that something was wrong  
They didn't listen  
This problem was relatively easy to solve!





STEVEN SPIELBERG PRESENTS



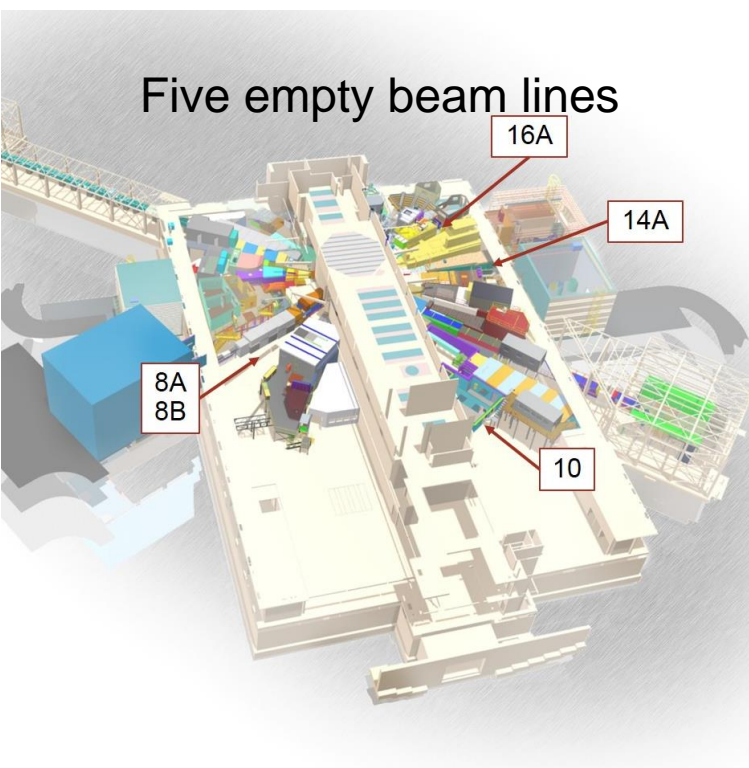
# BACK TO THE FUTURE

PG

A ROBERT ZEMECKIS FILM



# High priority instruments at the SNS First Target Station will leverage its high wavelength resolution strength



<b>HIGGS</b>	Inverse geometry spectrometer	BL-8A dc-p H <sub>2</sub> O
<b>MICRON</b>	Compact, texture, special purpose diffractometer	
<b>DISCOVER</b>	Medium resolution/flux diffractometer	BL-8B dc-p H <sub>2</sub> O
<b>VENUS</b>	Time-of-flight neutron imaging station	BL-10 dc-p para-H <sub>2</sub>
<b>INVENT</b>	Concept development station	BL-14A c para-H <sub>2</sub>
<b>SANS/GI-SANS</b>	SANS and/or GI-SANS	
<b>BeFAST</b>	Beryllium filter spectrometer	BL-16A dc-p H <sub>2</sub> O
<b>HiResPD</b>	High Resolution Powder Diffractometer	Needs dc-p para-H <sub>2</sub> 100 m flight path

dc-p: decoupled, poisoned

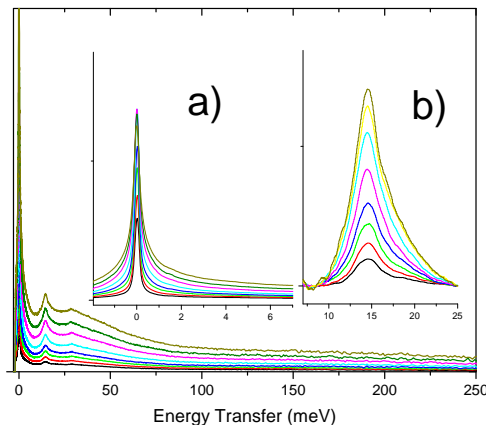
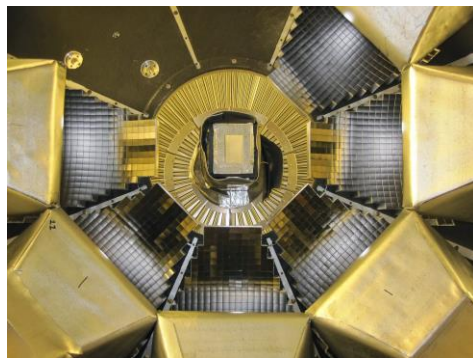
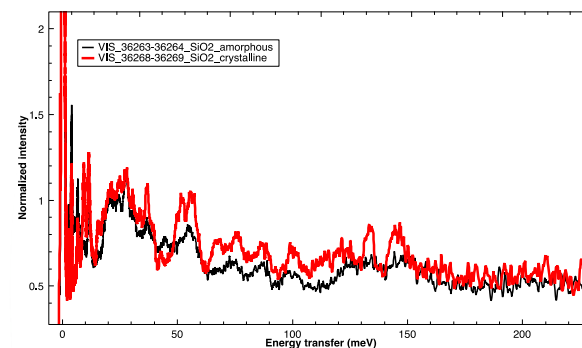
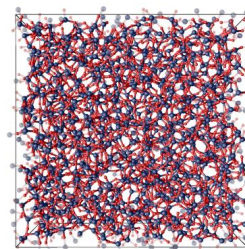
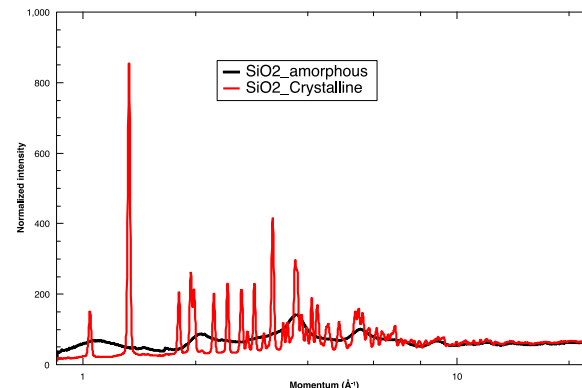
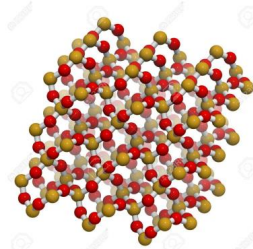
c: coupled

H<sub>2</sub>O: thermal neutrons

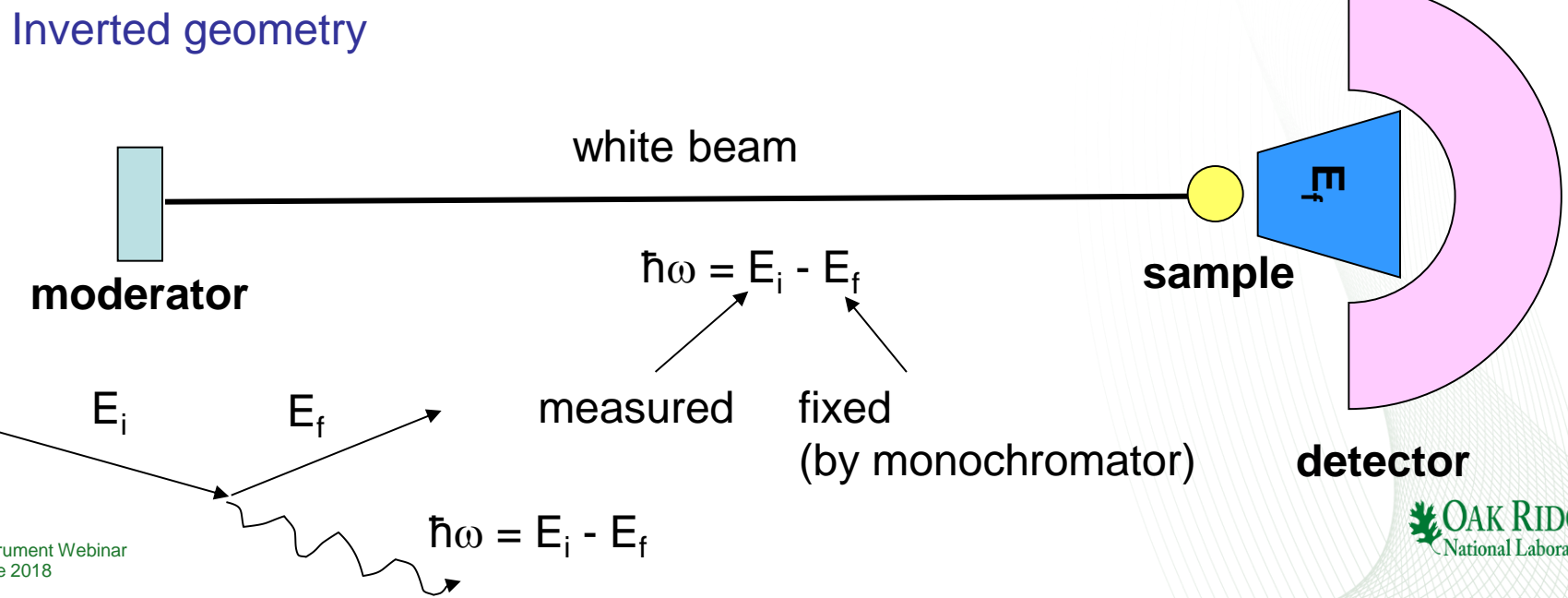
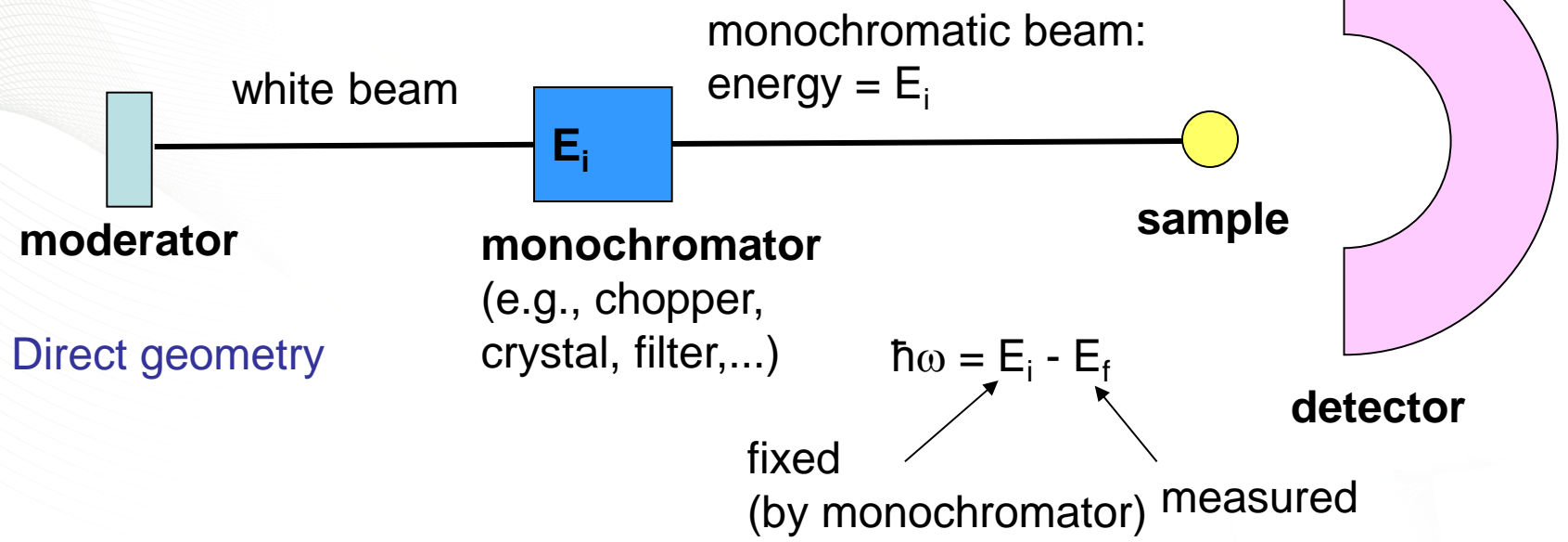
para-H<sub>2</sub>: cold neutrons

# VISION: Inelastic, Diffraction and QENS

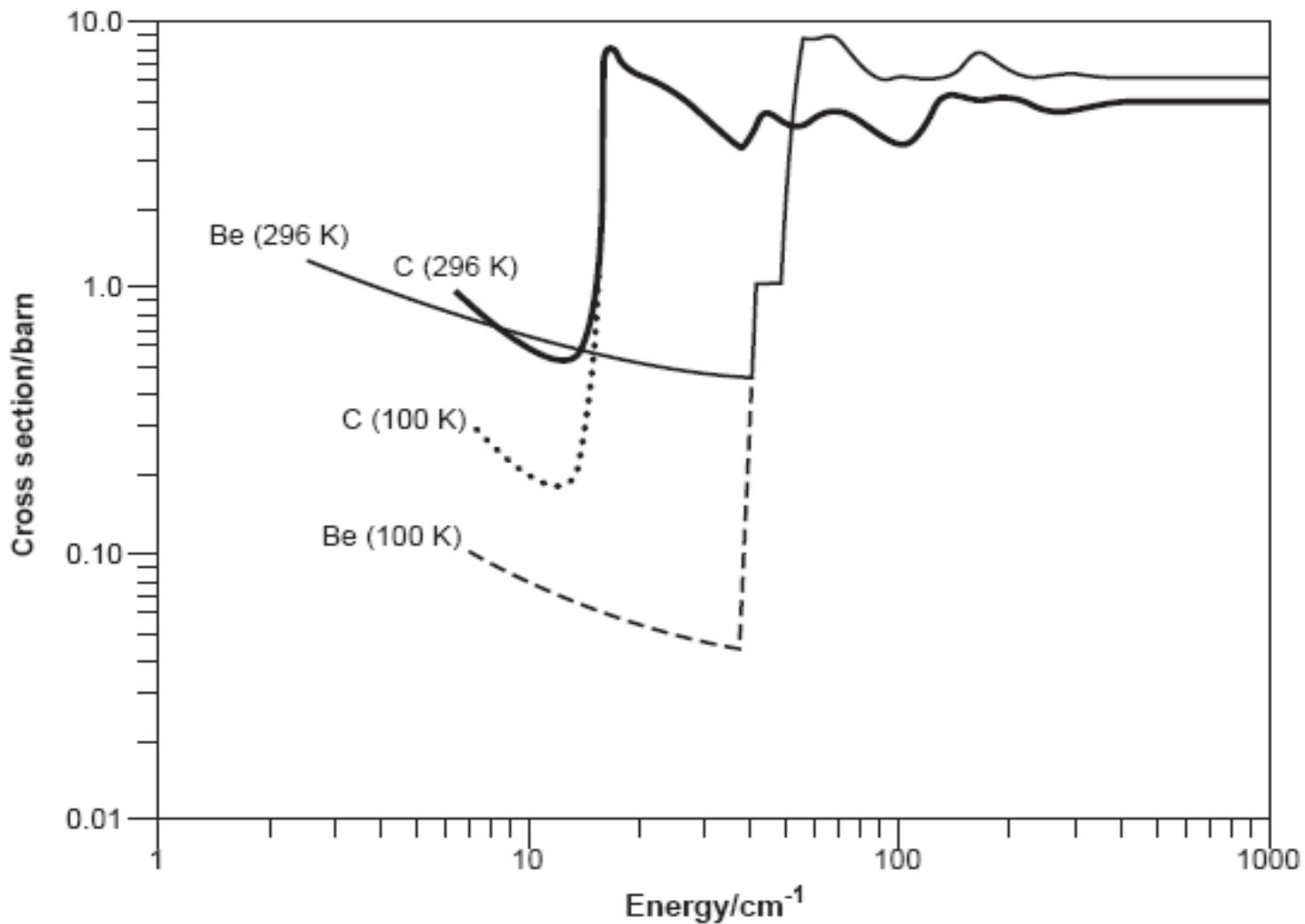
- Chemistry Oriented INS spectrometer
- White incident beam, fixed final energy (indirect geometry)
- High flux ( $\sim 5 \times 10^7$  neutrons/cm<sup>2</sup>/s) and double-focusing
- Broadband (-2 to 1000 meV at 30Hz, 5 to 500 meV at 60 Hz)
- Constant  $\Delta E/E$  throughout the spectrum ( $\sim 1.5\%$ )
- Elastic line HMF<sub>W</sub>  $\sim 120 \mu\text{eV}$
- Backward and 90° diffraction banks
- 4000 x its predecessor



## Spectrometer configurations



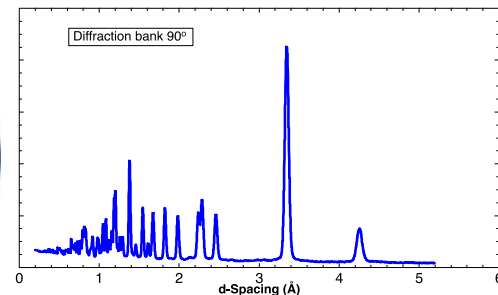
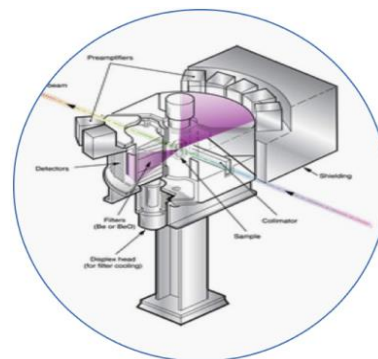
# Be filter



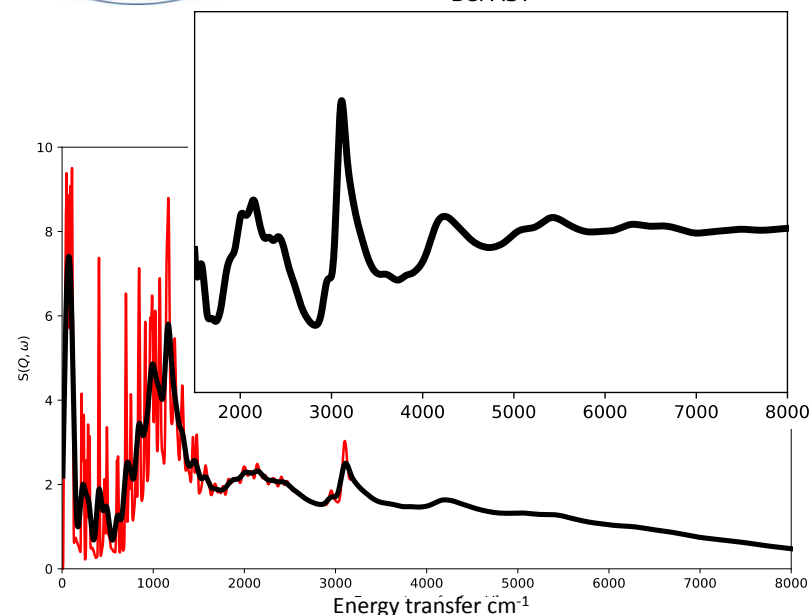
# BeFAST – Beryllium Filter Analyzer Spectrometer

- NScD Point-of-Contact: Timmy Ramirez-Cuesta
- Community Point-of-Contact: TBD
- Science Themes: catalysis, “real-world” systems
- Day 1 capabilities: BeFAST will measure in situ reactions by tracking CH or OH stretching and bond breaking (data collection as fast as 10 min)

- Measure inelastic neutron scattering to 8000  $\text{cm}^{-1}$ , overlapping Raman and IR spectroscopy
- Complements the VISION spectrometer with over 3-order of magnitude signal gain in the range 3000 to 8000  $\text{cm}^{-1}$  with slightly lower resolution
- Will have a 90° diffraction bank
- Large solid angle coverage
- Very compact, probably only instrument that will fit in BL-16A
- Inexpensive and simple, can be built in a short time
- Low m value guide (for half of the length), and needs one frame-overlap chopper



BeFAST



Neutron inelastic spectra from triphenylmethane. Red is calculated for VISION, black for BeFAST. BeFAST increased flux above 1800 $\text{cm}^{-1}$

# Thank you!

















