



by

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Nuclear Scattering of Neutrons

Overview

Introduction and theory of neutron scattering

- Advantages/disadvantages of neutrons for scattering measurements
- Neutron properties
- Comparison with other structural probes
- Definition of scattering cross sections
- Fermi pseudopotential
- Elastic scattering and definition of the structure factor, $S(Q)$
- Coherent & incoherent scattering
- References

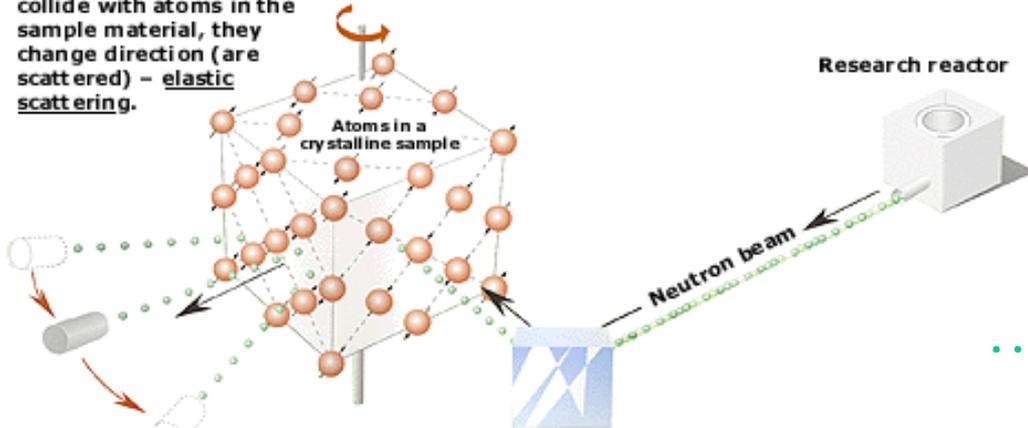
Why do Neutron Scattering?

- To determine the positions and motions of atoms in condensed matter
 - 1994 Nobel Prize to Shull and Brockhouse cited these areas
(see <http://www.nobel.se/physics/educational/poster/1994/neutrons.html>)
- Neutron advantages:
 - Wavelength comparable with interatomic spacings
 - Kinetic energy comparable with that of atoms in a solid
 - Penetrating => bulk properties are measured & sample can be contained
 - Weak interaction with matter aids interpretation of scattering data
 - Isotopic sensitivity allows contrast variation
 - Neutron magnetic moment couples to \mathbf{B} => neutron “sees” unpaired electron spins
- Neutron Disadvantages
 - Neutron sources are weak => low signals, need for large samples etc
 - Some elements (e.g. Cd, B, Gd) absorb strongly
 - Kinematic restrictions (can't access all energy & momentum transfers)

The 1994 Nobel Prize in Physics – Shull & Brockhouse

Neutrons show where the atoms are....

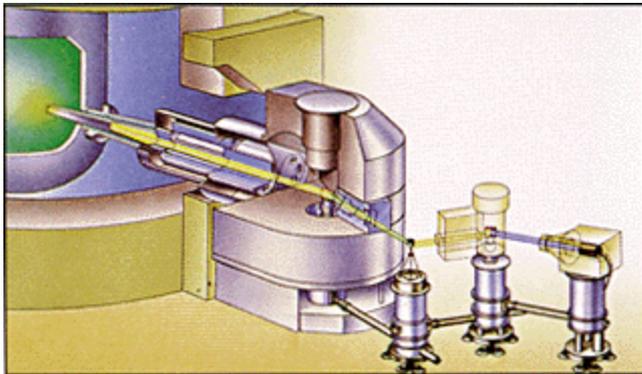
When the neutrons collide with atoms in the sample material, they change direction (are scattered) – elastic scattering.



Detectors record the directions of the neutrons and a diffraction pattern is obtained.

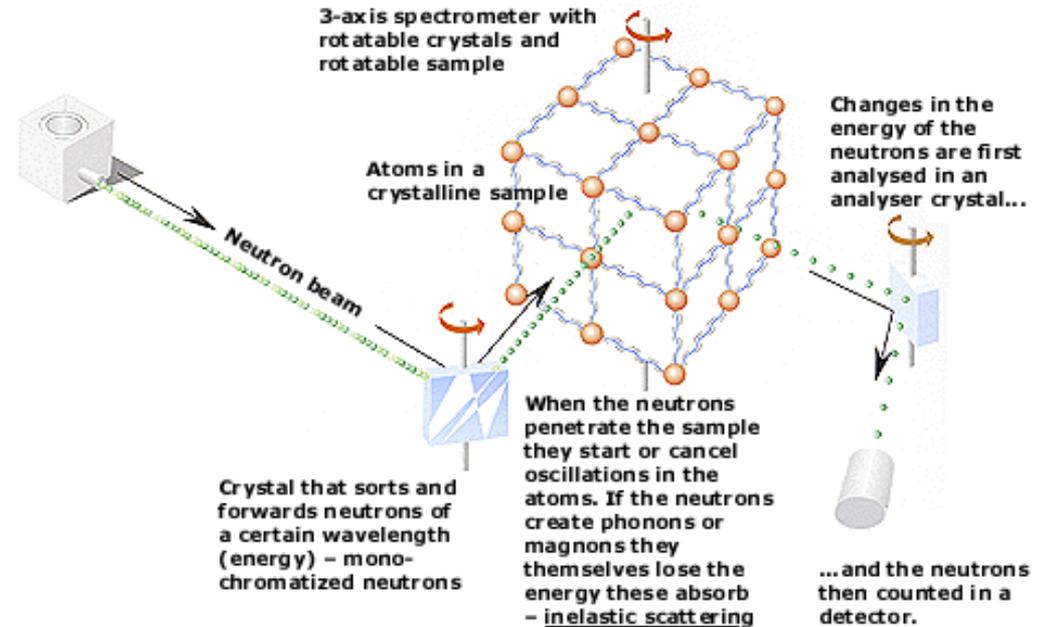
The pattern shows the positions of the atoms relative to one another.

Crystal that sorts and forwards neutrons of a certain wavelength (energy) – monochromatized neutrons



3-axis spectrometer

...and what the atoms do.



The Neutron has Both Particle-Like and Wave-Like Properties

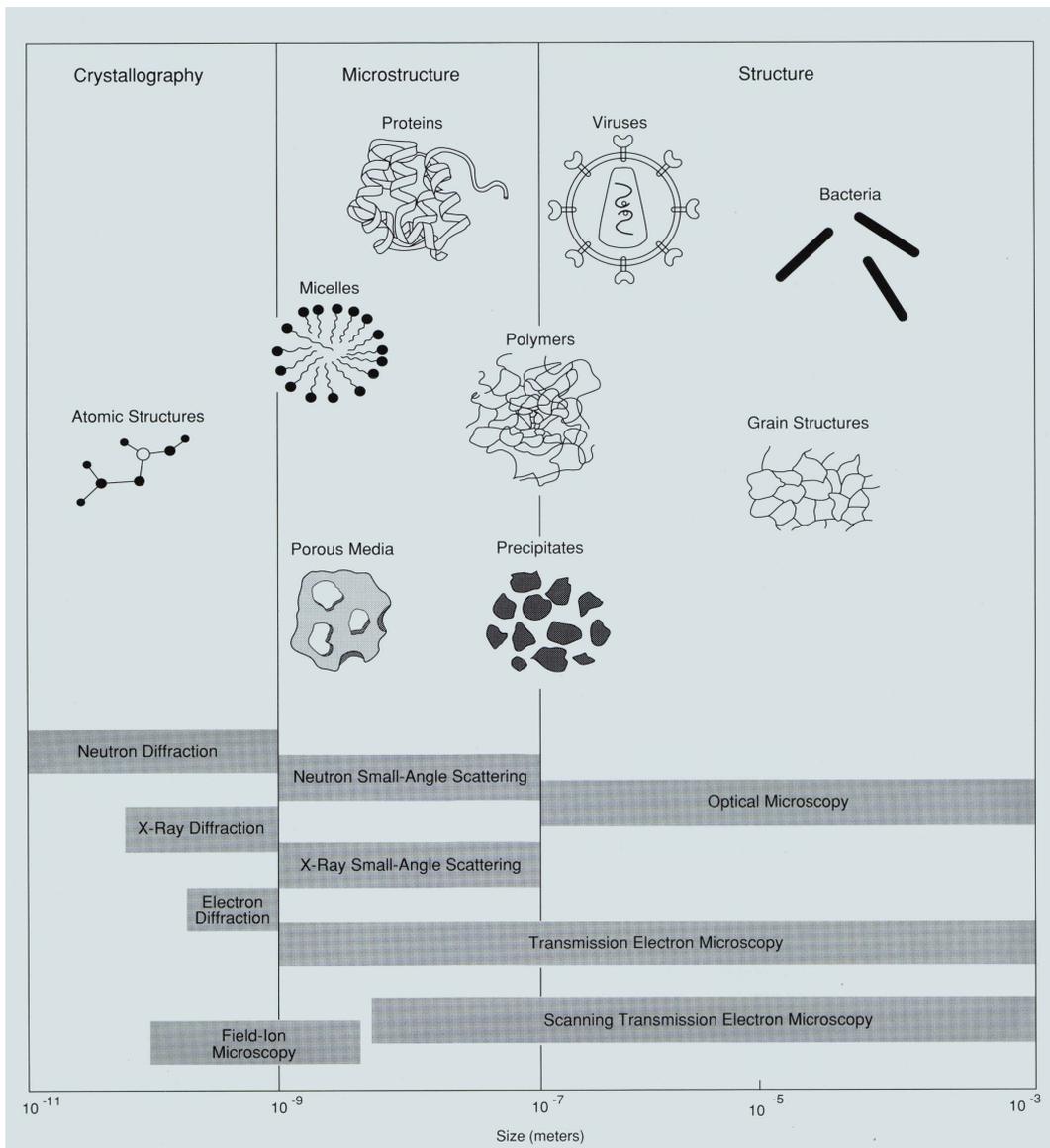
- Mass: $m_n = 1.675 \times 10^{-27}$ kg
- Charge = 0; Spin = $\frac{1}{2}$
- Magnetic dipole moment: $\mu_n = -1.913 m_N$
- Nuclear magneton: $\mu_N = eh/4\pi m_p = 5.051 \times 10^{-27}$ J T⁻¹
- Velocity (v), kinetic energy (E), wavevector (k), wavelength (λ), temperature (T).
- $E = m_n v^2/2 = k_B T = (hk/2\pi)^2/2m_n$; $k = 2\pi/\lambda = m_n v/(h/2\pi)$

	<u>Energy (meV)</u>	<u>Temp (K)</u>	<u>Wavelength (nm)</u>
Cold	0.1 – 10	1 – 120	0.4 – 3
Thermal	5 – 100	60 – 1000	0.1 – 0.4
Hot	100 – 500	1000 – 6000	0.04 – 0.1

$$\lambda \text{ (nm)} = 395.6 / v \text{ (m/s)}$$

$$E \text{ (meV)} = 0.02072 k^2 \text{ (k in nm}^{-1}\text{)}$$

Comparison of Structural Probes



Note that scattering methods provide statistically averaged information on structure rather than real-space pictures of particular instances

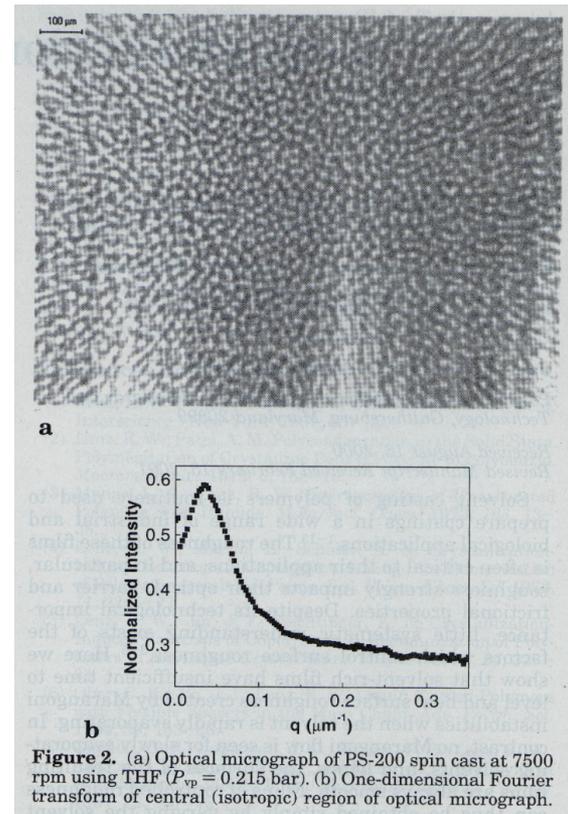
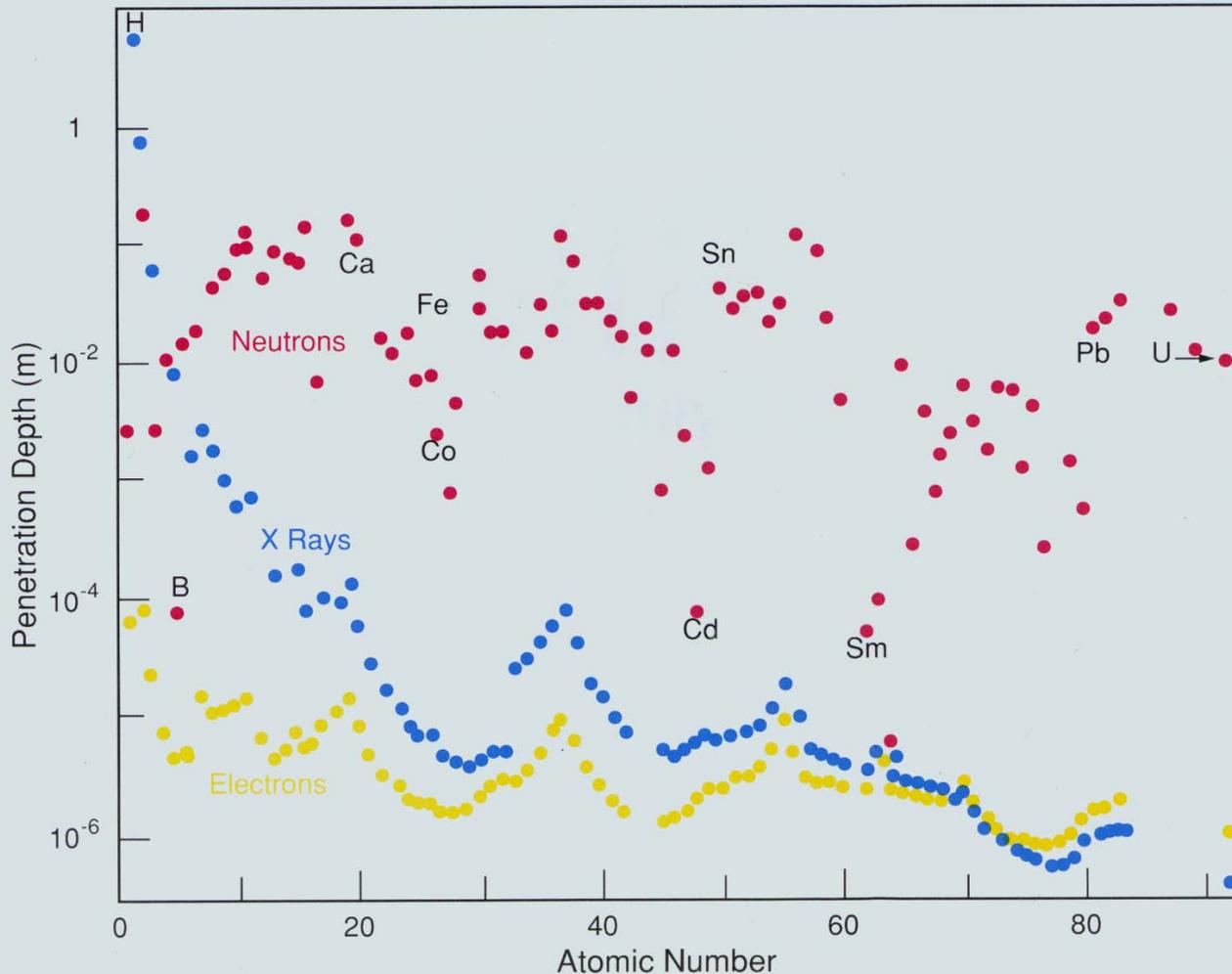


Figure 2. (a) Optical micrograph of PS-200 spin cast at 7500 rpm using THF ($P_{sp} = 0.215$ bar). (b) One-dimensional Fourier transform of central (isotropic) region of optical micrograph.

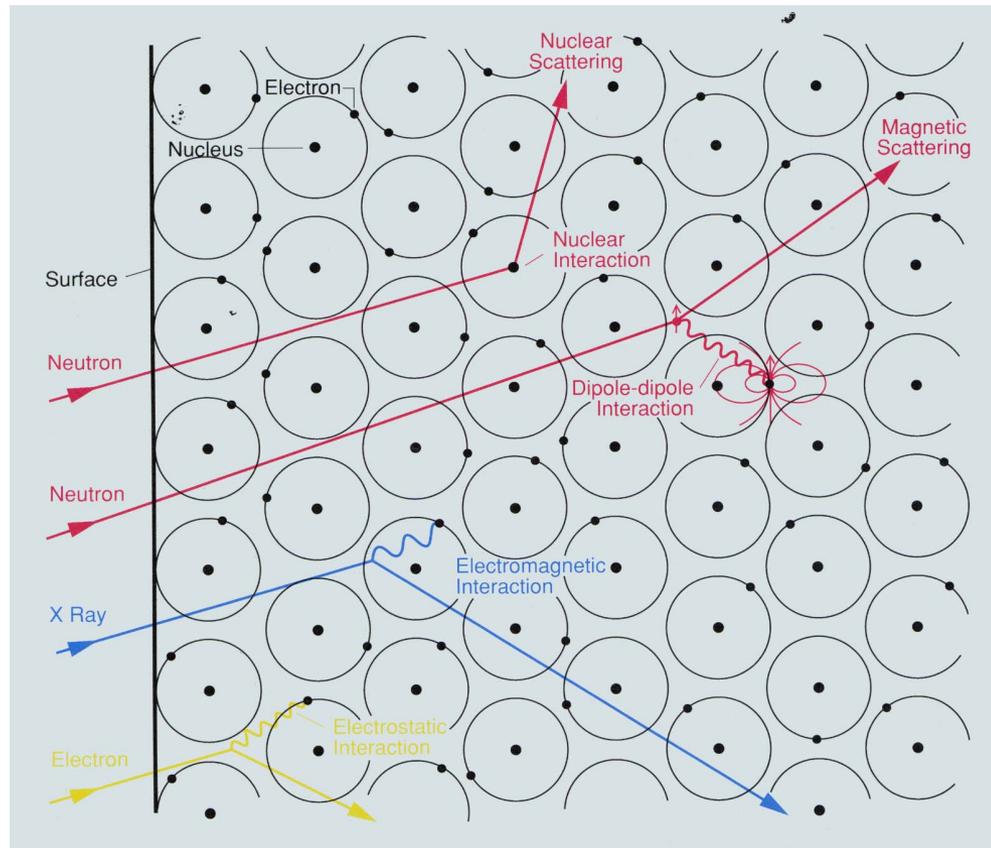
Thermal Neutrons, 8 keV X-Rays & Low Energy Electrons:- Absorption by Matter



Note for neutrons:

- H/D difference
- Cd, B, Sm
- no systematic A dependence

Interaction Mechanisms

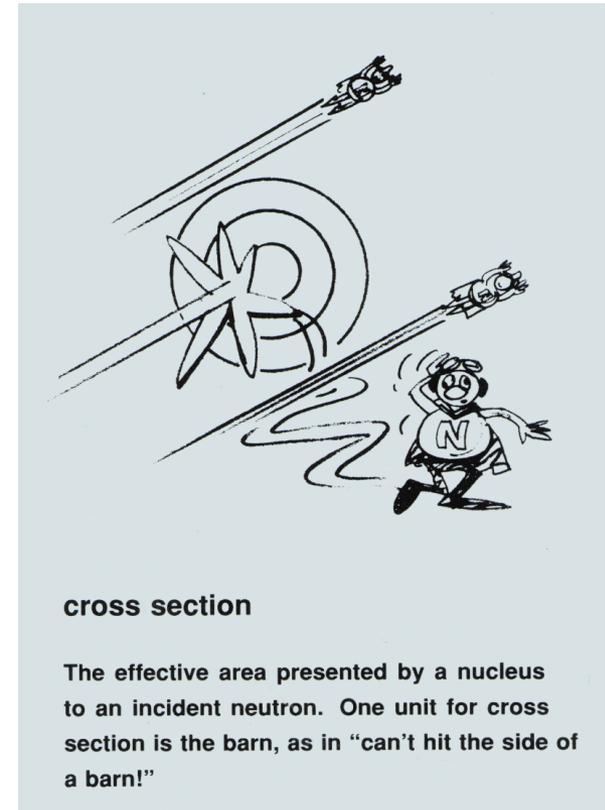
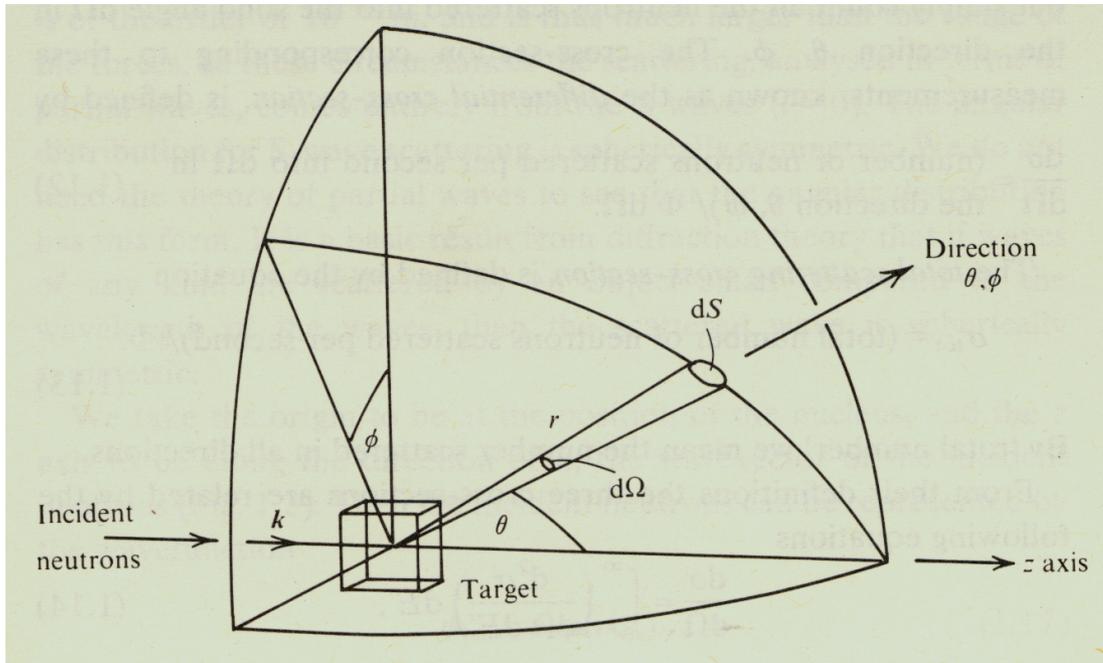


- Neutrons interact with atomic nuclei via very short range (\sim fm) forces.
- Neutrons also interact with unpaired electrons via a magnetic dipole interaction.

Brightness & Fluxes for Neutron & X-Ray Sources

	<i>Brightness</i> ($s^{-1} m^{-2} ster^{-1}$)	<i>dE/E</i> (%)	<i>Divergence</i> ($mrad^2$)	<i>Flux</i> ($s^{-1} m^{-2}$)
Neutrons	10^{15}	2	10 x 10	10^{11}
Rotating Anode	10^{16}	3	0.5 x 10	5×10^{10}
Bending Magnet	10^{24}	0.01	0.1 x 5	5×10^{17}
Wiggler	10^{26}	0.01	0.1 x 1	10^{19}
Undulator (APS)	10^{33}	0.01	0.01 x 0.1	10^{24}

Cross Sections



Φ = number of incident neutrons per cm^2 per second

σ = total number of neutrons scattered per second / Φ

$$\frac{d\sigma}{d\Omega} = \frac{\text{number of neutrons scattered per second into } d\Omega}{\Phi d\Omega}$$

$$\frac{d^2\sigma}{d\Omega dE} = \frac{\text{number of neutrons scattered per second into } d\Omega \text{ \& } dE}{\Phi d\Omega dE}$$

σ measured in barns:

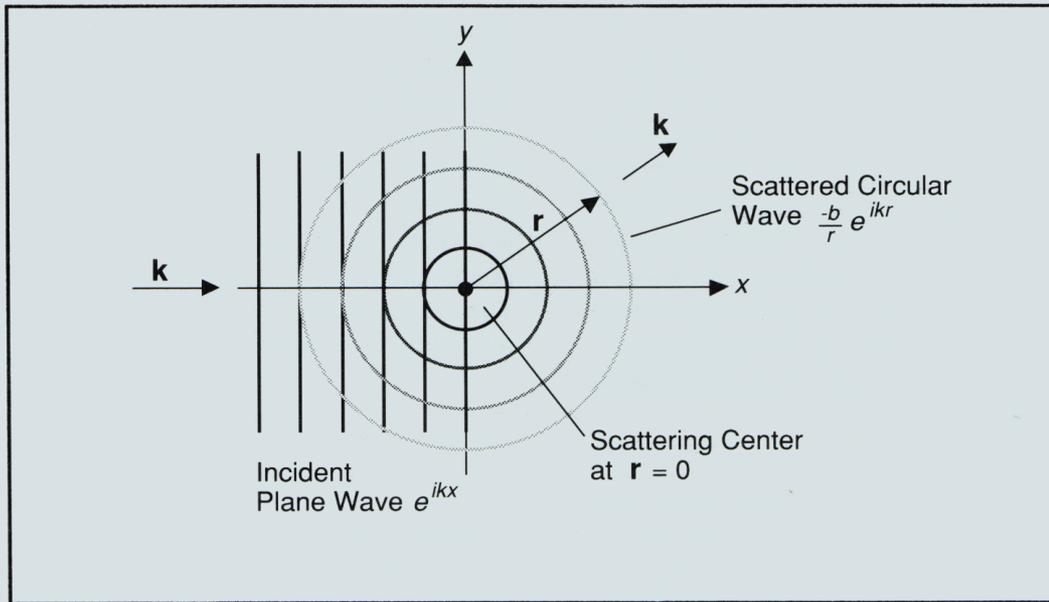
$$1 \text{ barn} = 10^{-24} \text{ cm}^2$$

Attenuation = $\exp(-N\sigma t)$

N = # of atoms/unit volume

t = thickness

Scattering by a Single (fixed) Nucleus



- range of nuclear force (~ 1 fm) is \ll neutron wavelength so scattering is “point-like”
- energy of neutron is too small to change energy of nucleus & neutron cannot transfer KE to a fixed nucleus \Rightarrow scattering is elastic
- we consider only scattering far from nuclear resonances where neutron absorption is negligible

If v is the velocity of the neutron (same before and after scattering), the number of neutrons passing through an area dS per second after scattering is :

$$v \, dS \, |\psi_{\text{scat}}|^2 = v \, dS \, b^2/r^2 = v \, b^2 \, d\Omega$$

Since the number of incident neutrons passing through unit areas is : $\Phi = v |\psi_{\text{incident}}|^2 = v$

$$\frac{d\sigma}{d\Omega} = \frac{v \, b^2 \, d\Omega}{\Phi \, d\Omega} = b^2$$

$$\text{so } \sigma_{\text{total}} = 4\pi b^2 \quad (\text{note units})$$

Adding up Neutrons Scattered by Many Nuclei

At a nucleus located at \vec{R}_i the incident wave is $e^{i\vec{k}_0 \cdot \vec{R}_i}$

so the scattered wave is $\psi_{\text{scat}} = \sum e^{i\vec{k}_0 \cdot \vec{R}_i} \left[\frac{-b_i}{|\vec{r} - \vec{R}_i|} e^{i\vec{k}' \cdot (\vec{r} - \vec{R}_i)} \right]$

$$\therefore \frac{d\sigma}{d\Omega} = \frac{vdS|\psi_{\text{scat}}|^2}{vd\Omega} = \frac{dS}{d\Omega} \left| b_i e^{i\vec{k}' \cdot \vec{r}} \sum \frac{1}{|\vec{r} - \vec{R}_i|} e^{i(\vec{k}_0 - \vec{k}') \cdot \vec{R}_i} \right|^2$$

If we measure far enough away so that $r \gg R_i$ we can use $d\Omega = dS/r^2$ to get

$$\frac{d\sigma}{d\Omega} = \sum_{i,j} b_i b_j e^{i(\vec{k}_0 - \vec{k}') \cdot (\vec{R}_i - \vec{R}_j)} = \sum_{i,j} b_i b_j e^{-i\vec{Q} \cdot (\vec{R}_i - \vec{R}_j)}$$

where the wavevector transfer Q is defined by $\vec{Q} = \vec{k}' - \vec{k}_0$

Coherent and Incoherent Scattering

The scattering length, b_i , depends on the nuclear isotope, spin relative to the neutron & nuclear eigenstate. For a single nucleus:

$$b_i = \langle b \rangle + \delta b_i \quad \text{where } \delta b_i \text{ averages to zero}$$

$$b_i b_j = \langle b \rangle^2 + \langle b \rangle (\delta b_i + \delta b_j) + \delta b_i \delta b_j$$

but $\langle \delta b \rangle = 0$ and $\langle \delta b_i \delta b_j \rangle$ vanishes unless $i = j$

$$\langle \delta b_i^2 \rangle = \langle b_i - \langle b \rangle \rangle^2 = \langle b^2 \rangle - \langle b \rangle^2$$

$$\therefore \frac{d\sigma}{d\Omega} = \langle b \rangle^2 \sum_{i,j} e^{-i\vec{Q} \cdot (\vec{R}_i - \vec{R}_j)} + (\langle b^2 \rangle - \langle b \rangle^2) N$$



Coherent Scattering

(scattering depends on the direction & magnitude of \mathbf{Q})



Incoherent Scattering

(scattering is uniform in all directions)

Note: N = number of atoms in scattering system

Nuclear Spin Incoherent Scattering

Consider a single isotope with spin I . The spin of the nucleus - neutron system can be $(I + 1/2)$ or $(I - 1/2)$.

The number of states with spin $(I + 1/2)$ is $2(I + 1/2) + 1 = 2I + 2$

The number of states with spin $(I - 1/2)$ is $2(I - 1/2) + 1 = 2I$

If the neutrons and the nuclear spins are unpolarized, each spin state has the same *a priori* probability.

The frequency of occurrence of b^+ state is $f^+ = (2I + 2)/(4I + 2)$

The frequency of occurrence of b^- state is $f^- = (2I)/(4I + 2)$

Thus $\langle b \rangle = \frac{1}{2I + 1} [(I + 1)b^+ + Ib^-]$ and $\langle b^2 \rangle = \frac{1}{2I + 1} [(I + 1)(b^+)^2 + I(b^-)^2]$

Values of s_{coh} and s_{inc}

Nuclide	s_{coh}	s_{inc}	Nuclide	s_{coh}	s_{inc}
^1H	1.8	80.2	V	0.02	5.0
^2H	5.6	2.0	Fe	11.5	0.4
C	5.6	0.0	Co	1.0	5.2
O	4.2	0.0	Cu	7.5	0.5
Al	1.5	0.0	^{36}Ar	24.9	0.0

- Difference between H and D used in experiments with soft matter (contrast variation)
- Al used for windows
- V used for sample containers in diffraction experiments and as calibration for energy resolution
- Fe and Co have nuclear cross sections similar to the values of their magnetic cross sections
- Find scattering cross sections at the NIST web site at:
<http://webster.ncnr.nist.gov/resources/n-lengths/>

Coherent Elastic Scattering measures the Structure Factor $S(\vec{Q})$ I.e. correlations of atomic positions

$$\frac{d\sigma}{d\Omega} = \langle b \rangle^2 N \cdot S(\vec{Q}) \quad \text{for an assembly of similar atoms where} \quad S(\vec{Q}) = \frac{1}{N} \left\langle \sum_{i,j} e^{-i\vec{Q} \cdot (\vec{R}_i - \vec{R}_j)} \right\rangle_{\text{ensemble}}$$

$$\text{Now } \sum_i e^{-i\vec{Q} \cdot \vec{R}_i} = \int d\vec{r} \cdot e^{-i\vec{Q} \cdot \vec{r}} \sum_i \delta(\vec{r} - \vec{R}_i) = \int d\vec{r} \cdot e^{-i\vec{Q} \cdot \vec{r}} \rho_N(\vec{r}) \quad \text{where } \rho_N \text{ is the nuclear number density}$$

so

$$S(\vec{Q}) = \frac{1}{N} \left\langle \left| \int d\vec{r} \cdot e^{-i\vec{Q} \cdot \vec{r}} \rho_N(\vec{r}) \right|^2 \right\rangle$$

$$\text{or } S(\vec{Q}) = \frac{1}{N} \int d\vec{r}' \int d\vec{r} \cdot e^{-i\vec{Q} \cdot (\vec{r} - \vec{r}')} \langle \rho_N(\vec{r}) \rho_N(\vec{r}') \rangle = \frac{1}{N} \int d\vec{R} \int d\vec{r} e^{-i\vec{Q} \cdot \vec{R}} \langle \rho_N(\vec{r}) \rho_N(\vec{r} - \vec{R}) \rangle$$

ie

$$S(\vec{Q}) = 1 + \int d\vec{R} \cdot \{g(\vec{R}) - \bar{\rho}\} \cdot e^{-i\vec{Q} \cdot \vec{R}}$$

$$\text{where } g(\vec{R}) = \sum_{i \neq 0} \langle \delta(\vec{R} - \vec{R}_i + \vec{R}_0) \rangle \quad \text{is a function of } \vec{R} \text{ only.}$$

$g(\vec{R})$ is known as the **static pair correlation function**. It gives the probability that there is an atom, i , at distance R from the origin of a coordinate system, given that there is also a (different) atom at the origin of the coordinate system at the same instant in time.

S(Q) and g(r) for Simple Liquids

- Note that $S(Q)$ and $g(r)/r$ both tend to unity at large values of their arguments
- The peaks in $g(r)$ represent atoms in “coordination shells”
- $g(r)$ is expected to be zero for $r <$ particle diameter – ripples are truncation errors from Fourier transform of $S(Q)$

Fig. 5.1 The structure factor $S(\kappa)$ for ^{36}Ar at 85 K. The curve through the experimental points is obtained from a molecular dynamics calculation of Verlet based on a Lennard-Jones potential. (After Yarnell *et al.*, 1973.)

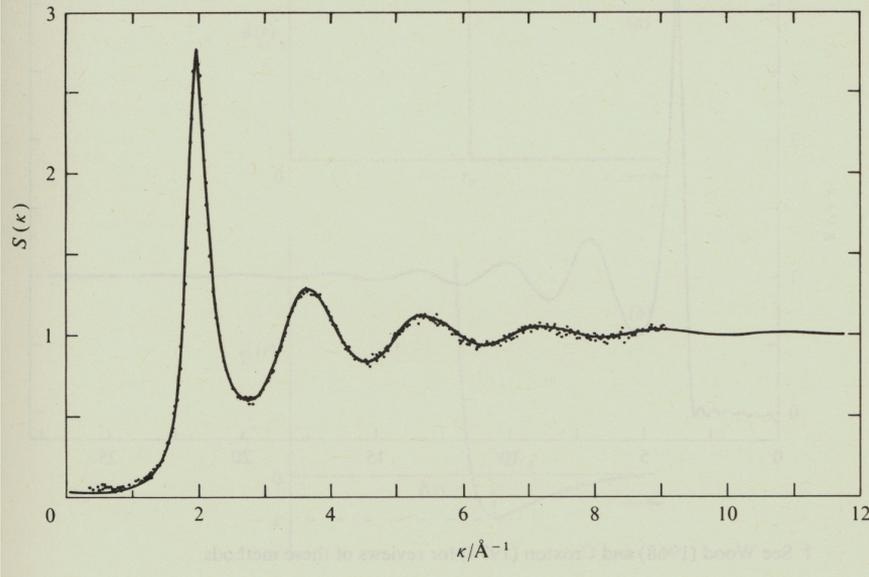
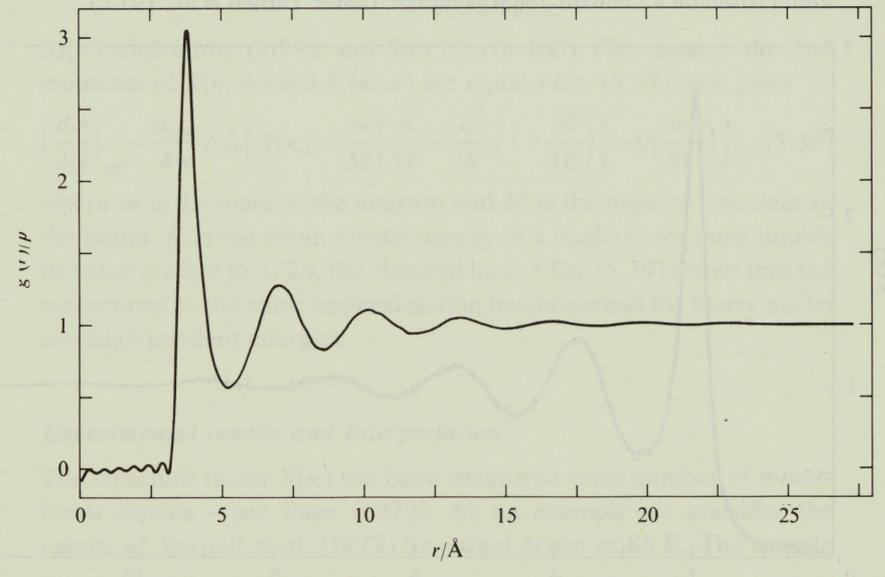


Fig. 5.2 The pair-distribution function $g(r)$ obtained from the experimental results in Fig. 5.1. The mean number density is $\rho = 2.13 \times 10^{28}$ atoms m^{-3} . (After Yarnell *et al.*, 1973.)

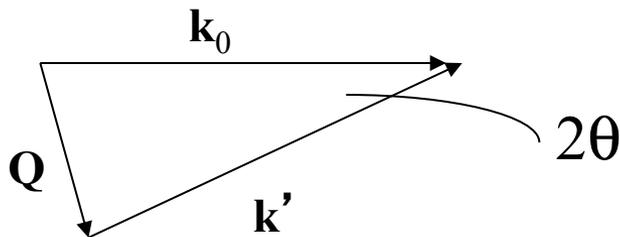


Summarizing:

$$\frac{d\sigma}{d\Omega} = \frac{\text{number of neutrons scattered through angle } 2\theta \text{ per second into } d\Omega}{\text{number of incident neutrons per square cm per second}}$$

$$\left(\frac{d\sigma}{d\Omega}\right)_{coh} = \sum_{i,j} b_i^{coh} b_j^{coh} e^{i(\vec{k}_0 - \vec{k}') \cdot (\vec{R}_i - \vec{R}_j)} = \sum_{i,j} b_i^{coh} b_j^{coh} e^{-i\vec{Q} \cdot (\vec{R}_i - \vec{R}_j)}$$

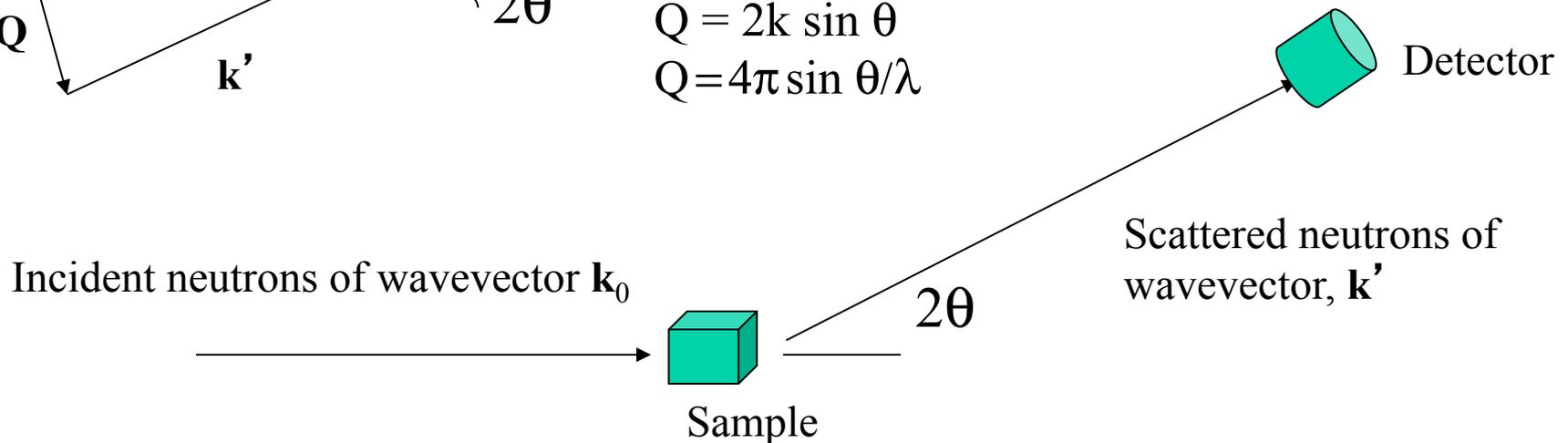
where the wavevector transfer \vec{Q} is defined by $\vec{Q} = \vec{k}' - \vec{k}_0$



For elastic scattering $k_0 = k' = k$:

$$Q = 2k \sin \theta$$

$$Q = 4\pi \sin \theta / \lambda$$



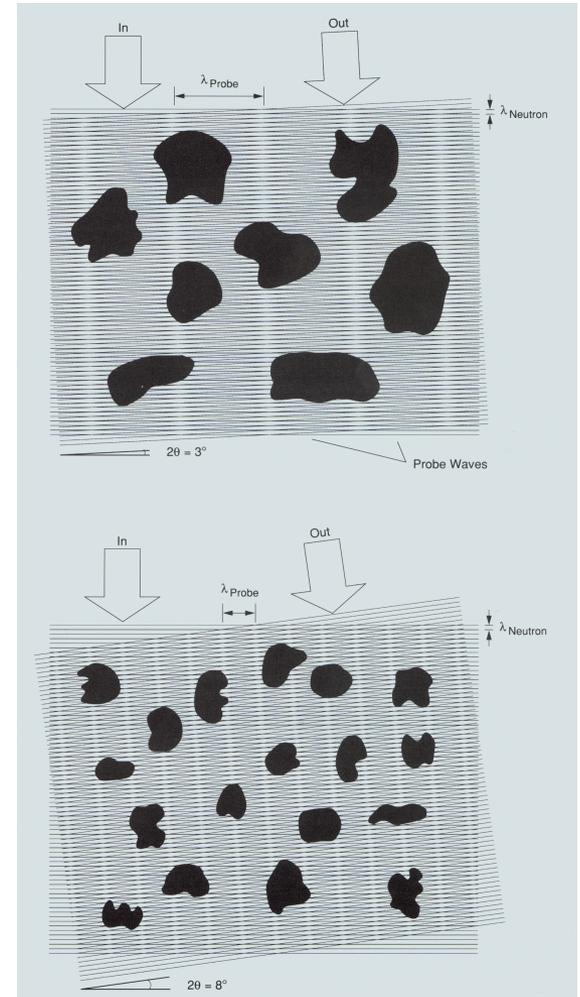
Neutron Diffraction

- Neutron diffraction is used to measure the differential cross section, $d\sigma/d\omega$
 - Crystalline solids
 - Unit cell size; crystal symmetry; atomic arrangement and thermal motions (ellipsoids)
 - Liquids and amorphous materials
 - Large scale structures

- Depending on the scattering angle, structure on different length scales, d , is measured:

$$2\pi / Q = d = \lambda / 2 \sin(\theta)$$

- For crystalline solids & liquids, use wide angle diffraction. For large structures, e.g. polymers, colloids, micelles, etc. use small-angle neutron scattering



The Kinematical Approximation

- Note that the approximation we have just seen ignores
 - Depletion of the incident beam by scattering or absorption
 - Multiple scatteringi.e. energy is not conserved
- This so-called “kinematic approximation” is OK for weak scattering, very small crystals or “bad” crystals
- It is usually used for interpreting diffraction experiments, though “extinction corrections” are often needed with single crystals
 - If it’s not adequate, use dynamical theory
- In addition, we have so-far ignored thermal motion of atoms

Bragg Scattering from Crystals

Working through the math (see, for example, Squires' book), we find :

$$\left(\frac{d\sigma}{d\Omega} \right)_{\text{Bragg}} = N \frac{(2\pi)^3}{V_0} \sum_{hkl} \delta(\vec{Q} - \vec{G}_{hkl}) |F_{hkl}(\vec{Q})|^2$$

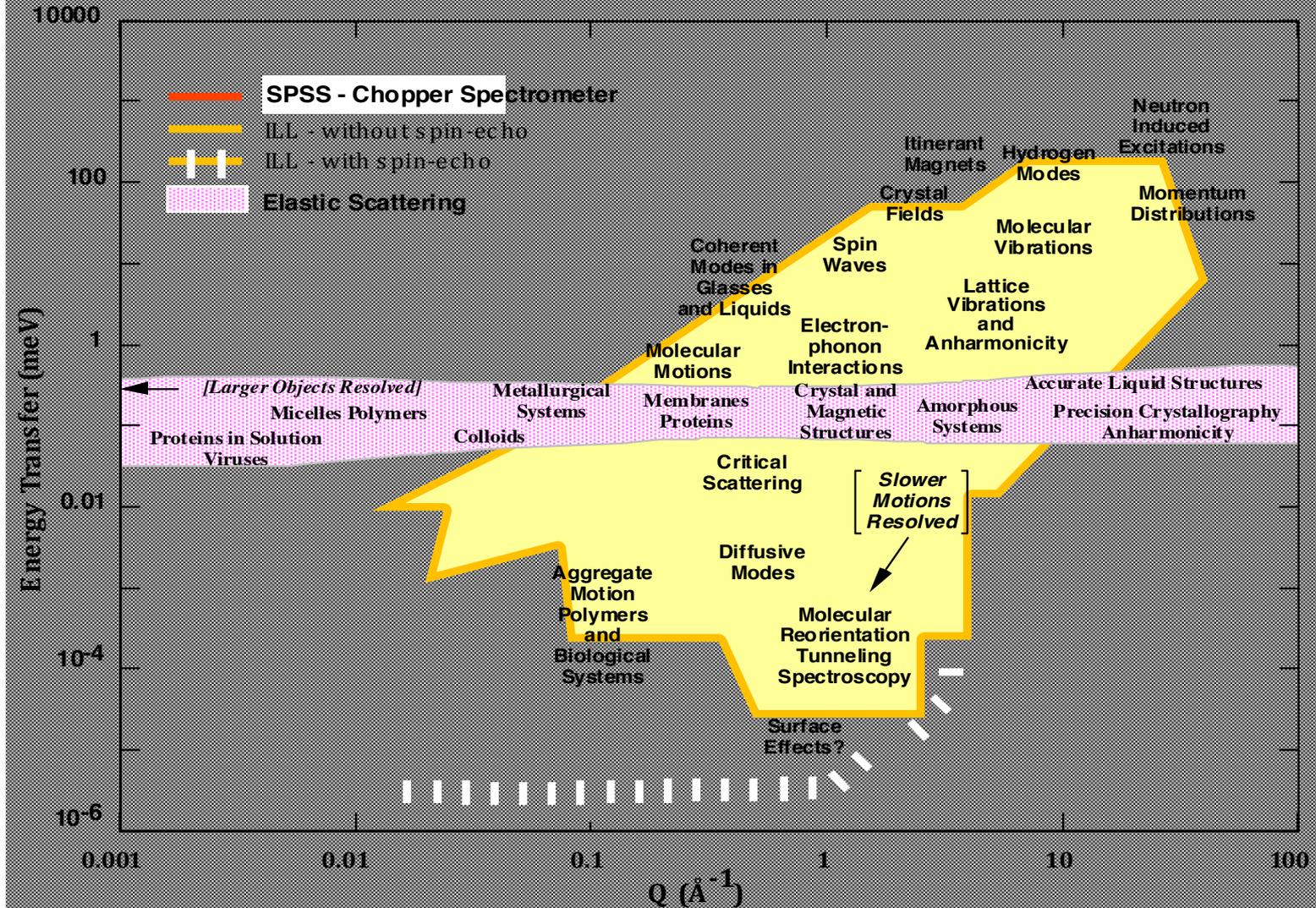
where the unit - cell structure factor is given by

$$F_{hkl}(\vec{Q}) = \sum_d \bar{b}_d e^{i\vec{Q} \cdot \vec{d}} e^{-W_d}$$

and W_d is the Debye - Waller factor that accounts for thermal motions of atoms

- Using either single crystals or powders, neutron diffraction can be used to measure F^2 (which is proportional to the intensity of a Bragg peak) for various values of (hkl).
- Direct Fourier inversion of diffraction data to yield crystal structures is not possible because we only measure the magnitude of F , and not its phase => models must be fit to the data
- Neutron powder diffraction has been particularly successful at determining structures of new materials, e.g. high T_c materials

Neutrons in Condensed Matter Research



Neutron scattering experiments measure the number of neutrons scattered at different values of the wavevector and energy transferred to the neutron, denoted Q and E . The phenomena probed depend on the values of Q and E accessed.

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